

Last trends in scintillators development; theory and practice

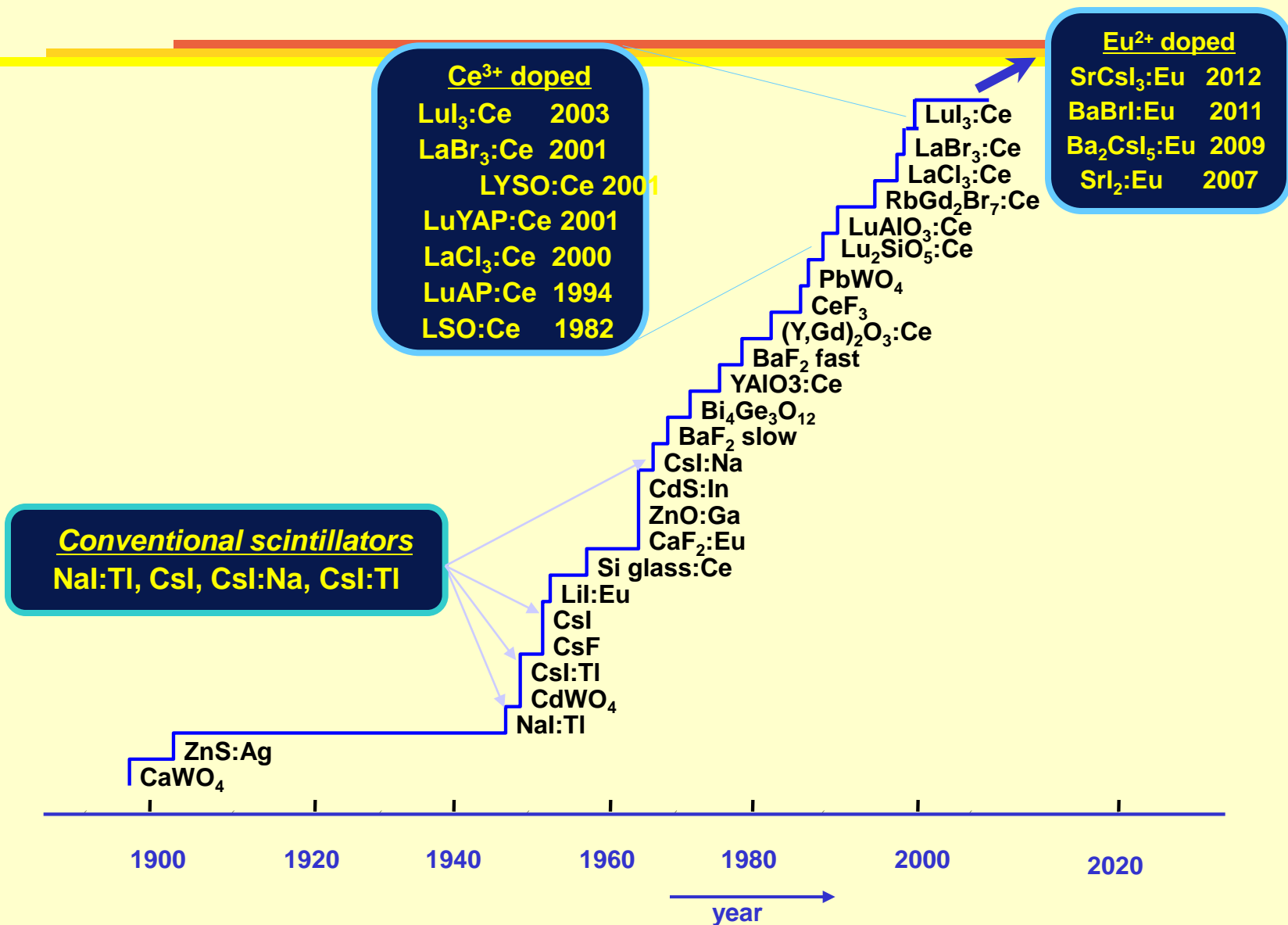
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**Skobeltsyn Institute of Nuclear Physics, Lomonosov Moscow State University, Leninskie Gory 1(2),
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Scintillation discovery





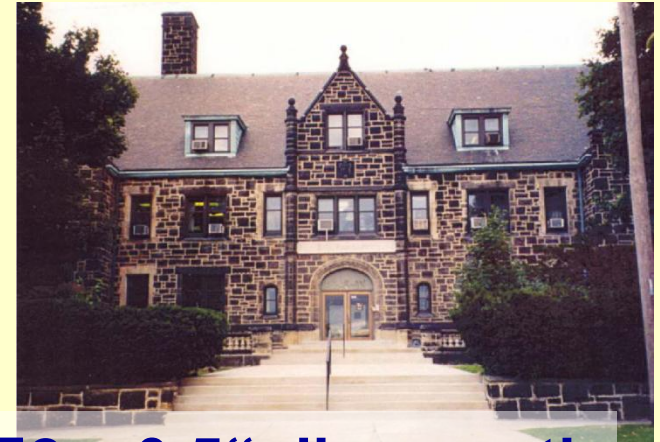
NaI(Tl) way to Crystal Ball

1949 – NaI(Tl) invention
1963 – Gamma-camera prototype

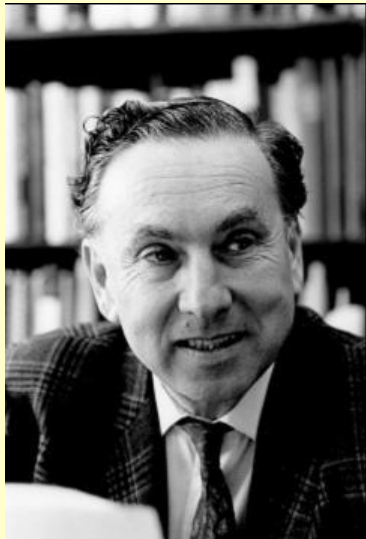


Large NaI(Tl) needs

Harshaw Chemical Company



1952 – 0,5" dia growth
1974 – 21" dia growth



Dr. Carl Swinehart
In Harshaw from
1932 to 1990

R&D supervisors !!! :
Prof. R. Hofstadter (Stanford)
and Dr. D. Stockbarger (MIT)





ISMA, Scintillator Developments

New scintillator ?

New projects?

2011-12 LPS

2012 - CALIFA (FAIR)

2007 LGSO

2008 GLAST (Int. NASA)

2004 dedicated imaging system

2007 AGILE (EC)

2005 NEMO (Int. NASA)

2005 OPERA (EC)

2000 PiBeta (Switzerland)

2000 SPECT

1998 BaBar (USA)

1994 CsI(Tl) CRYSTALS of 525 kg WEIGHT

1996 BELLE (Japan)

1992 PWO

1992-2008 LHC (CERN)

1989 CsI(CO₃)

1987 UNDOPED CsI SINGLE CRYSTALS

1986 CdWO₄, BiGe₃O₁₂

1985 "CRYSTAL" TECHNOLOGY

1978 "ROST" TECHNOLOGY

1974 ZnSe(Te)

1964 CsI(Na)

1961 PMMA

1959 NaI(Tl)

1955 ISC WAS FOUNDED



PWO - INNOVATION AT HIGH ENERGY PHYSICS CRYSTAL



L.Nagornaya



Material: PWO
Invention: 1991

PWO milestones:

1992 : Crystal-2000; PWO promotion

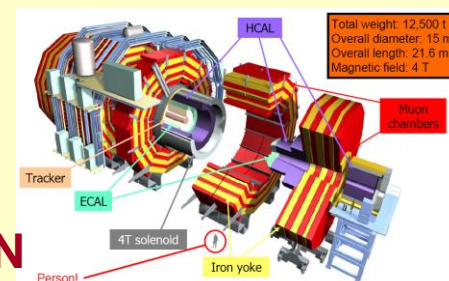
1994 : Choice of PWO for CMS e-cal

1994-1998 : extensive R&D on PWO

1998-2000 : Preproduction of 6000 crystals

2001 : Start of the production

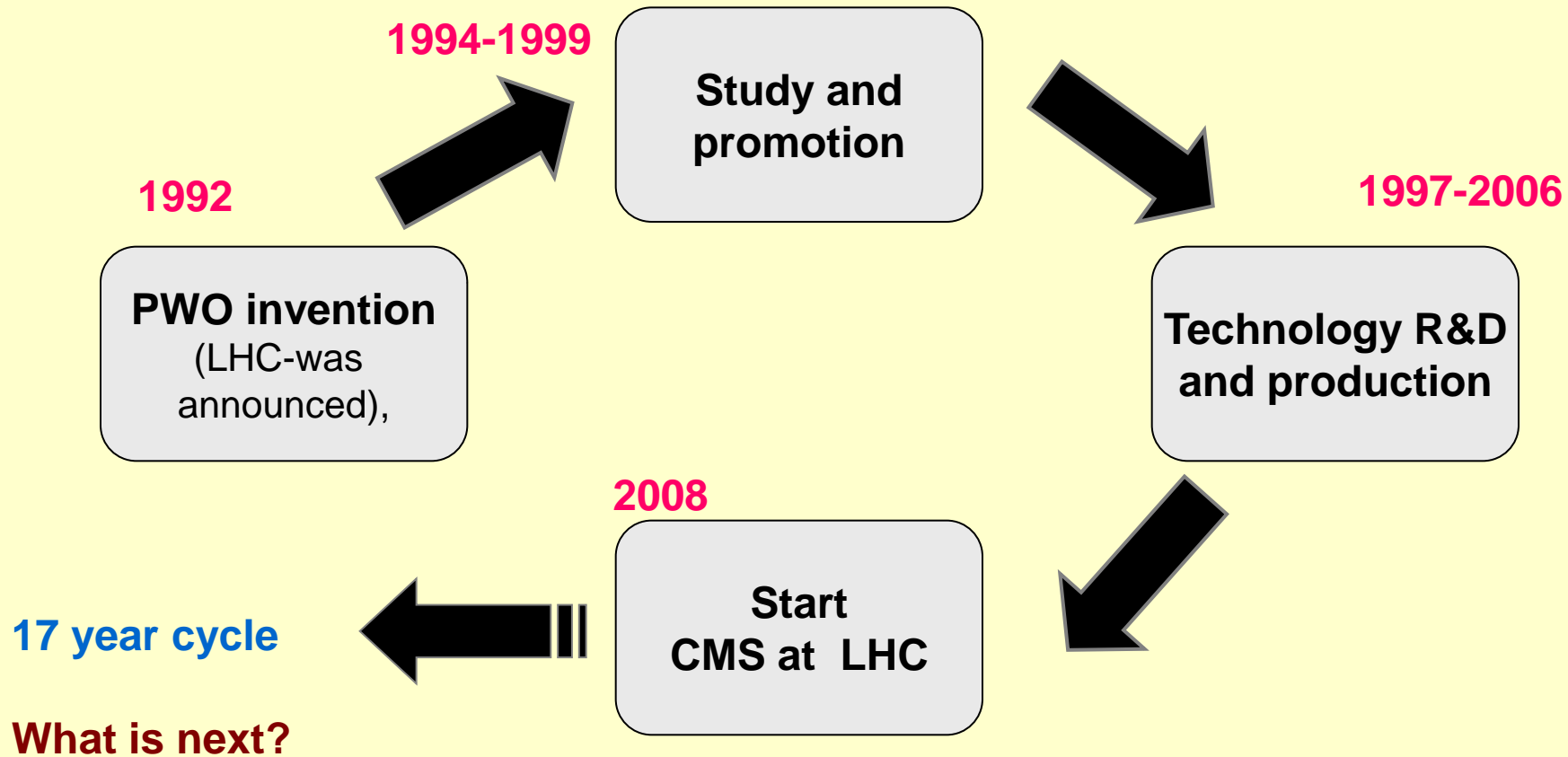
2008: E-cal complete



LHC - CERN

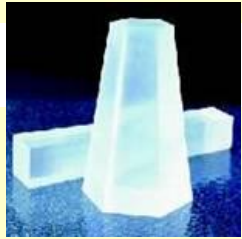


PbWO₄(PWO) cycle. From invention to LHC use





ISMA for International Collaborations



High Energy and Nuclear Physics



CMS



PANDA



NEMO



OPERA

BELLE (Japan)



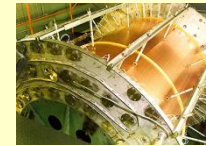
BaBar (USA)



PiBeta (Switzerland)



KEDR



CALIFA (FAIR)



Space Missions

GLAST (USA, Sweden...)



AGILE (Italy)





Is it possible to improve scintillation efficiency of conventional scintillators?

Optimal scintillator

Effective = efficient + available + cheap

efficient ~ 100.000 ph/MeV, 3% resolution (662 keV)

available ~ size 400 mm

cheap ~ 3 \$/cc

History (2008-2009) and start point:

Two ways to obtain (develop) new efficient scintillator:

A - search of new compounds (Successfully done!)

(-) Deep scientific search !?

(-) Ability to grow large crystal ?

(-) Crystal cost ?

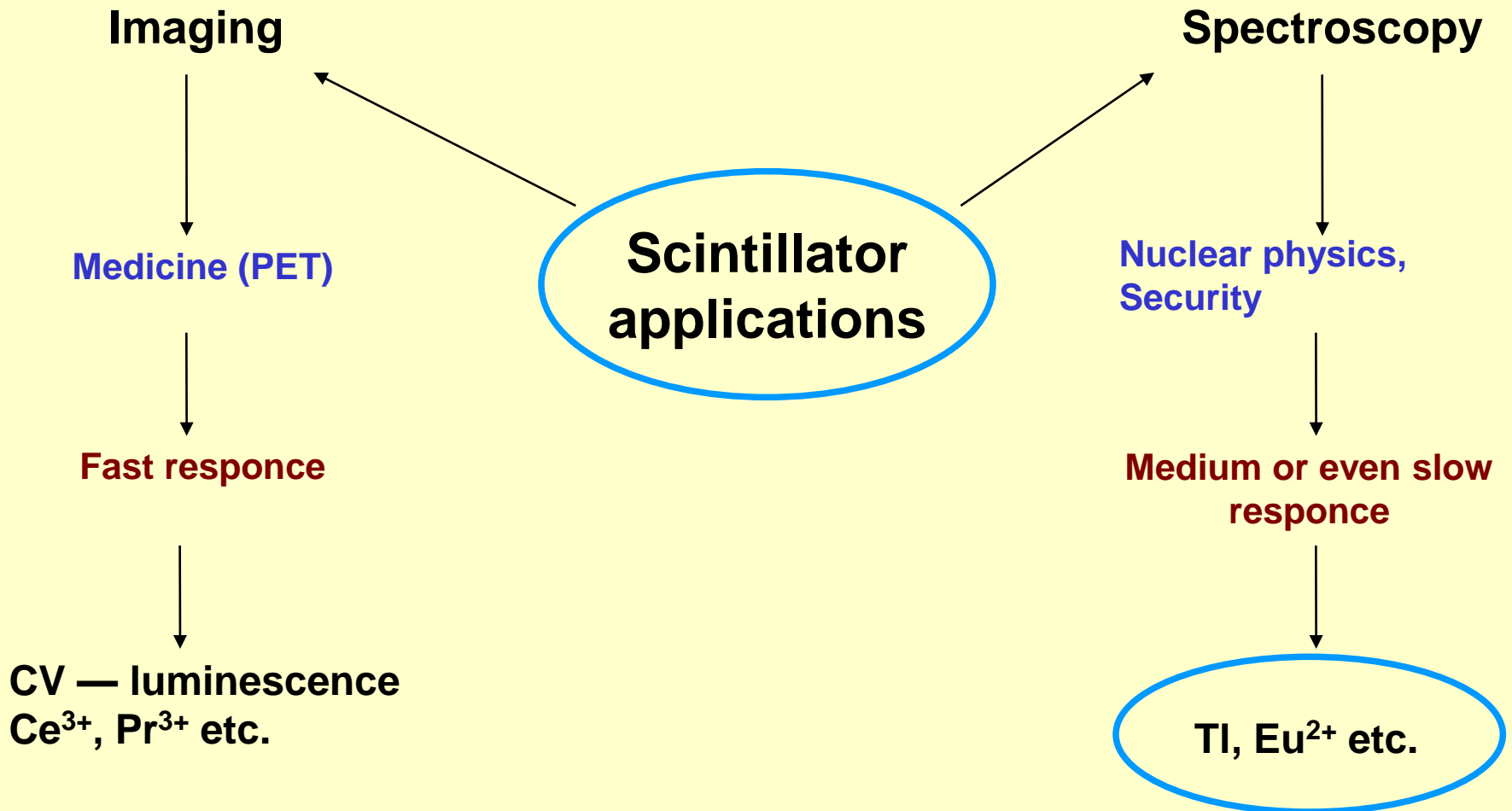
B - modification of conventional scintillators

(-) Need in modification idea ?!

(+) Existed technologies of industrial growth!



Logics and motivation





Most efficient new scintillators

Crystal	ρ g/cm ³	Lum λ , nm	LY ph/Mev	R, % Cs ¹³⁷	Decay τ , ns	Hygro- scopy	References
CaI ₂ :Eu	3.96	467	110.000	5,2	1.000	strong	Cherepy, Moses, Derenzo, Bizarri, Bourret et al. 2007 - 2012
SrI ₂ :Eu	4.55	435	115.000	2.6	1.500	strong	
Ba ₂ CsI ₅ :Eu	4.9	435	102.000	2.55	383;1.50 0	medium	
SrCsI ₃ :Eu	4,25	458	73.000	3.9	2.200	medium	Zhuravleva et al. 2012
BaBrI :Eu	5.2	413	97.000	3,4	500	low	Bizarri et al. 2011
NaI : Tl	3.67	415	44.000	5.6	230	strong	
CsI : Tl	4.53	560	56,000	6.0	980	no	
CsI : Na		420	46,000	6.4	600	low	



Maximal scintillator light yield

Scintillator efficiency:

$$N_{\text{ph}} = \beta S Q$$

$$\beta = \frac{E_{\gamma}}{E_{e-h}}$$

E_{γ} quantum energy

$$E_{e-h} = \sim 2.4 E_g$$

S energy transfer efficiency

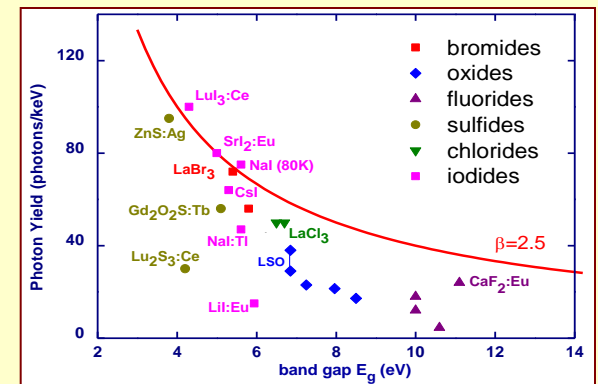
Q luminescence center efficiency

β – e-h creation efficiency is a key to the new material search and investigation

Q is ~ 1 for many typical activators, Ce, Eu etc

S is also ~ 1 for many hosts.

1-5% of uniform distributed activator minimizes the transfer length to 2-5 a (lattice parameters)

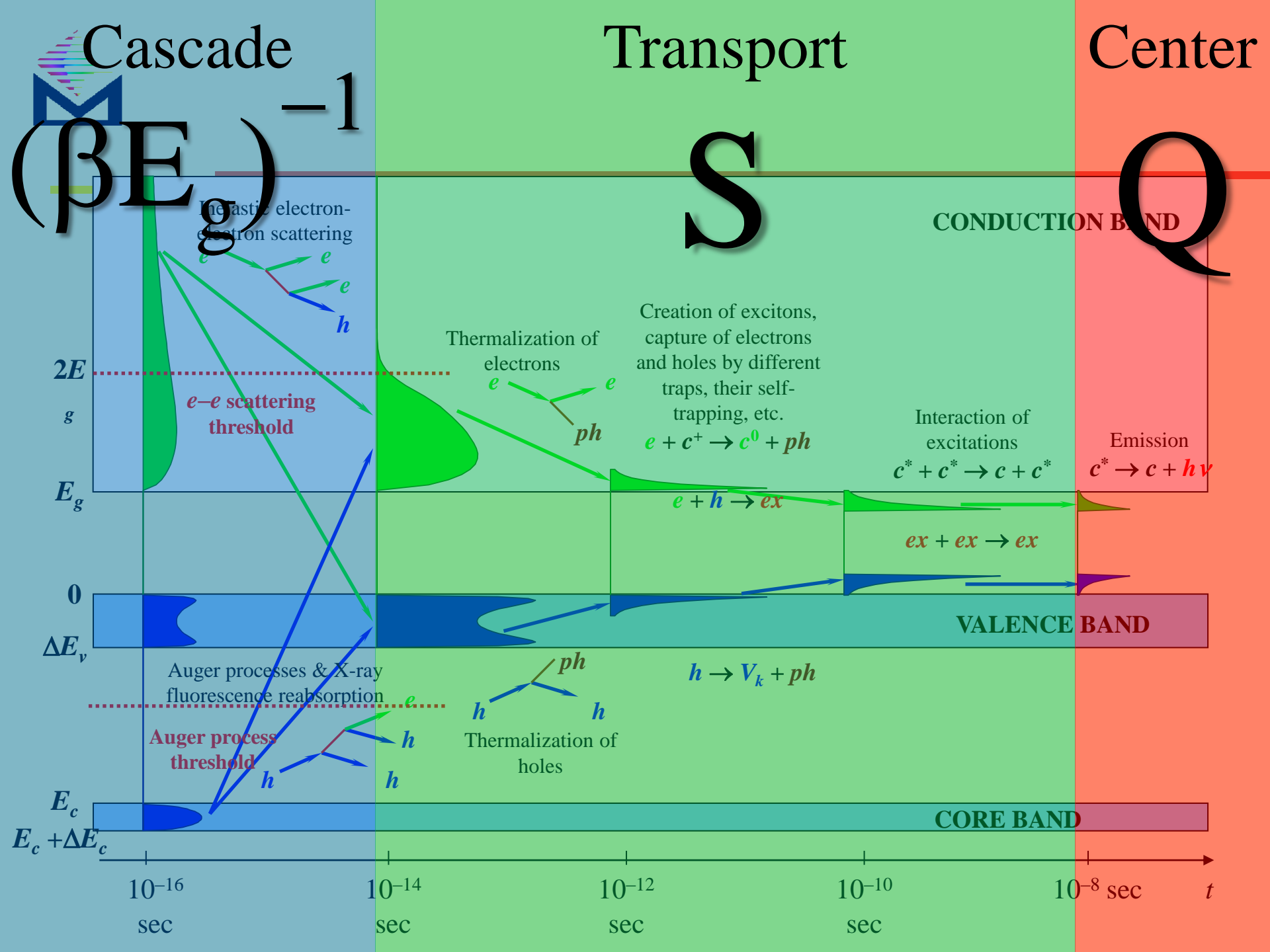


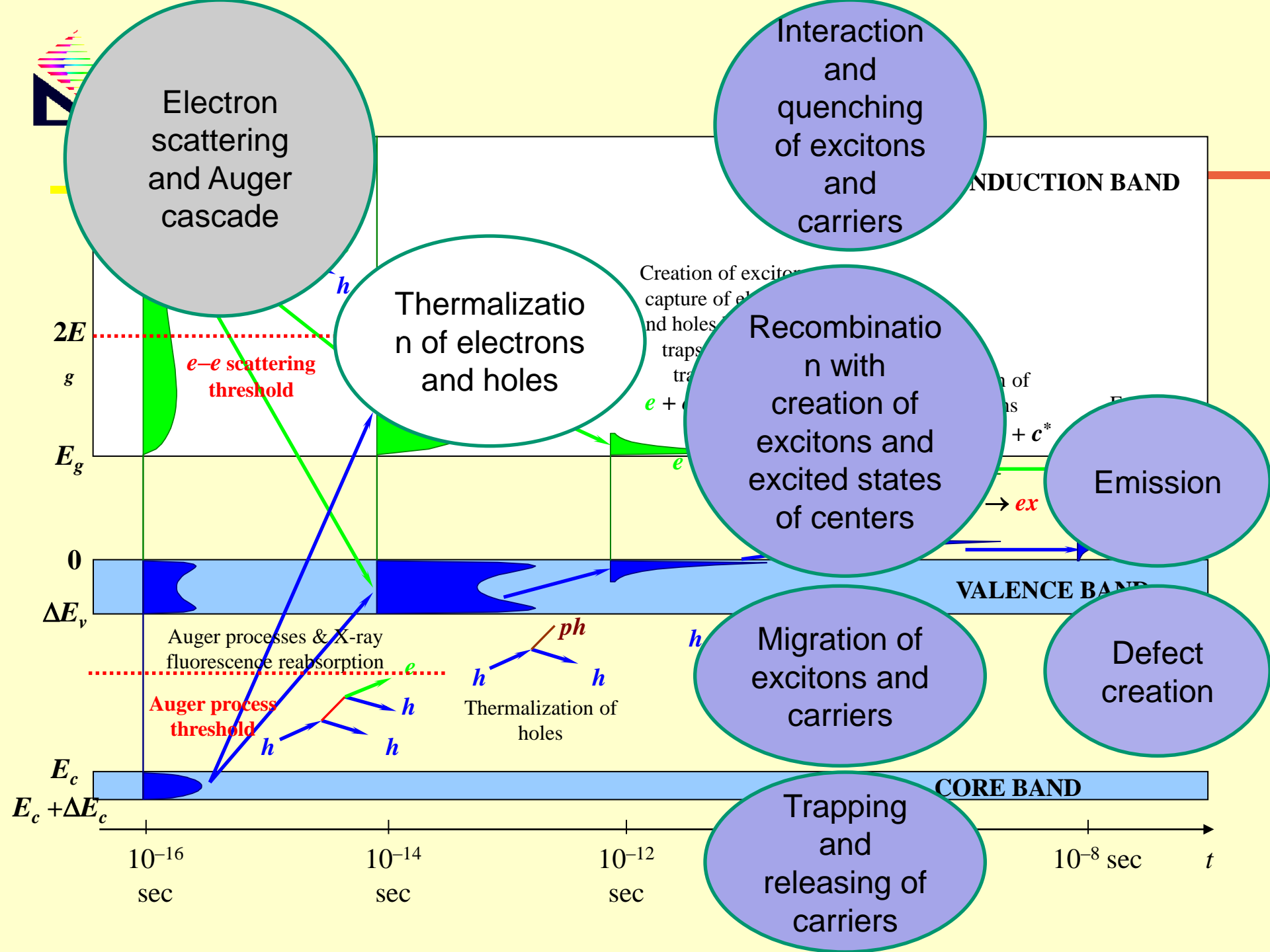
P.Dorenbos, SCINT, 2009



What was done last years?

- ✓ Maximal yield for alkali halides is far from the fundamental limit . *Some limits could be described*
- ✓ Why alkali-earth (AE) halides are more efficient? *The yield is close to fundamental limit*
- ✓ Can we obtain (grow) scintillators with the large size and high industrial efficiency? *Why not? What are the problems?*
- ✓ Natural “bottle neck” (self absorption). *Overpass ways*







Electron scattering and Auger cascade

Interaction and quenching of excitons and carriers

Thermalization of electrons and holes

Recombination with creation of excitons and excited states of centers

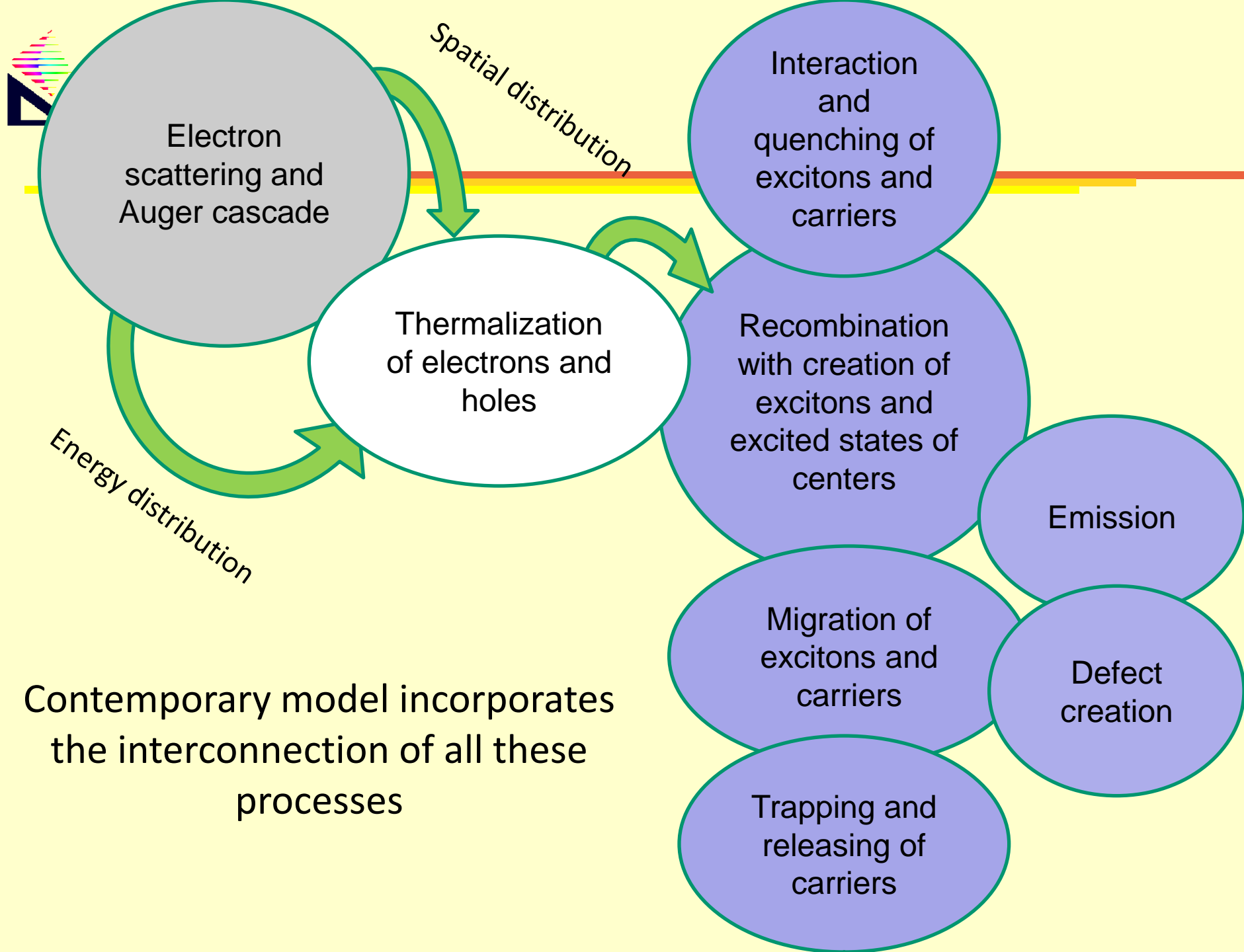
Emission

Migration of excitons and carriers

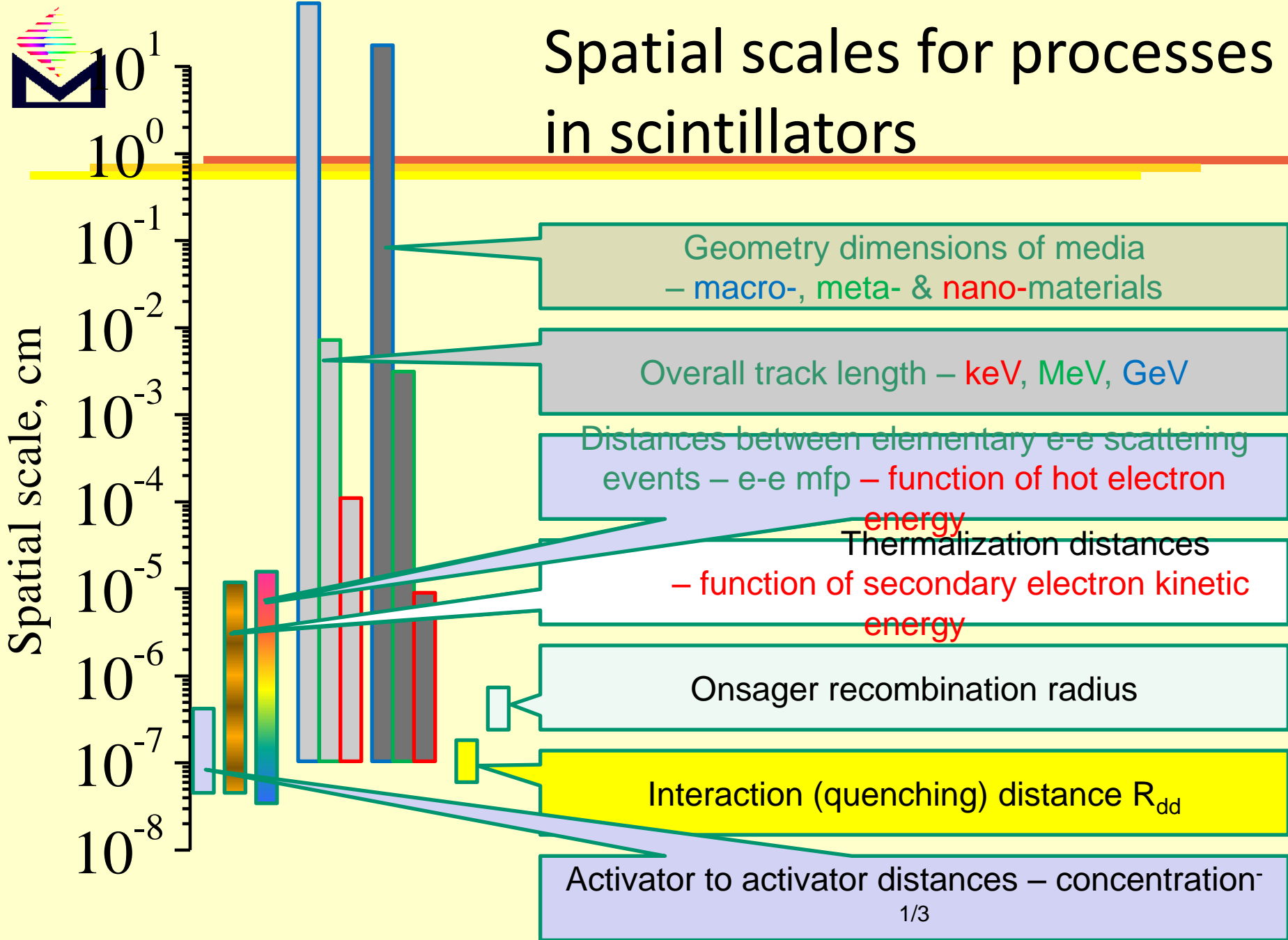
Defect creation

Trapping and releasing of carriers

Are these processes independent?
NO!



Spatial scales for processes in scintillators





Outline

Spatial scales for processes in scintillators

Nanoparticles as scintillators

Cascade, thermalization and recombination

Different types of mobilities

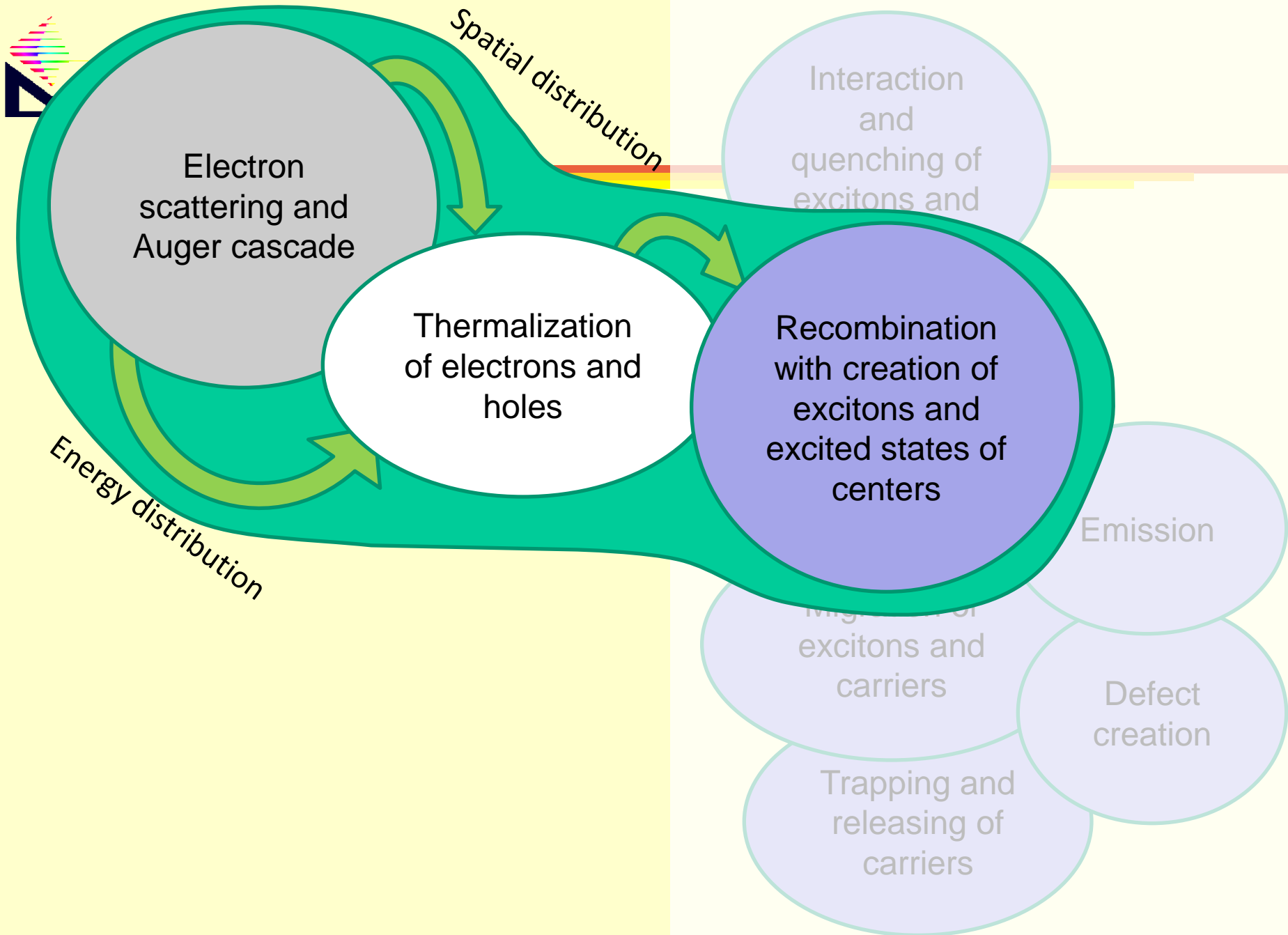
Thermalization length for different types of crystals

Interconnection of cascade, thermalization and recombination stages in binary iodides

Why cascade is so effective in CsI?

Thermalization length and impurities

Concluding remarks



Electron scattering and Auger cascade

Thermalization of electrons and holes

Recombination with creation of excitons and excited states of centers

Interaction and quenching of excitons and

Emission

Trapping and releasing of carriers

Defect creation

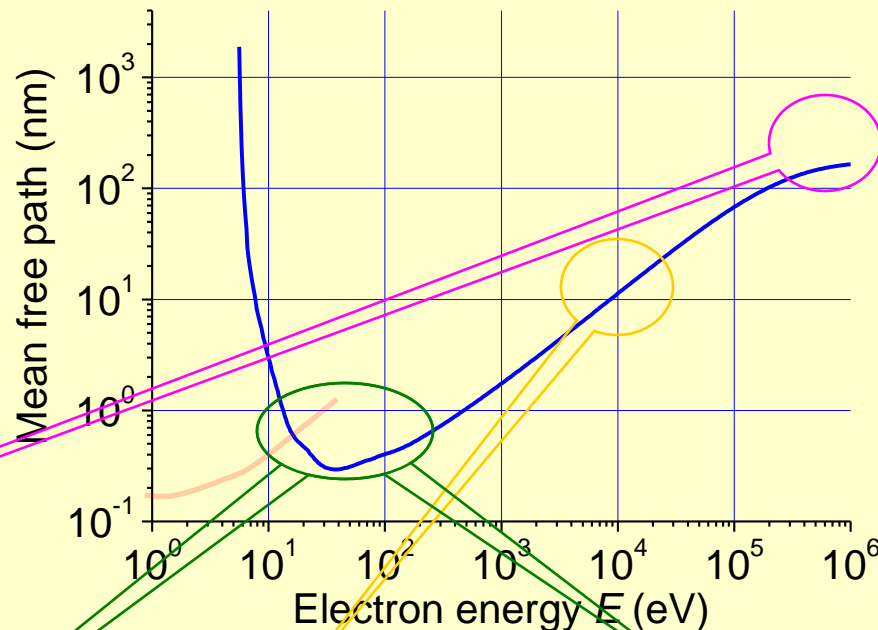
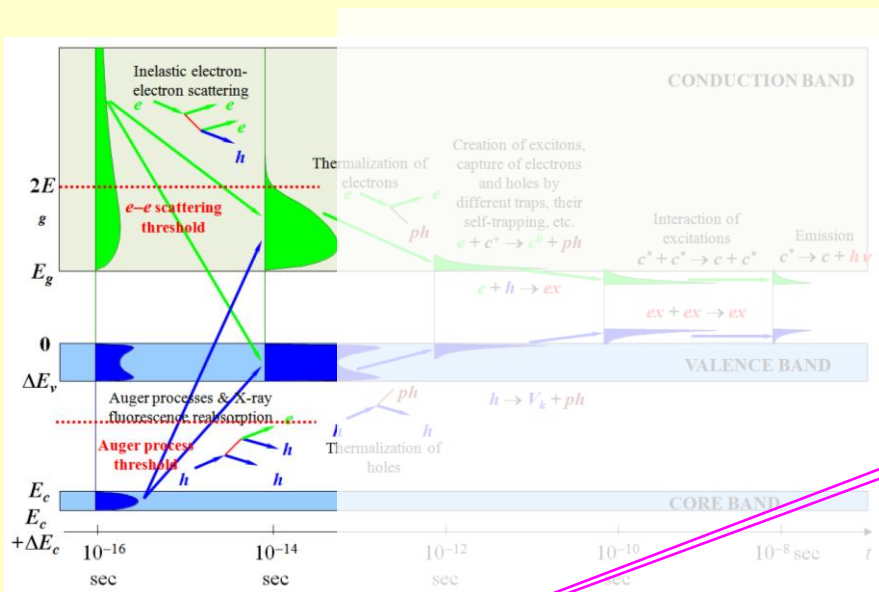
Trapping and releasing of carriers

Spatial distribution

Energy distribution



Spatial track structure for e-e scattering stage (prior to thermalization)



'Real' track structure

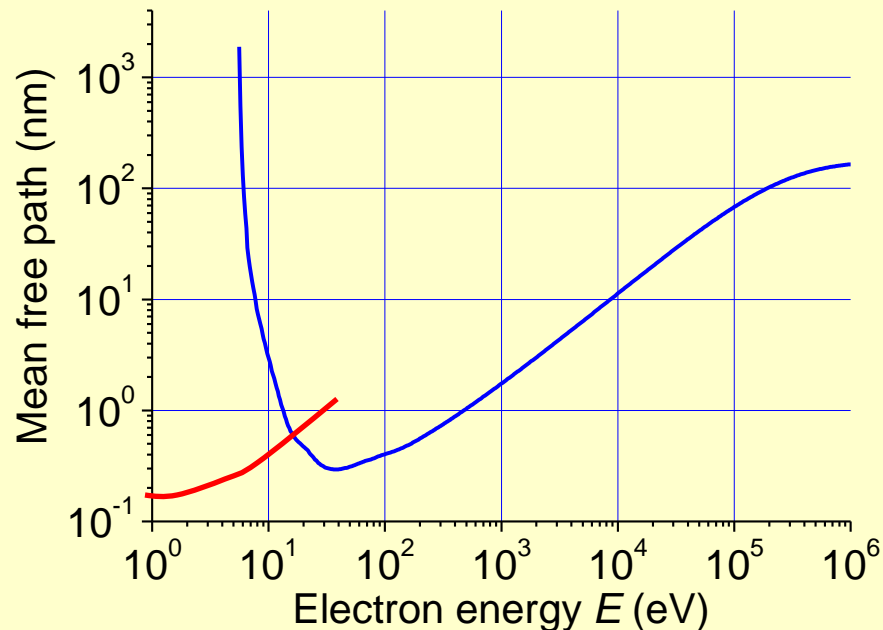
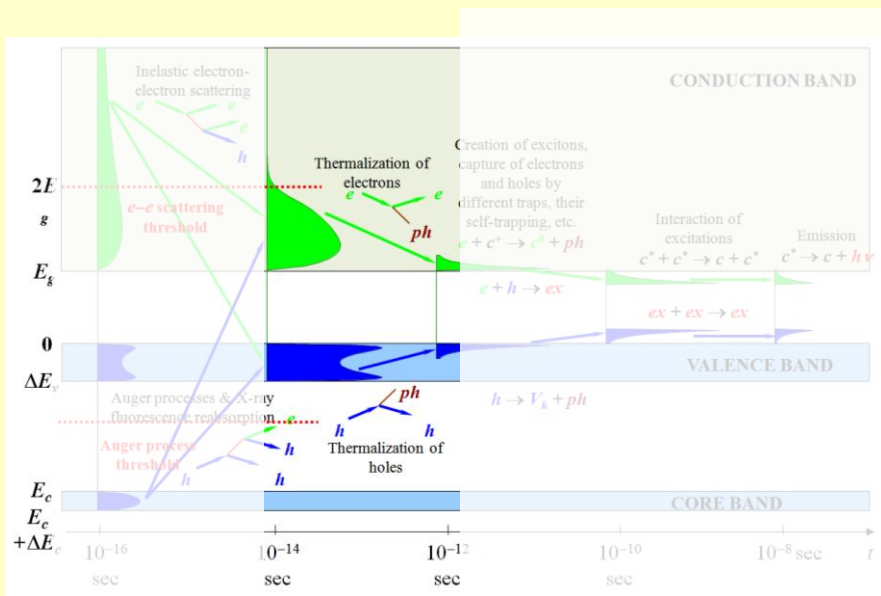


100 nm

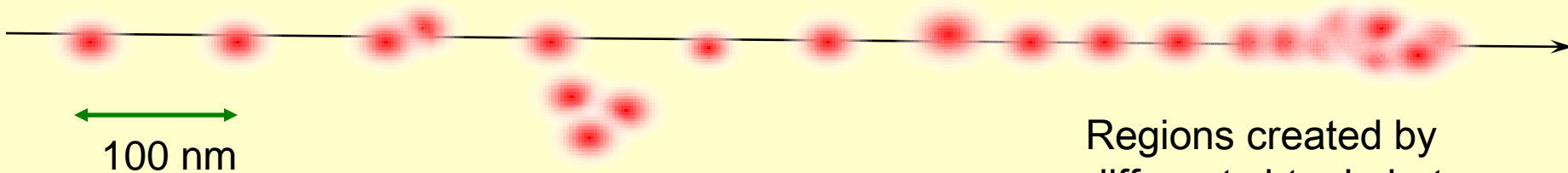
Regions created by different virtual photons are overlapped



Spatial track structure for phonon scattering stage (after thermalization) for small thermalization radius



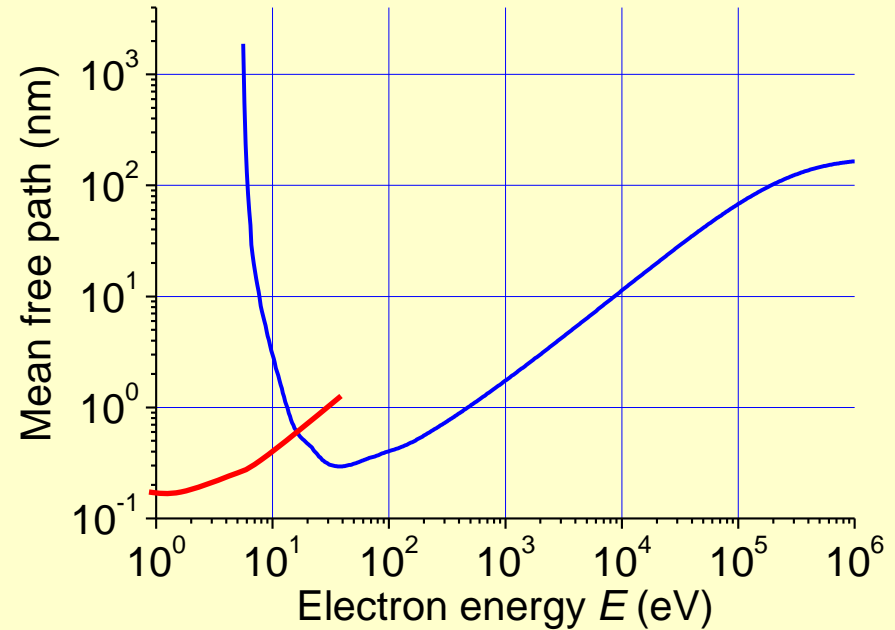
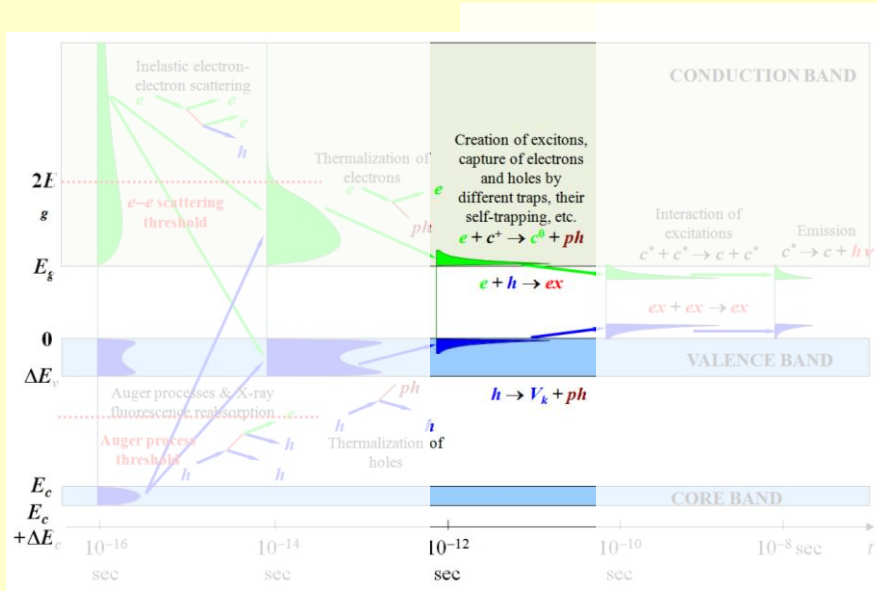
'Real' track structure



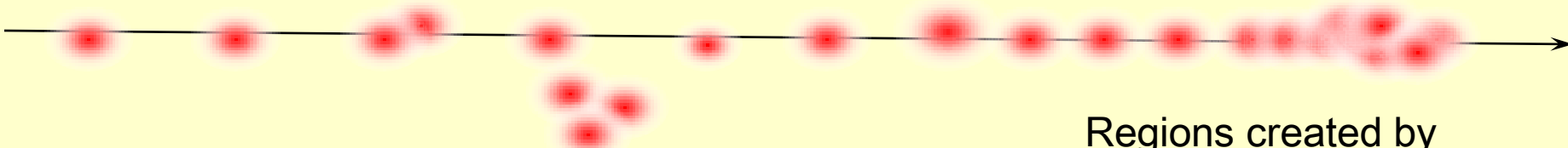
Regions created by different virtual photons are overlapped



Spatial track structure for e-h Onsager recombination stage for small thermalization radius



'Real' track structure



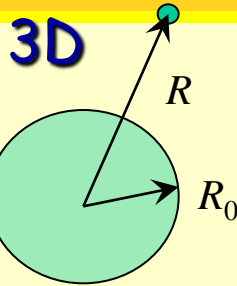
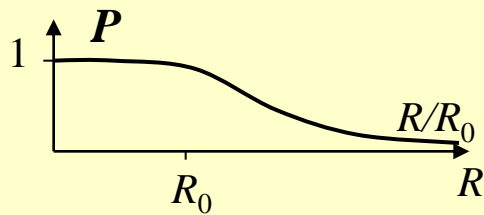
● Onsager radius 10 nm

Regions created by different virtual photons are overlapped



3D diffusion-controlled recombination

Recombination probability



Black sphere

$$P = \begin{cases} 1, & r_{eh} < R_0 \\ R_0/r_{eh}, & r_{eh} > R_0 \end{cases}$$

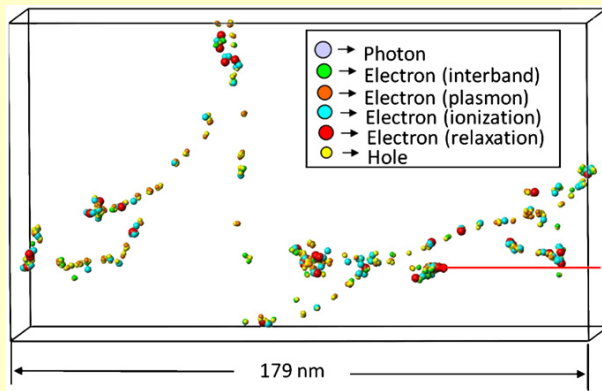
Coulomb $P = 1 - \exp(-R_{Ons}/r_{eh})$

$$\frac{e^2}{\epsilon R_{Ons}} = k_B T$$

$\epsilon=5.7$ $T=300K$ $R_{Ons}=10$ nm

$T=77K$ $R_{Ons}=38$ nm

$T=10K$ $R_{Ons}=300$ nm ???

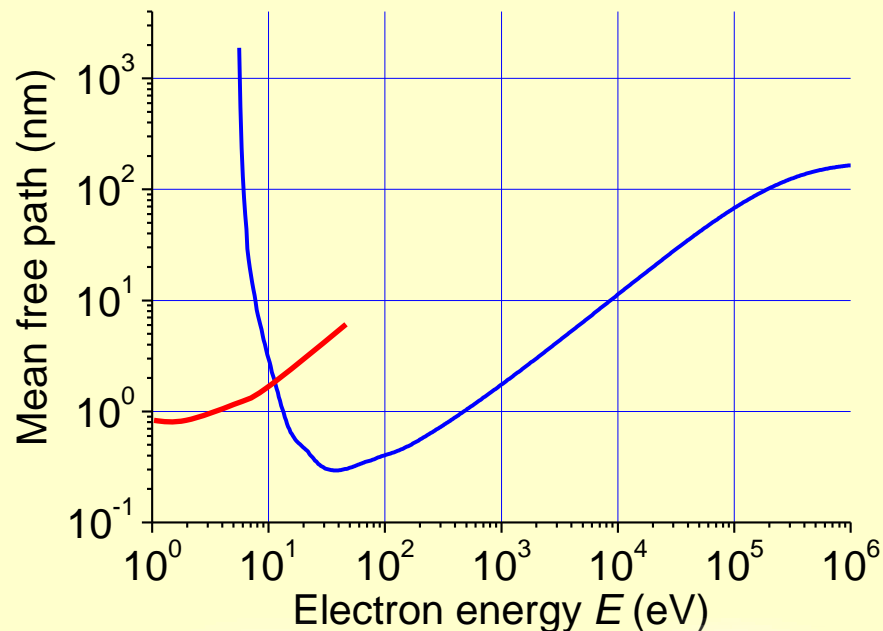
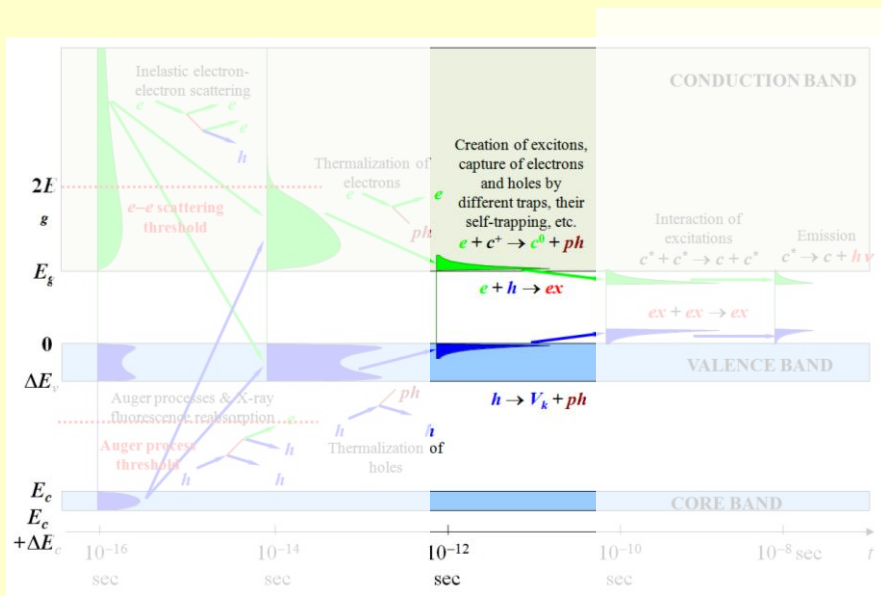


For thermalized excitations $R_{Ons}/r_{eh} \ll 1$
 – exciton yield after thermalization should be low

Simulated spatial distribution of e–h pairs for a 10 keV photon event in CsI, where electrons and holes are distinguished by size and color, as indicated in legend. NWEGRIM code.



Spatial track structure for e-h Onsager recombination stage for large thermalization radius

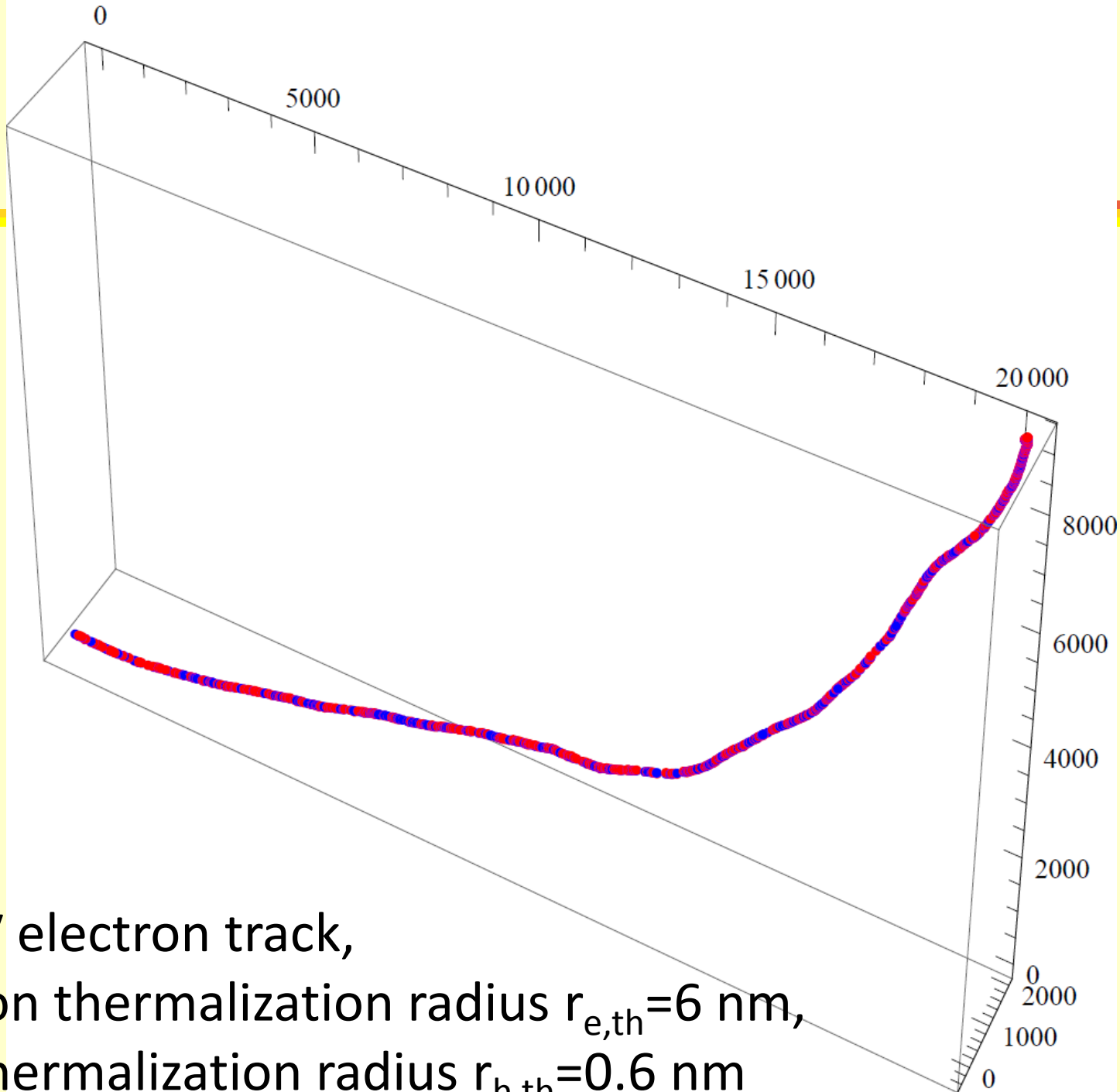


'Real' track structure

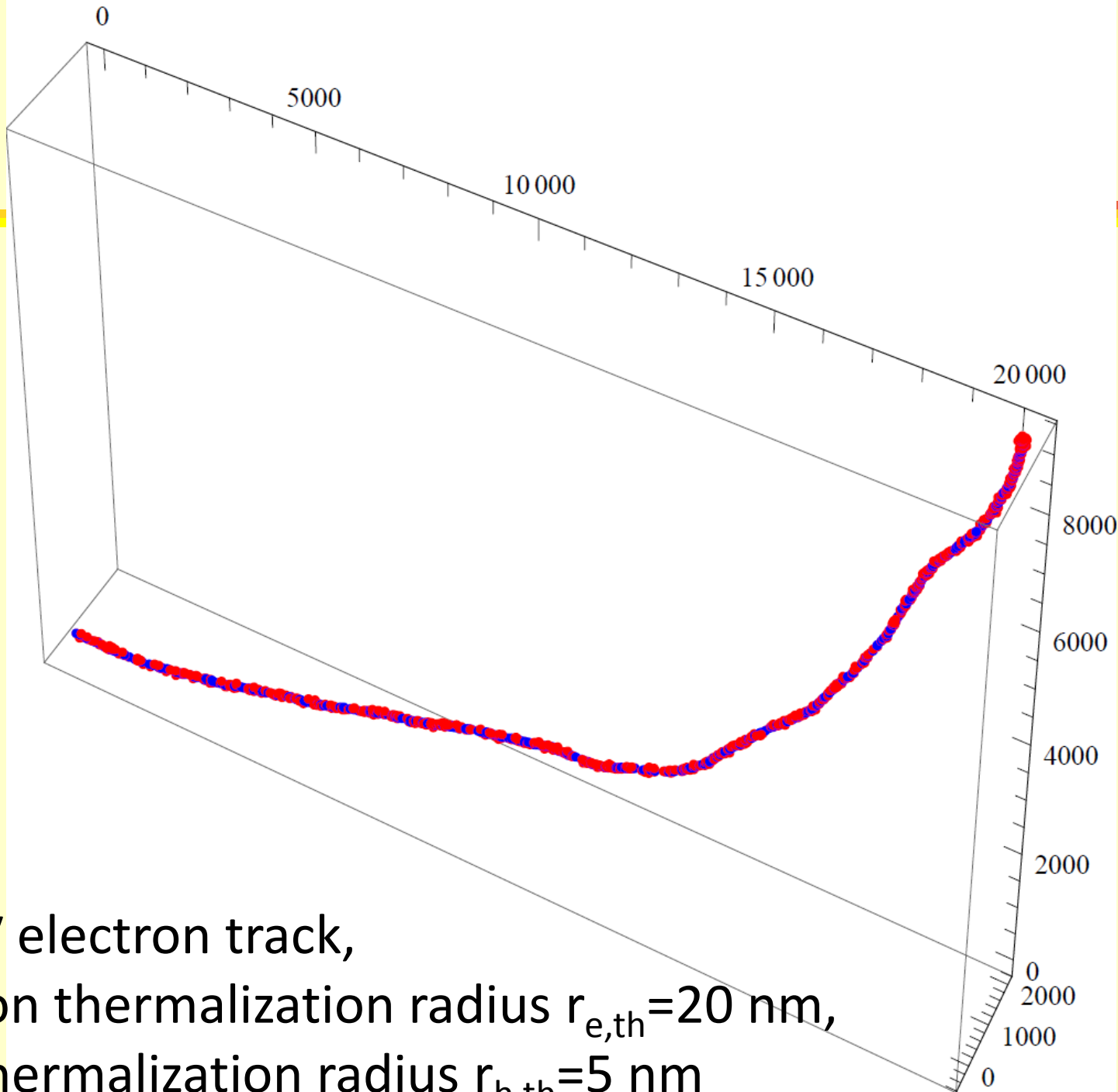
● Onsager radius 10 nm



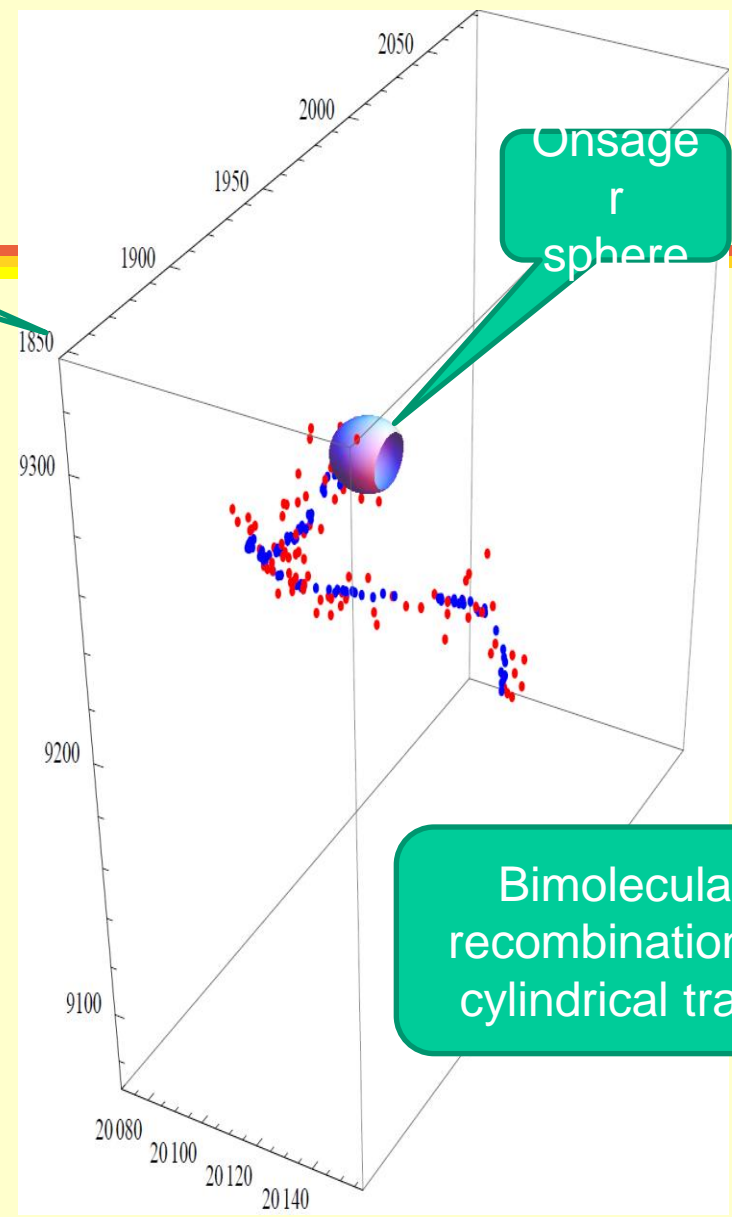
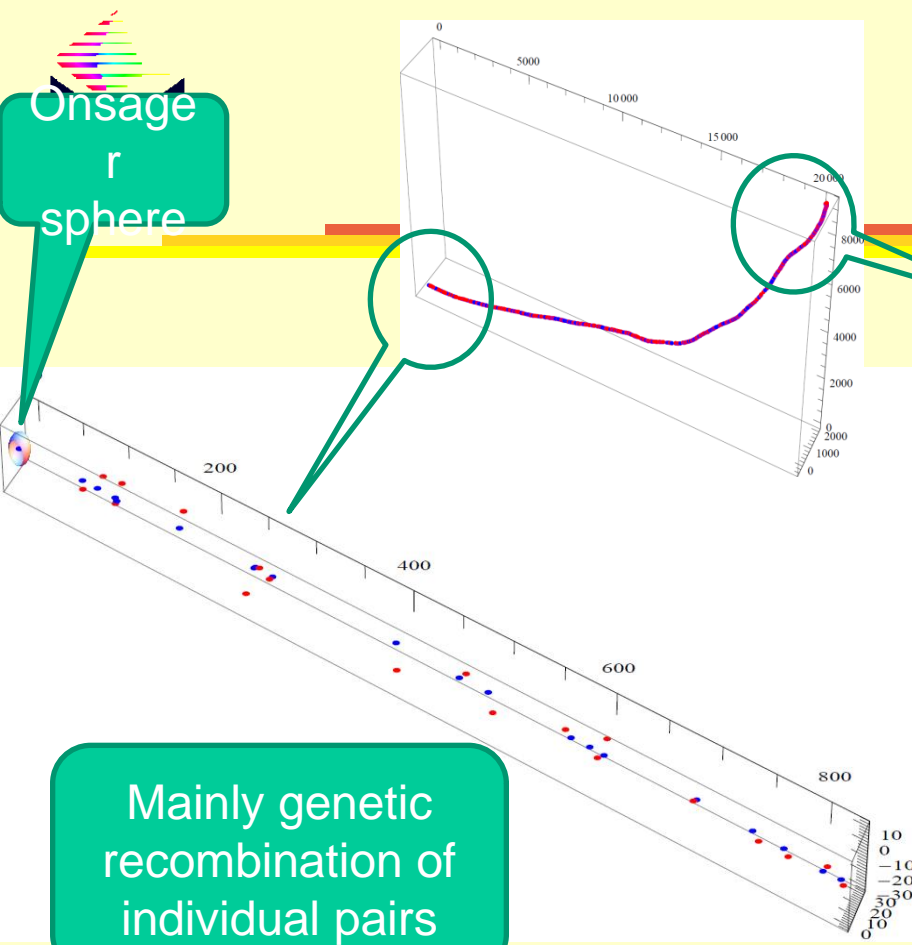
Example of structure of excited region after 30 keV electron passage



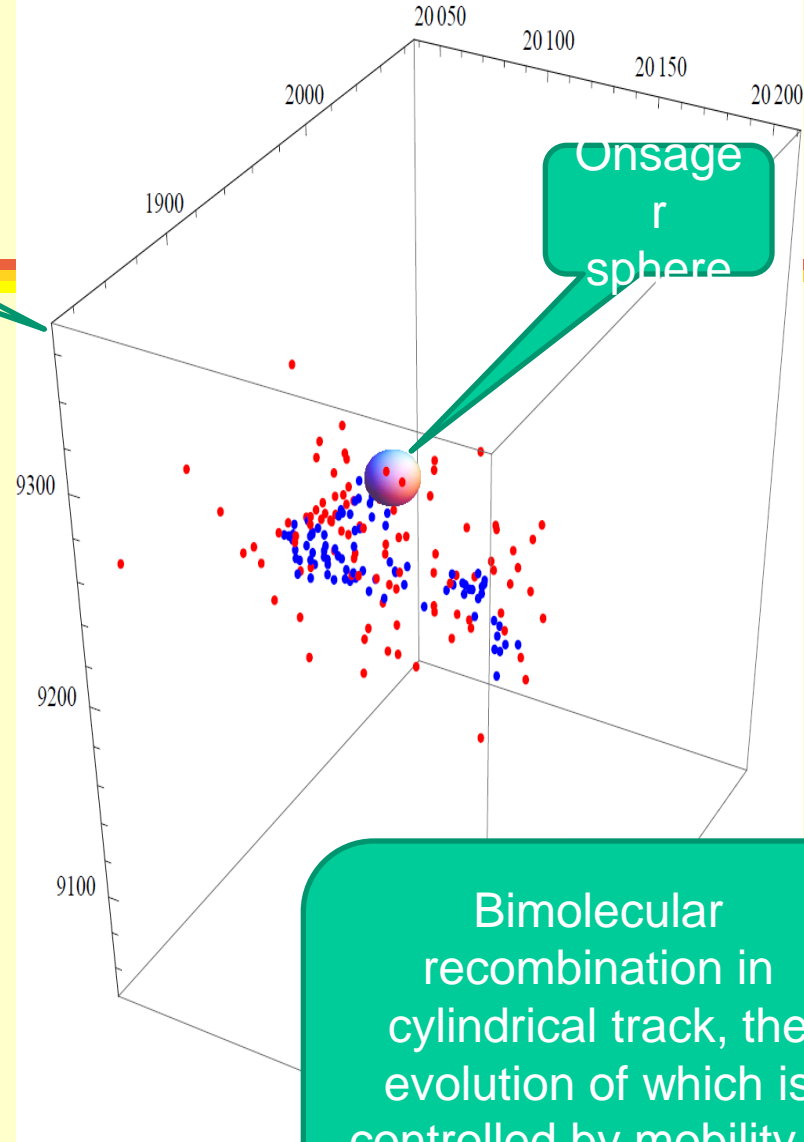
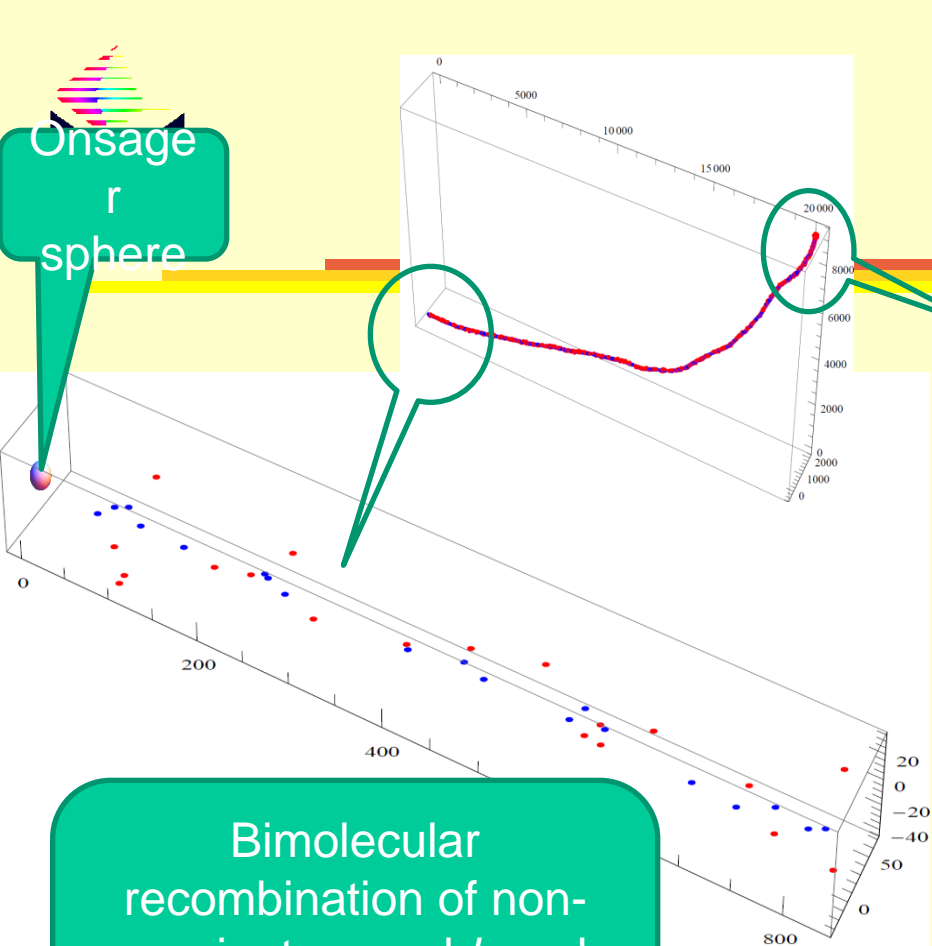
30 keV electron track,
electron thermalization radius $r_{e,th} = 6 \text{ nm}$,
hole thermalization radius $r_{h,th} = 0.6 \text{ nm}$



30 keV electron track,
electron thermalization radius $r_{e,th} = 20$ nm,
hole thermalization radius $r_{h,th} = 5$ nm



30 keV electron track,
 $r_{e,th} = 6 \text{ nm}$,
 $r_{h,th} = 0.6 \text{ nm}$ (red=e, blue=h)



30 keV electron track,
 $r_{e,th} = 20$ nm,
 $r_{h,th} = 5$ nm (red= e , blue= h)



Outline

Spatial scales for processes in scintillators

Nanoparticles as scintillators

Cascade, thermalization and recombination

Different types of mobilities

Thermalization length for different types of crystals

Interconnection of cascade, thermalization and recombination stages in binary iodides

Why cascade is so effective in CsI?

Thermalization length and impurities

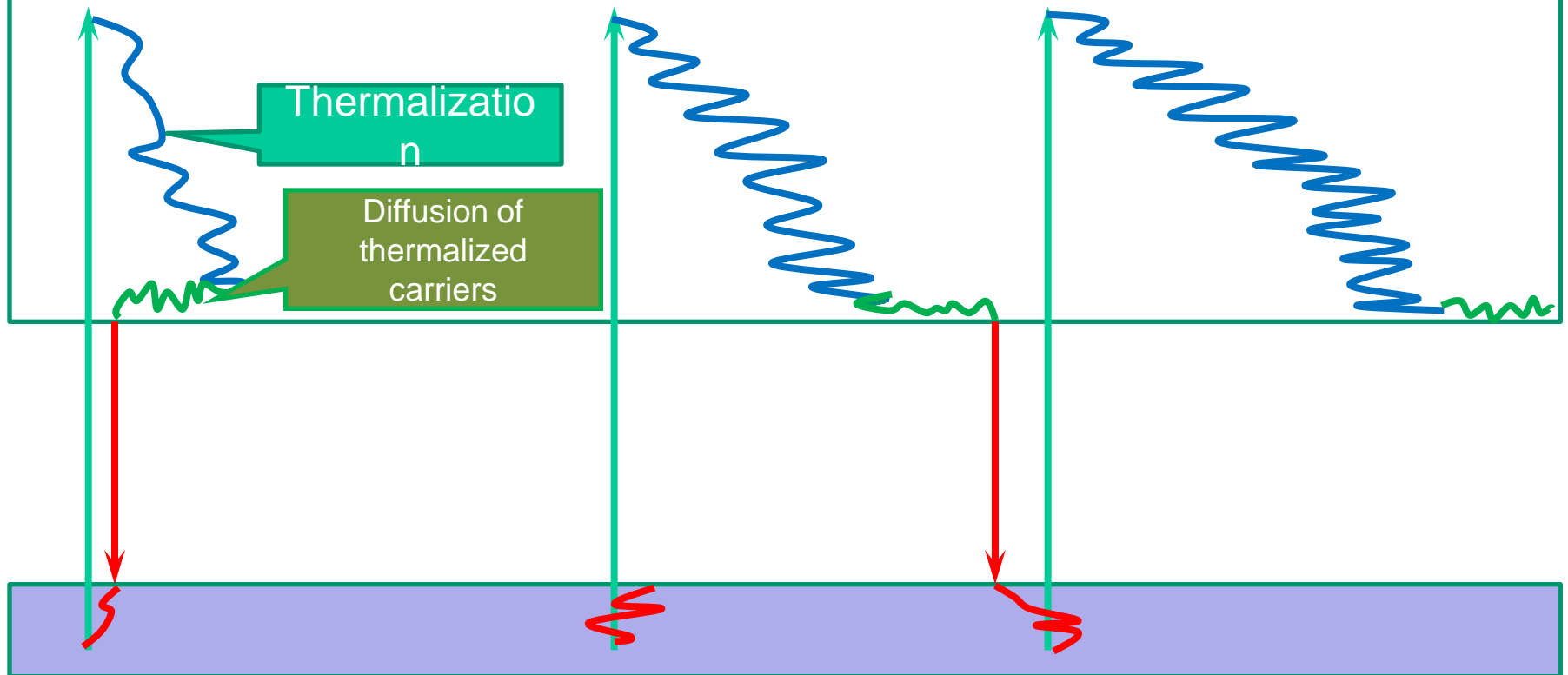
Concluding remarks



Electron-hole separation and recombination

Geminate recombination

Bimolecular recombination and escape

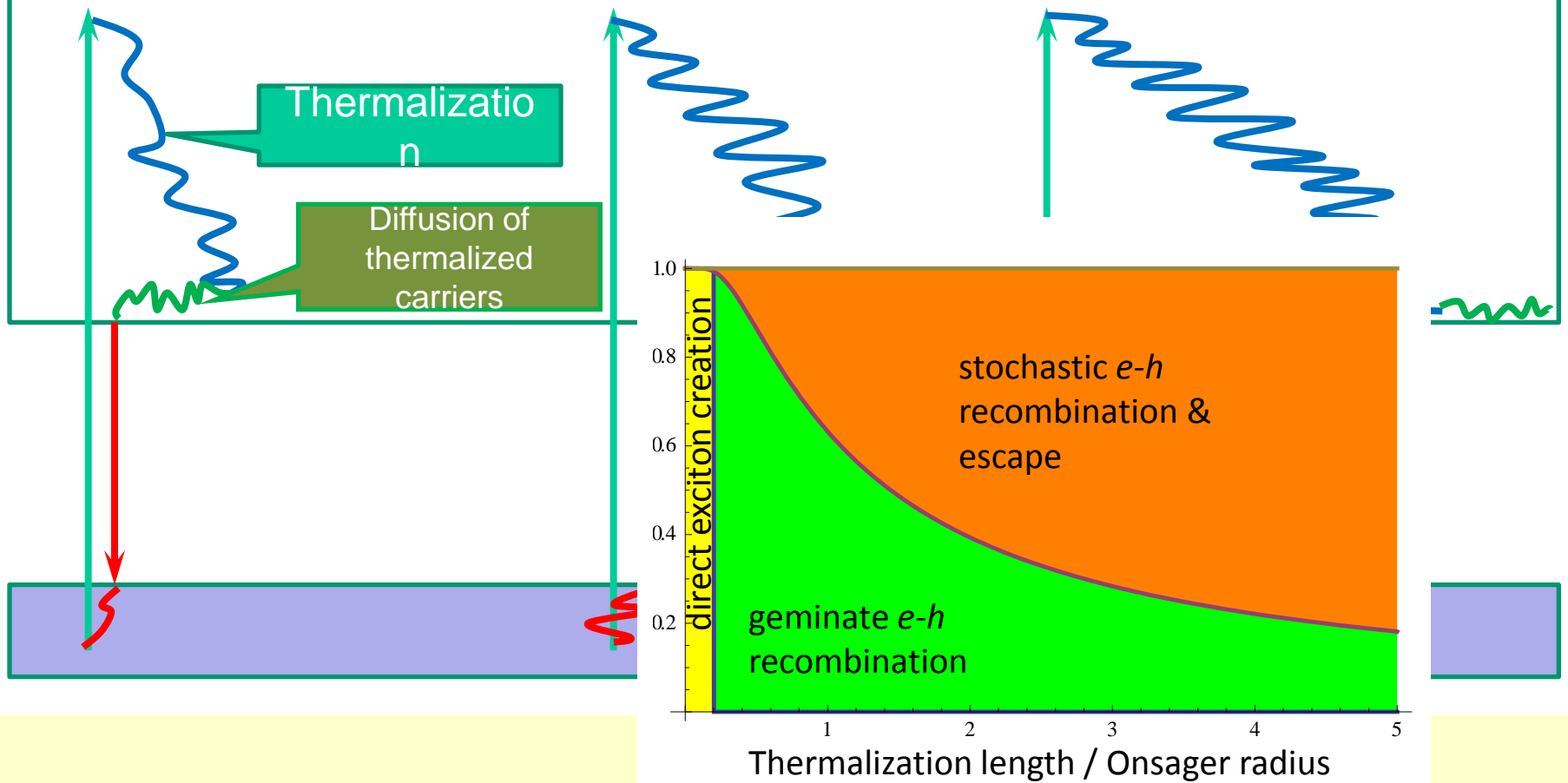




Electron-hole separation and recombination

Geminate recombination

Bimolecular recombination and escape

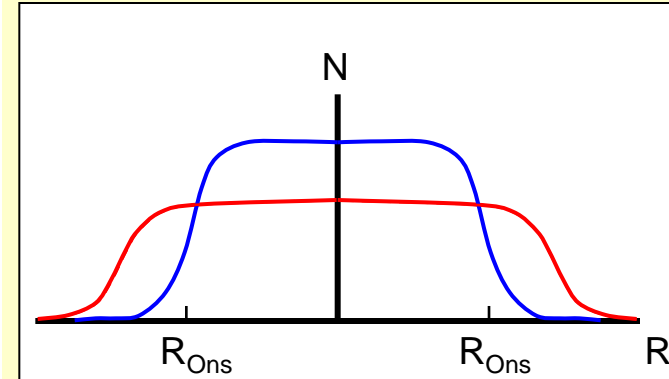




How we can manage thermalization length?

What we have to do to improve the yield?

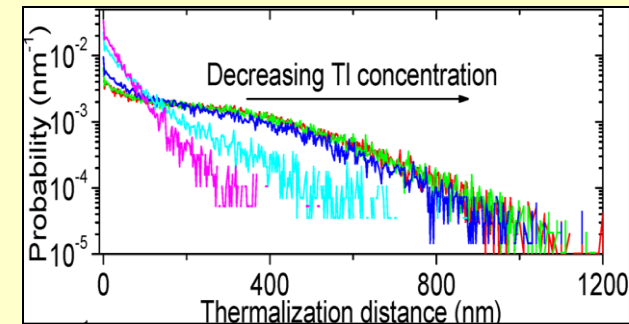
The goal is to concentrate e-h pairs at the distance less than Onsager radius, to minimize the volume of stochastic recombination and escape losses.



Two ways for e-h separation management

Doped/activated crystals (rare solutions)

Mixed crystals (hard solutions)



Z. Wang, Y. Xie, B. D. Cannon... 2011



Coupled processes of thermalization and spatial diffusion

Mean square of the thermalization distance $\langle r^2 \rangle_{E_{e0} \rightarrow E_e^{kin}} = 6 \int_{E_e^{kin}}^{E_{e0}} \frac{D^R(E')}{S(E')} dE'$

Spatial distribution function $f(r, l_e(E_{e0})) = \frac{3\sqrt{6} r^2}{\sqrt{\pi} l_e^3(E_{e0})} \exp\left(-\frac{3r^2}{2l_e^2(E_{e0})}\right)$

$$l_e(E_{e0}) = \sqrt{\langle r^2 \rangle_{E_{e0} \rightarrow k_B T}}$$

where thermalization length is

Thermalization length for one LO phonon branch

$$\begin{aligned} l_{e,LO}^2(E_{e0}) &= \frac{8}{3} a_B^2 \left(\frac{\tilde{\epsilon}}{m_e^*/m_0} \right)^2 \tanh\left(\frac{\hbar\Omega_{LO}}{2k_B T}\right) \int_{\hbar\Omega_{LO}}^{E_{e0}} \left(\frac{E'}{\hbar\Omega_{LO}} \right)^2 \frac{1}{\ln(4E'/\hbar\Omega_{LO})} \frac{dE'}{\hbar\Omega_{LO}} \\ &= \frac{1}{24} a_B^2 \left(\frac{\tilde{\epsilon}}{m_e^*/m_0} \right)^2 \tanh\left(\frac{\hbar\Omega_{LO}}{2k_B T}\right) \text{Ei}\left(3 \ln\left(\frac{4E_{e0}}{\hbar\Omega_{LO}}\right)\right), \end{aligned}$$

We have to choose/engineer materials with

- higher effective masses in the whole relaxation region $E_{kin} < E_g$
- higher LO phonon energies



Spatial distribution of electrons, holes and excitons due to mobility in e-e passive energy domain

Two types of carrier mobilities: **thermalization length** (mobility of hot electrons and holes) and **mobility of thermalized excitations** (electrons, holes & excitons).

High-energy part of ionization track – individual electron-hole pairs and small non-overlapping clusters of excitations. **Negative role of mobility**: the higher **thermalization length** (in comparison with Onsager radius), the lower the recombination yield.

Low-energy part of ionization track – overlapping clusters of excitations. Mean distance between interacting excitations increases with increase of the mobility of excitons. **Positive role of mobility**: the higher the mobility, the lower the quenching of excitation due to high EE density.

“Ideal” scintillator: **Low hot mobility** (high yield of excitons) and **high thermalized mobility** (low interaction).



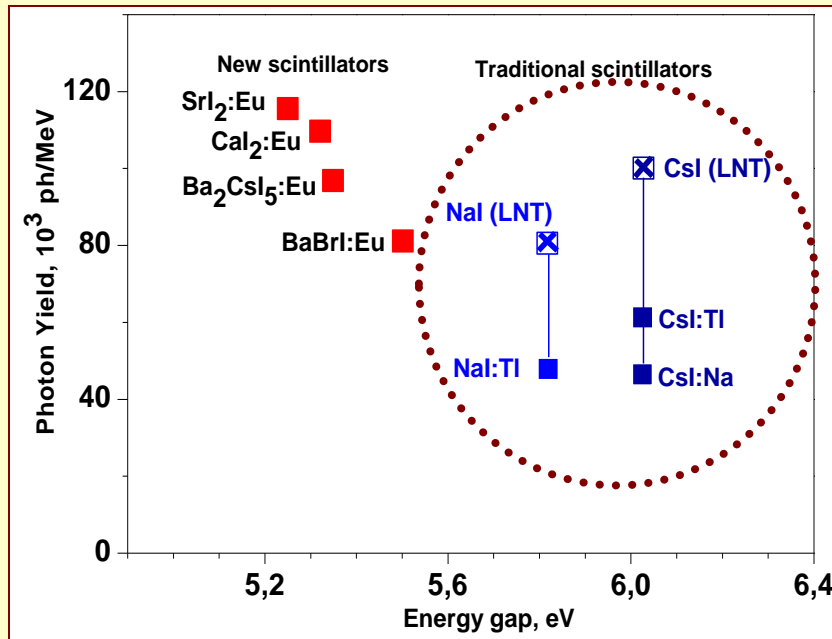
Binary alkali halides:

Can the yield achieve theoretical limit?

Can we improve conventional alkali halide scintillators?



Fundamental limits to Light Yield of NaI and CsI based scintillators (LY vs. E_g)



Crystal	E_g , eV	LY, ph/MeV theor.	LY, ph/MeV expim.
NaI (77K)	5.8	86.000	80.000
NaI:TI (RT)			45.000
CsI (77K)	6.1	82.000	~100.000
CsI:TI (RT)			56.000
CsI:Na (RT)			46.000

Experimental data are far from theoretic limit for NaI and CsI based crystals

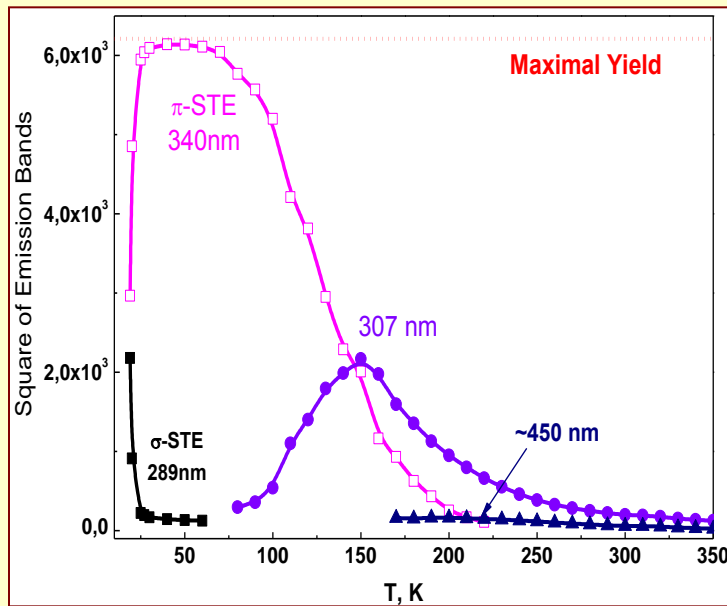
✓ Pure NaI and CsI possess extremely high photon yield at LNT

[V.Sciver, 1958; Persyk, 1980; Moszynski et al, 2010]

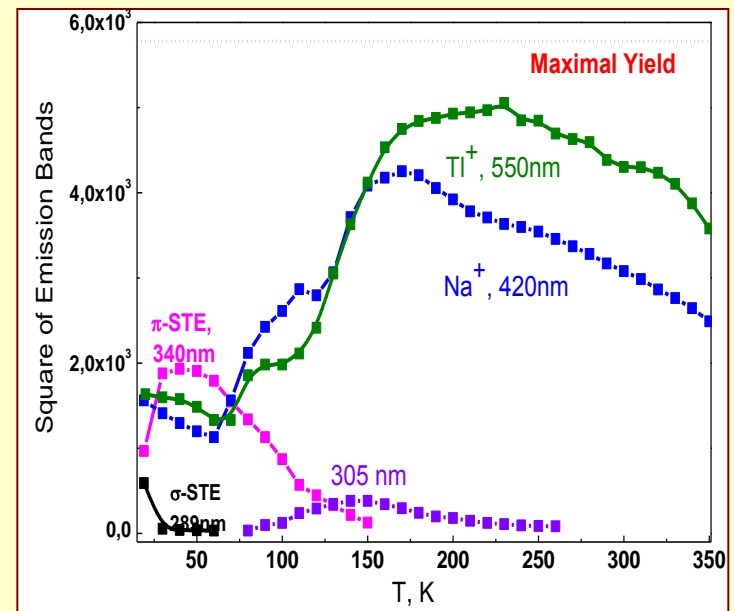


Exciton based luminescence

Pure CsI emission



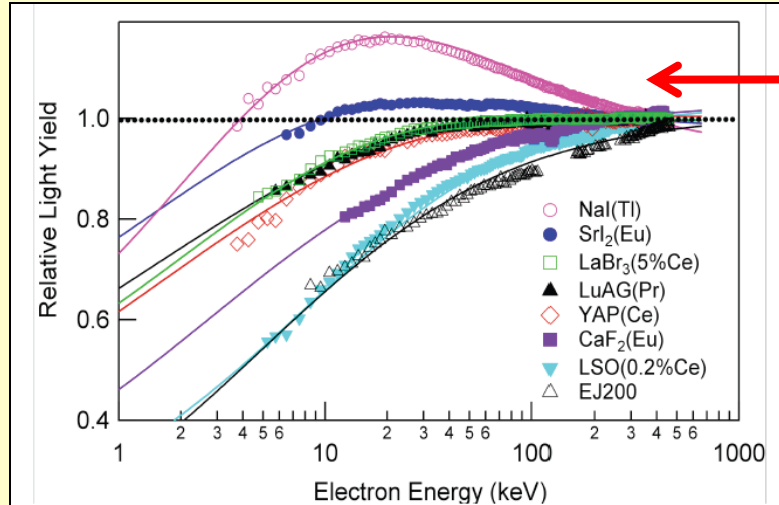
CsI:Na and CsI:TI emission



- Resume :**
- * It is possible to separate different types of emission
 - * STE and defect trapped exciton emissions are dominated
 - * Self trapping creates the best conditions for maximal yield
 - * Other relaxation mechanisms lead to an extra efficiency losses



Non-proportionality analysis for alkali halides



The high-energy decrease of the scintillator efficiency shows that significant fraction of individual electron-hole pairs are thermalized at distances larger than Onsager radius

$$p = 1 - \exp(-R_{Ons}/r_{eh})$$

$$\frac{e^2}{\epsilon R_{Ons}} = k_B T$$

Electron response of some scintillators

[S. A. Payne, W. W. Moses et al., *IEEE TNS*, 2011]

- 1) We can increase Onsager radius – by decreasing the temperature. Pure CsI and NaI have yield $\sim 100,000$ ph/MeV at 77K ! ($R_{ons,77K} = 4R_{ons,300K}$)
- 2) We can decrease thermalization distances - by choosing of complex halides.

What is the physics of the thermalization distances decrease in this case?

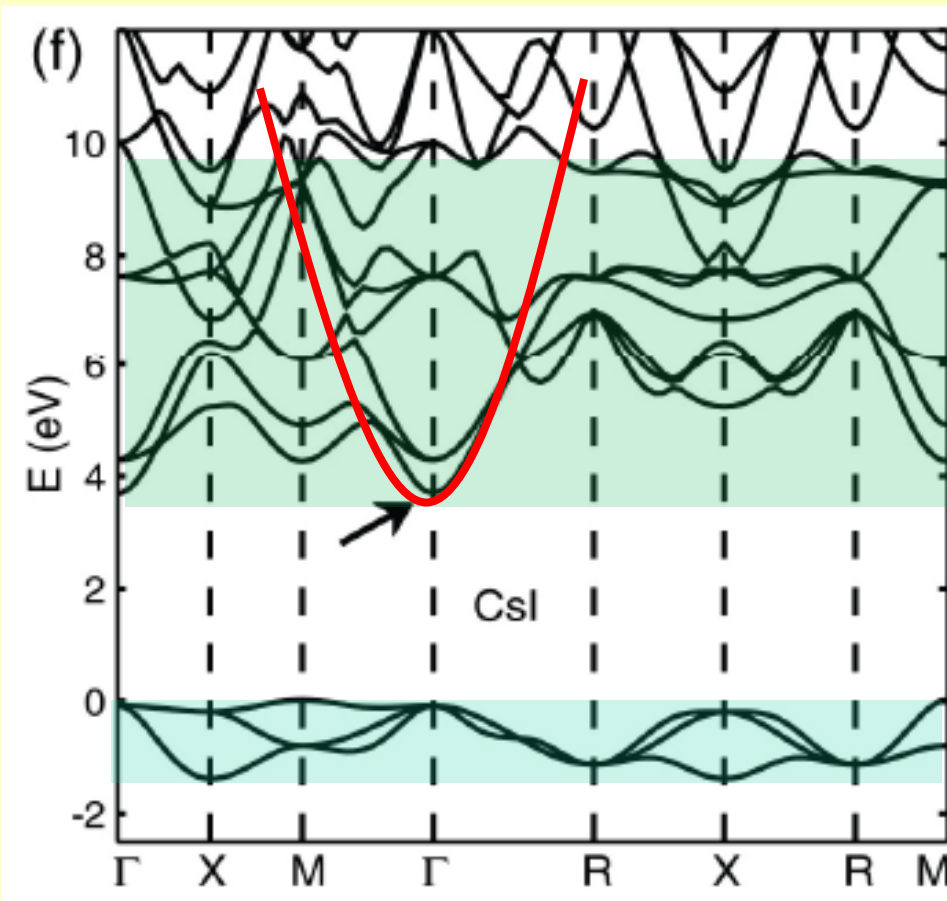
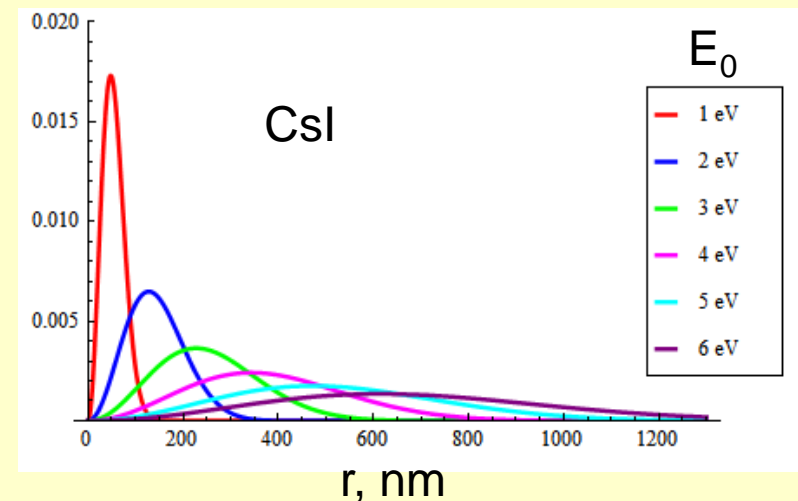
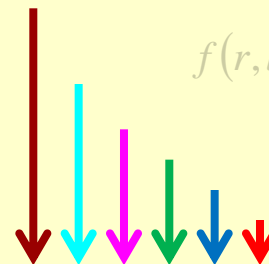


Starting states for thermalization ($E_{\text{kin}} < E_g$)

Thermalization for parabolic band and one LO phonon starting from energy E_0

$$l_{e,LO}^2(E_{e0}) = \frac{1}{24} a_B^2 \left(\frac{\xi}{m_e^*/m_0} \right)^2 \tanh \left(\frac{\hbar \Omega_{LO}}{2k_B T} \right) \text{Ei} \left(3 \ln \left(\frac{4E_{e0}}{\hbar \Omega_{LO}} \right) \right),$$

$$f(r, l_e(E_{e0})) = \frac{3\sqrt{6} r^2}{\sqrt{\pi} l_e^3(E_{e0})} \exp \left(-\frac{3r^2}{2l_e^2(E_{e0})} \right)$$



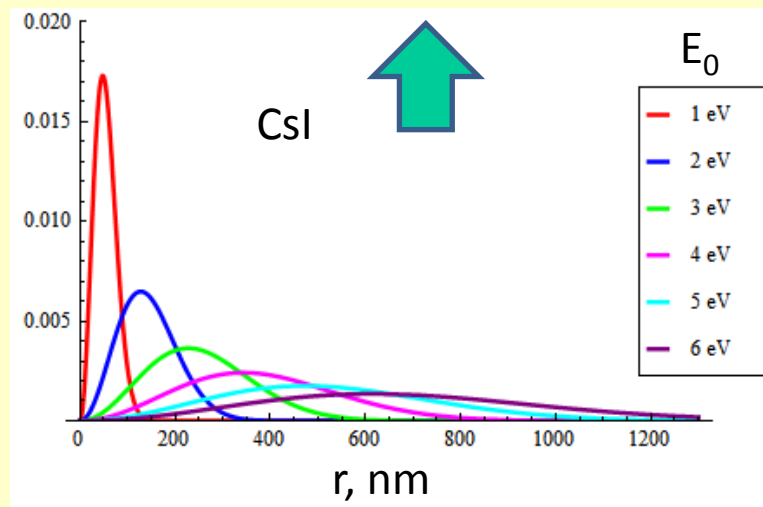
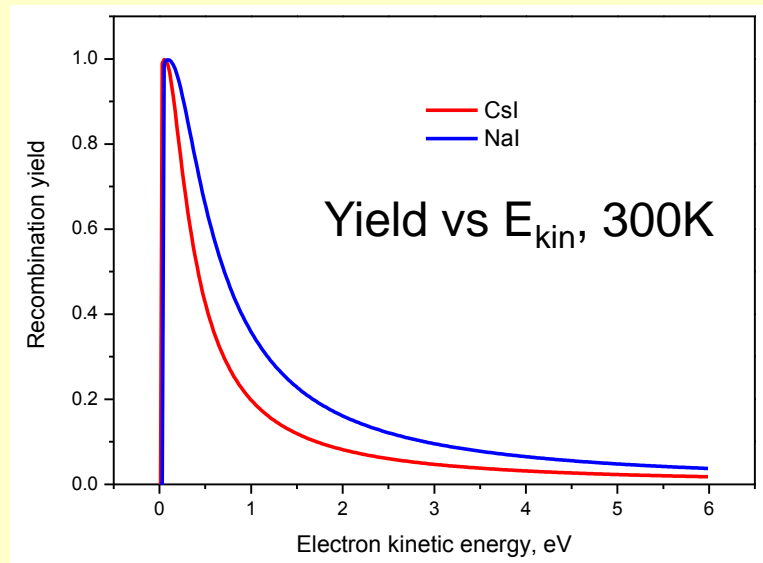
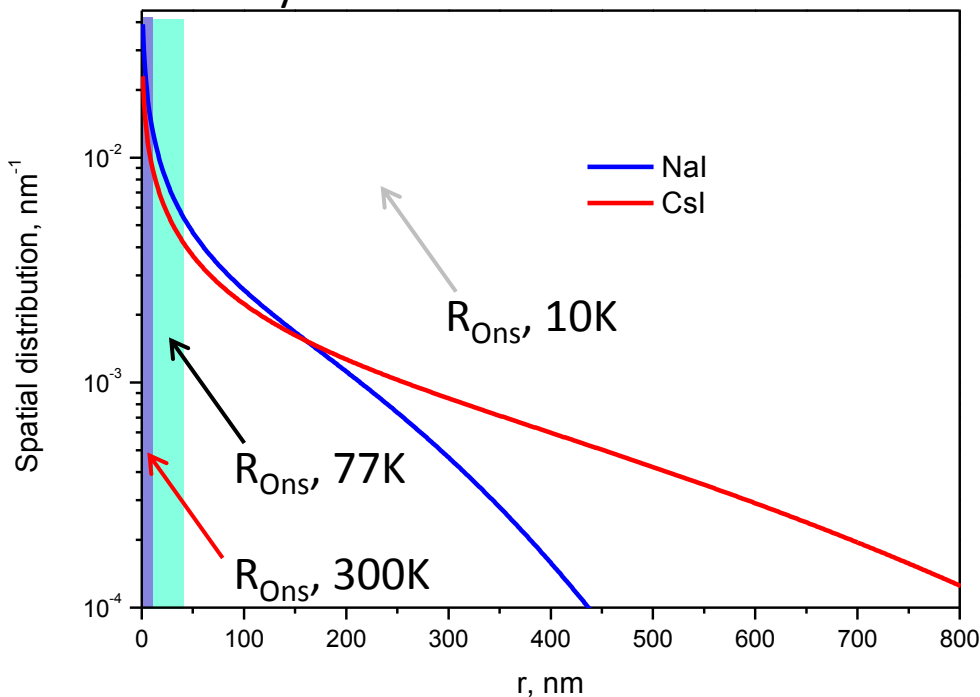
Band structure calculations

from W. Setyawan, R. M. Gaume et al. *IEEE TNS*, 2009



Spatial distribution of thermalized electrons (binary crystals)

Analytical estimation

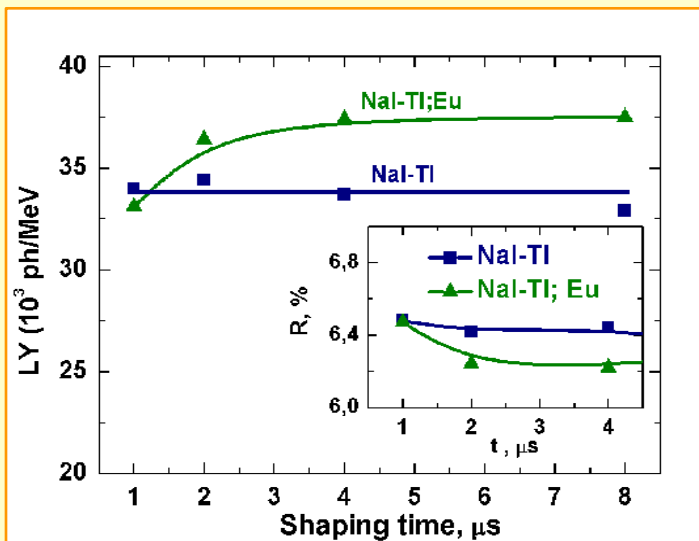


	$R_{O_{ns}}, 300K$	Yield, 300K	Yield, 77K
CsI	9.87 nm	0.24	0.44
NaI	9.05 nm	0.34	0.58

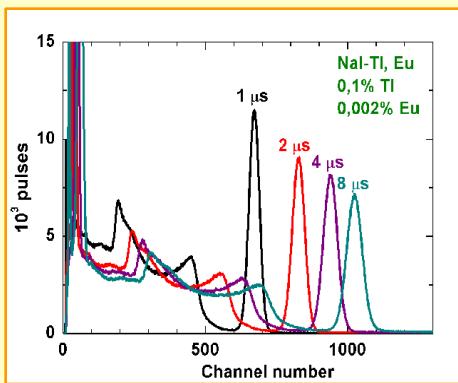
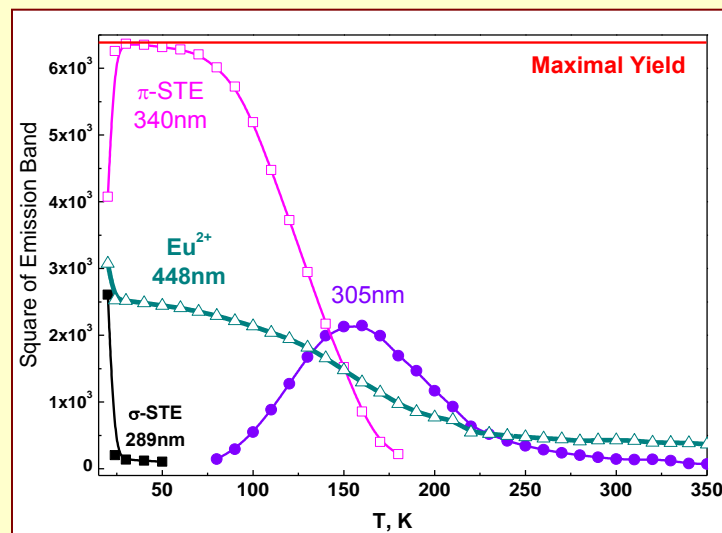


Can we use co-doping to collect more excitations?

Nal :Ti, Eu



CsI:Eu emission vs temperature



Eu, Ti – co-doping of Nal increases light yield comparing to conventional Nal:Ti



Scintillation efficiency improvement

“Synergy” of Eu and Tl co-doping in NaI crystals

Crystal	Tl, m%	Eu, m%	Lum, nm	Decay, ns	LY, %	R, %
NaI-Tl	$1 \cdot 10^{-1}$		415	230	100	6.4
NaI-Tl, Eu	$1 \cdot 10^{-1}$	$1 \cdot 10^{-3}$	445	230 (26%) 1000 (74%)	110	6.2
NaI-Eu		$1 \cdot 10^{-1}$	445	1000	60	9.5

- Introduce of hundred times lower Eu-concentration ($10^{-3}\%$) than Tl (0.1%) leads to Tl emission suppression and higher light output.
- Thus, Eu co-doping allows to get better scintillation performance of NaI:Tl.
- It costs some losses at decay and claim for larger integration time.



Binary alkali halides - the yield could not reach fundamental limit !

- ✓ The maximal CsI or NaI scintillation yield corresponds to STE relaxation at LNT
- ✓ Temperature rise leads to the STE luminescence quenching and DTE emission that lower due to transfer and stabilization losses. Eu^{2+} co-doping allows slightly increase the NaI(Tl) yield only
- ✓ Thermalization length is much higher than Onsager radius in alkali halides – therefore **geminate recombination yield is much less than unity.**
- ✓ We can decrease thermalization distances – by choosing of complex halides



What materials are characterized by lower thermalization length? Why **Alkali Earth Halides**?

How thermalization length depend on material parameters?

$$l_{e,LO}^2(E_{e0}) = \frac{1}{24} a_B^2 \left(\frac{\tilde{\varepsilon}}{m_e^*/m_0} \right)^2 \tanh\left(\frac{\hbar\Omega_{LO}}{2k_B T} \right) \text{Ei}\left(3 \ln\left(\frac{4E_{e0}}{\hbar\Omega_{LO}} \right) \right),$$

We should choose materials with

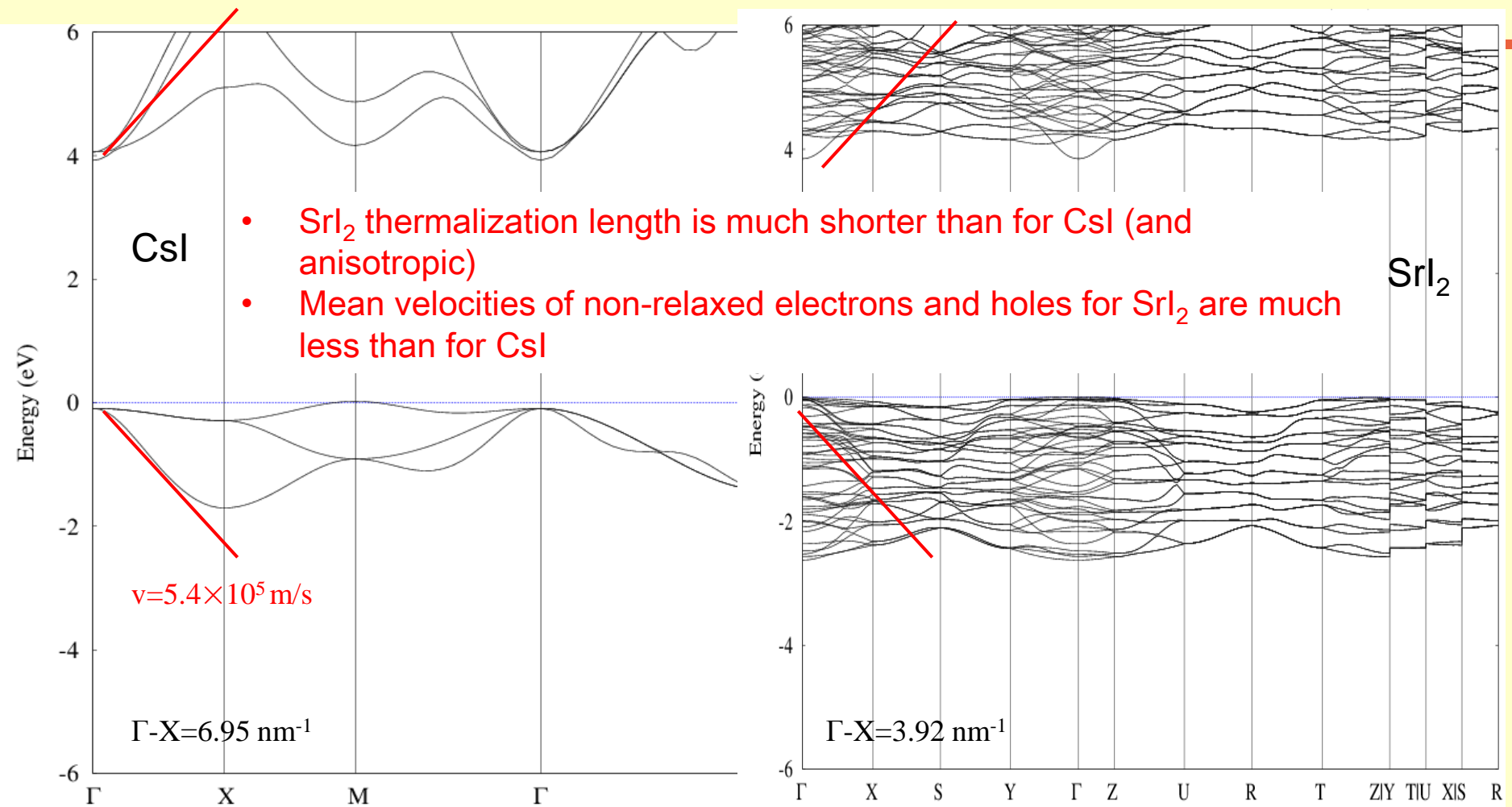
- **higher effective masses** in the whole relaxation region $E_{\text{kin}} < E_g$ and
- higher LO phonon energies

Why AE halides? - Lower thermalization length in complex halides result in higher yield of geminate recombination!

Complex halides with many atoms in elementary cell (e.g. SrI_2 with 24 atoms in elementary cell)



Comparison of CsI and Srl₂ electronic structures



[1] W. Setyawan and S. Curtarolo, *Comp. Mat. Sci.* 49, 299 (2010).

[2] S. Curtarolo, W. Setyawan, S. Wang et al. *Comp. Mat. Sci.* 58, 227 (2012).

[3] W. Setyawan, R. M. Gaumé, S. Lam et al. *ACS Comb. Sci.* 13(4), 382 (2011).



The model resume

For all complex halides we can obtain high yield due to high yield of geminate recombination and small bimolecular effects in non-proportionality curves

Lower thermalization length (in comparison with alkali halides) is connected with

- flat bands in whole energy thermalization region ($E_{\text{kin}} < E_g$)
and
- much more complicated LO phonon structure (and probably strongly anisotropic mobility due to layer structure)



Renaissance of Eu-doped scintillators (history and reality)

Luminescence study 1948 - 1975	
LiCl :Eu	Lehmann, 1975
LiI :Eu	Murray, 1958
CaI ₂ :Eu	Hofstadter, 1963 Lyskovich, 1970
CaF ₂ :Eu	Butement, 1948
SrCl ₂ :Eu	Lehmann, 1975
SrBr ₂ :Eu	
SrI ₂ :Eu	
SrI ₂ :Eu <i>scintillator</i>	Hofstadter, 1968, US Patent, 3373279

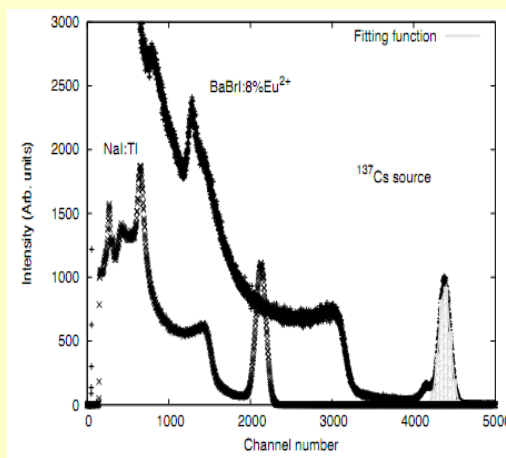
New demands have led to discovery
several new Eu-doped scintillators



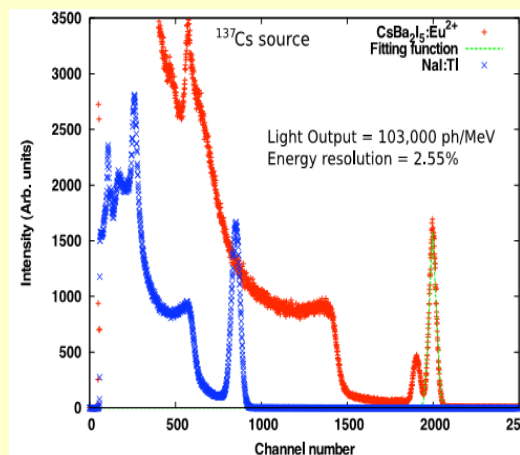
New scintillators 2007 - 2011	
CaI ₂ :Eu	LLNL, LBNL, USA Cherepy, Moses et al. 2007 - 2009
SrI ₂ :Eu	
Ba ₂ CsI ₅ :Eu	
BaBrI:Eu	LBNL, USA Bourret, Derenzo et al. 2010
BaFI:Eu	
SrCsI ₃ :Eu	SMRC, Tennessee, USA, Zhuravleva, Melcher et al. 2010
CsEuI ₃	
Cs ₃ EuI ₅	



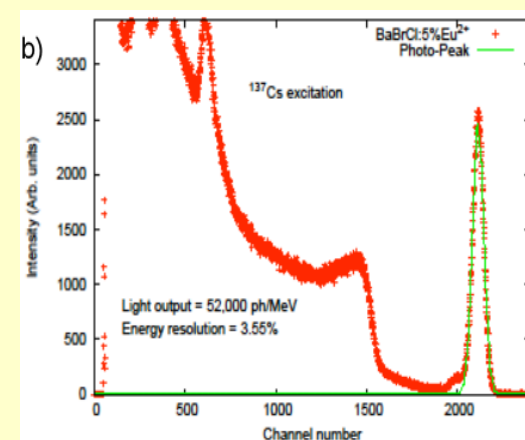
Pulse height spectra of record scintillators under γ ^{137}Cs excitation



BaBrI:Eu



CsBa₂I₅:Eu



BaBrCl:Eu

LBL, USA

Bourret, Derenzo, Bizarri et al. 2009-2012



AE scintillator performance progress (2007→2012)

Crystal	2007 - 2009		2011 - 2012	
	LY ph/Mev	R, % Cs ¹³⁷	LY ph/Mev	R, % Cs ¹³⁷
SrI ₂ :Eu	115.000	2.6	115.000	2.6
Ba ₂ CsI ₅ :Eu	97.000	3.8	102.000	2.55
SrCsI ₃ :Eu	65.000	5.2	73.000	3.9
BaBrI:Eu	81.000	4.8	97.000	3,4

Many AE halides possess with efficiency about fundamental limit

Selection of one (best) scintillator has to base on the technology advantages

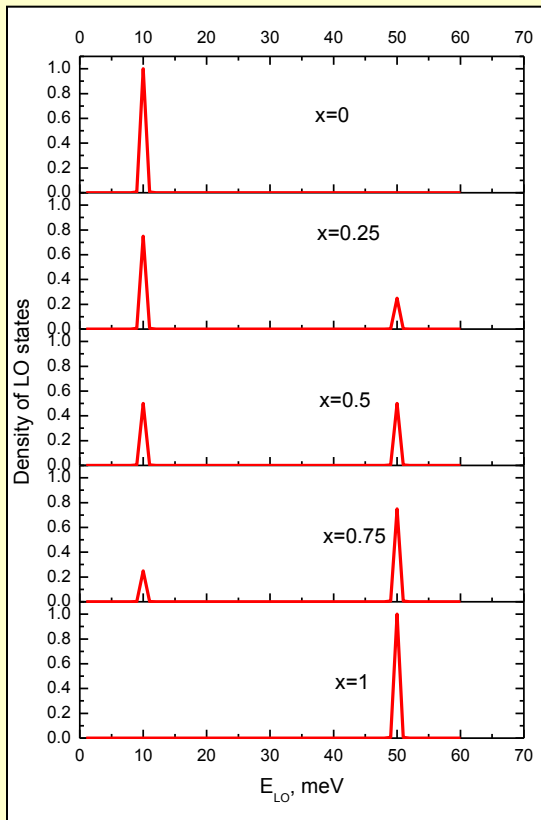


Modification of phonon spectrum (additional phonon branches)

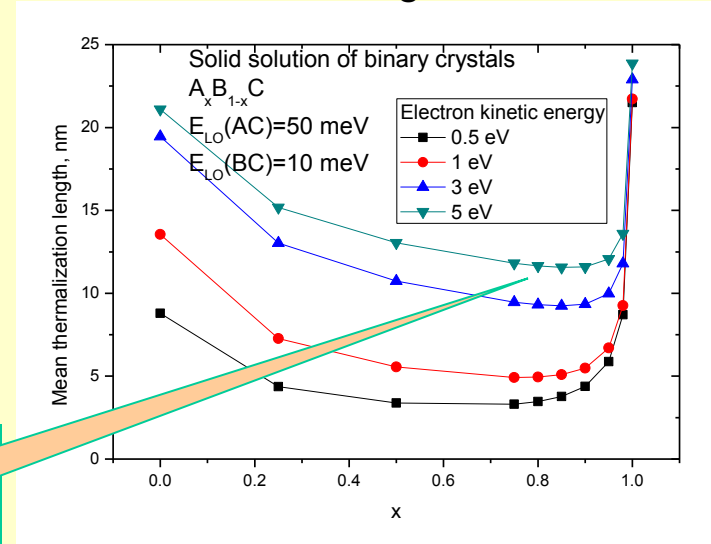
Mixed – $A_xB_{1-x}C$ crystal model

Mean thermalization length vs X concentration

Density of LO states

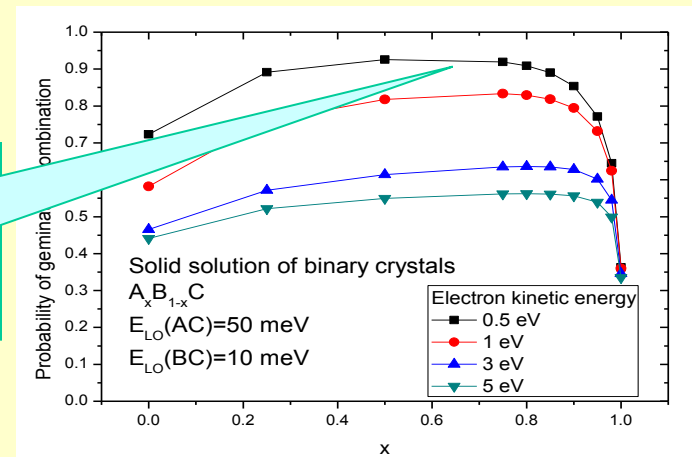


Thermalization length decrease



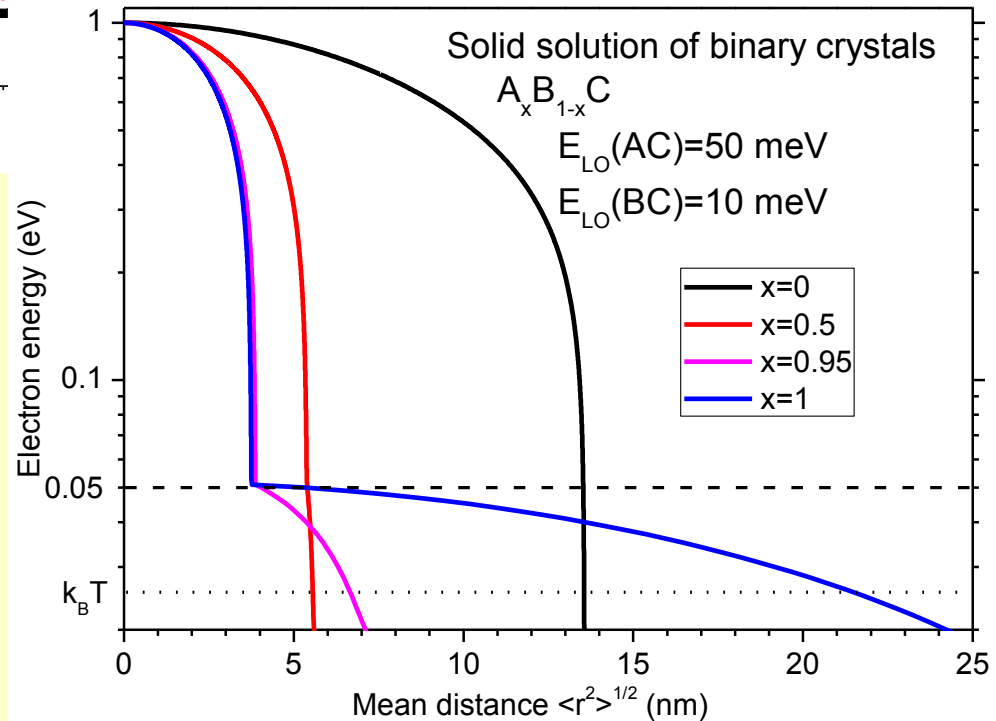
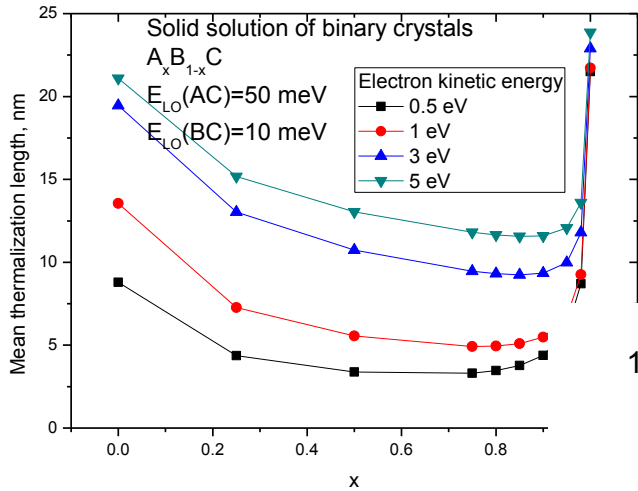
Recombination probability increase

Probability of geminate recombination



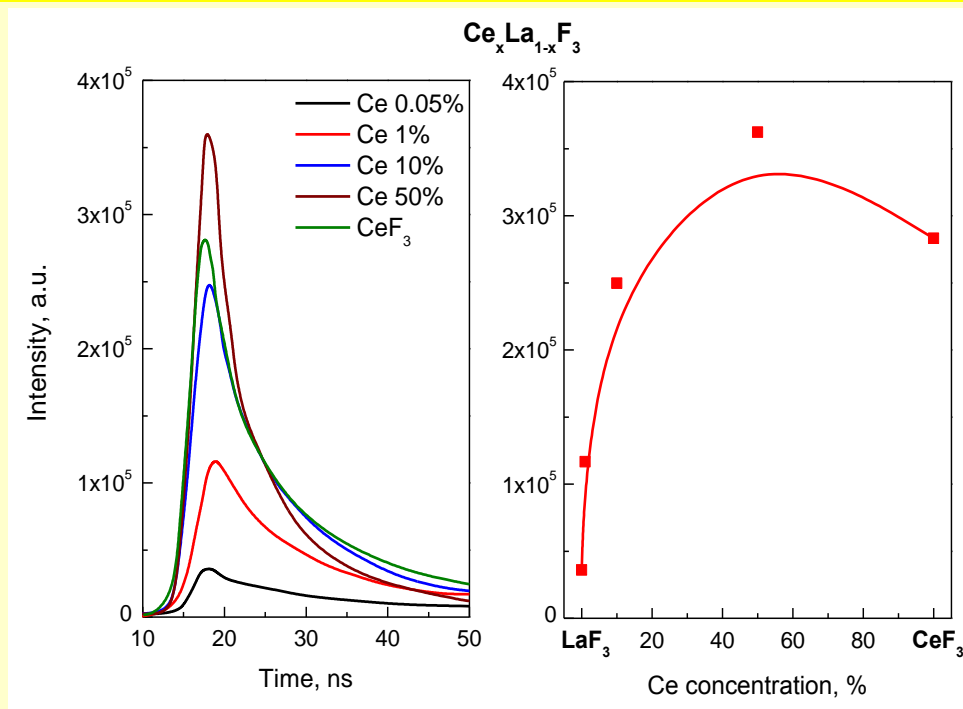


Modification of electron spectrum – increasing of elastic scattering





Mixed fluorides: $\text{Ce}_x\text{La}_{1-x}\text{F}_3$



A.N. Belsky, A.V. Gektin et al.,
Proceedings of SCINT'95, Delft,
1995

Pulse shape and decay kinetics of $\text{Ce}_x\text{La}_{1-x}\text{F}_3$

X-ray excitation (10 keV).

Left – original linear scale data; right –
intensity vs cation mixture rate.



Mixed halides... 20 years late

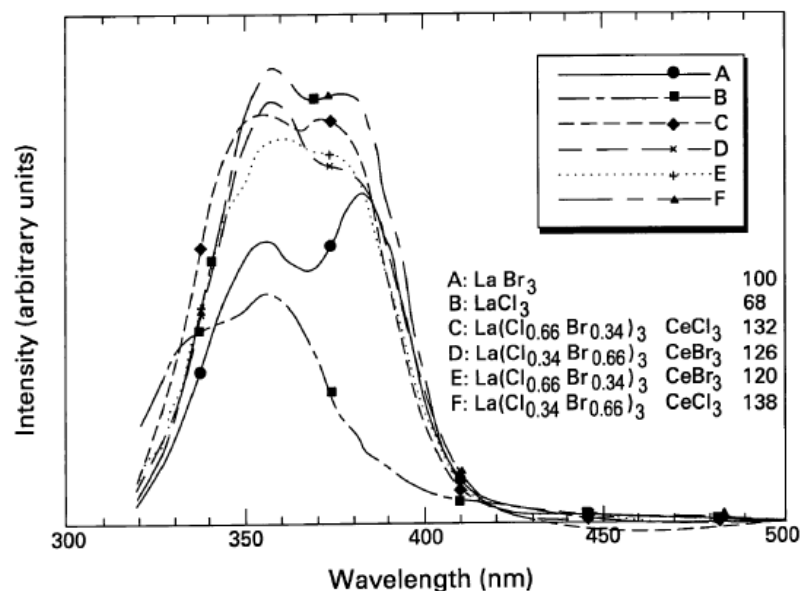
(54) SCINTILLATOR COMPOSITIONS, AND RELATED PROCESSES AND ARTICLES OF MANUFACTURE

Srivastava et al.

(45) Date of Patent:

Aug. 1, 2006

GE Research



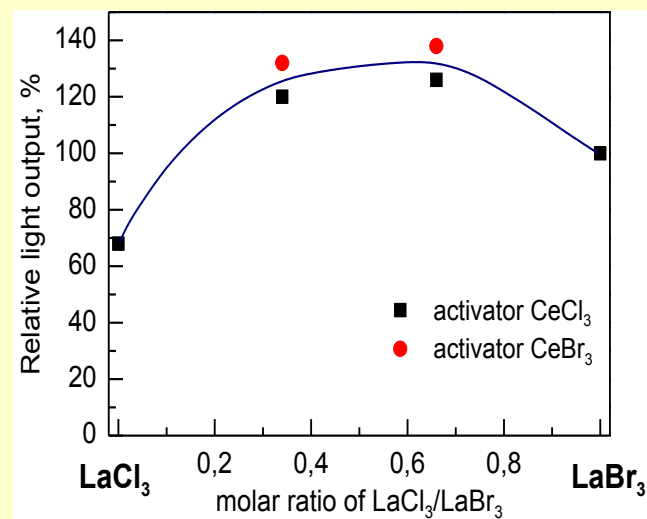
Emission spectra under UV excitation

TABLE 1

SAMPLE	COMPOSITION	ACTIVATOR	LIGHT OUTPUT*
A**	LaBr ₃	—	100
B**	LaCl ₃	—	68
C	La(Cl _{0.66} Br _{0.34}) ₃	CeCl ₃	132
D	La(Cl _{0.34} Br _{0.66}) ₃	CeBr ₃	126
E	La(Cl _{0.66} Br _{0.34}) ₃	CeBr ₃	120
F	La(Cl _{0.34} Br _{0.66}) ₃	CeCl ₃	138

*Relative percent for samples B–F, as compared to sample A.

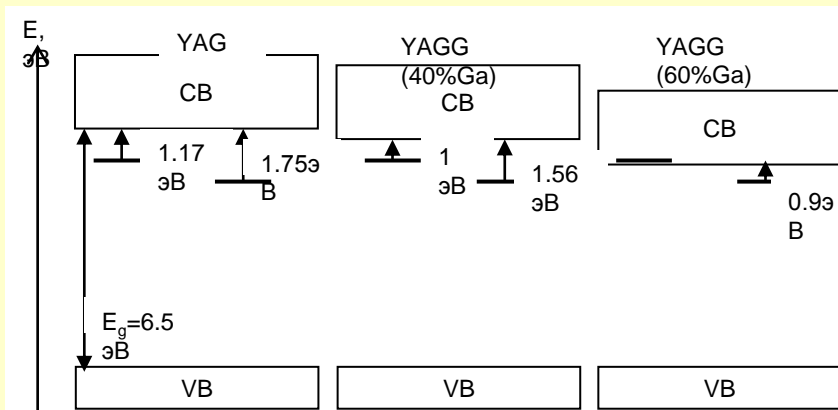
**Comparative samples.





e-h separation and/or conduction band modification?

Band structure change with Ga doping.



Ga doping (shift to mixed crystals)

- * Decrease the CB bottom level
- Decrease of shallow traps influence
-

M.Niki ...

* There are some alternative mechanisms that influence to light yield with similar or even higher rate

** Crystal performance, initial purity and activator concentration are crucial for the experimental study of phenomena

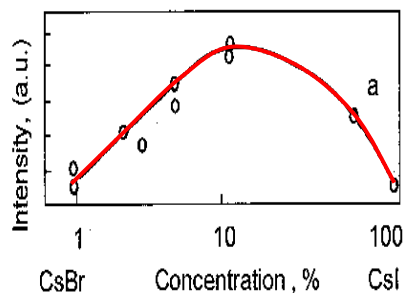
*** Decay time measurement could be more efficient for the model verification than yield test

**** We need in more detailed theoretical estimations for doped and mixed crystals

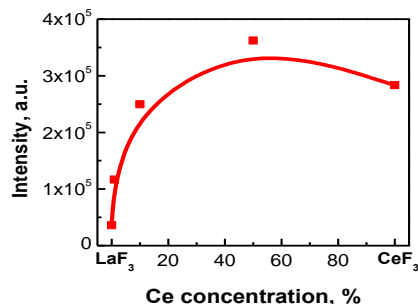


CONCLUSIONS

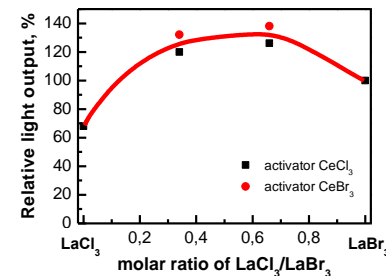
CsI-CsBr



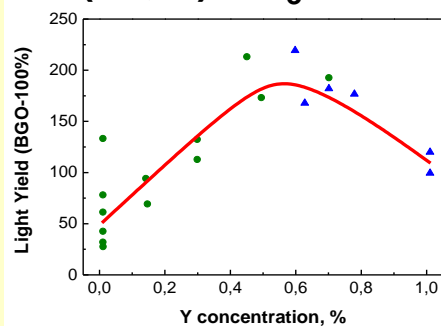
Ce_xLa_{1-x}F₃



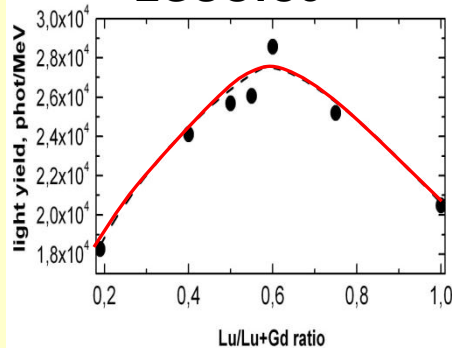
LaCl₃-LaBr₃



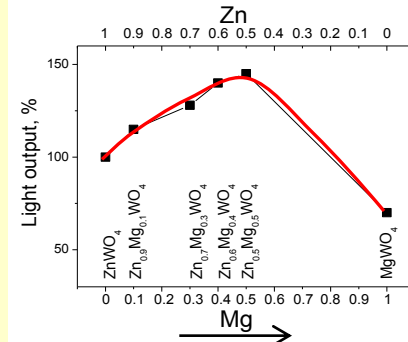
(Lu, Y)AlO₃: Ce



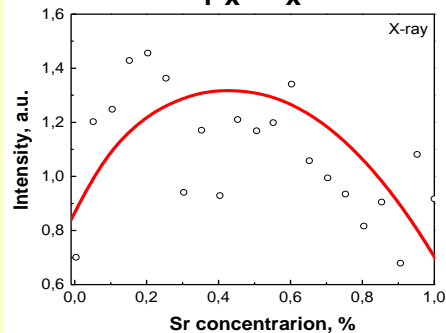
LGSO:Ce



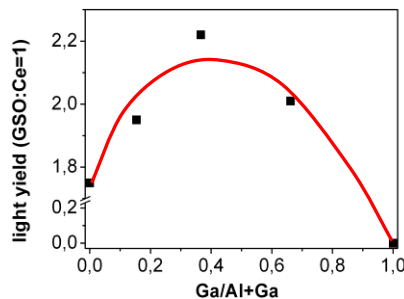
Zn_xMg_{1-x}WO₄



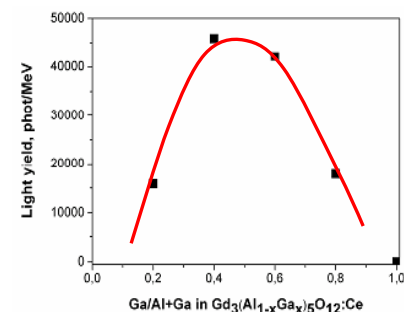
Ca_{1-x}Sr_xS



YAGG:Ce



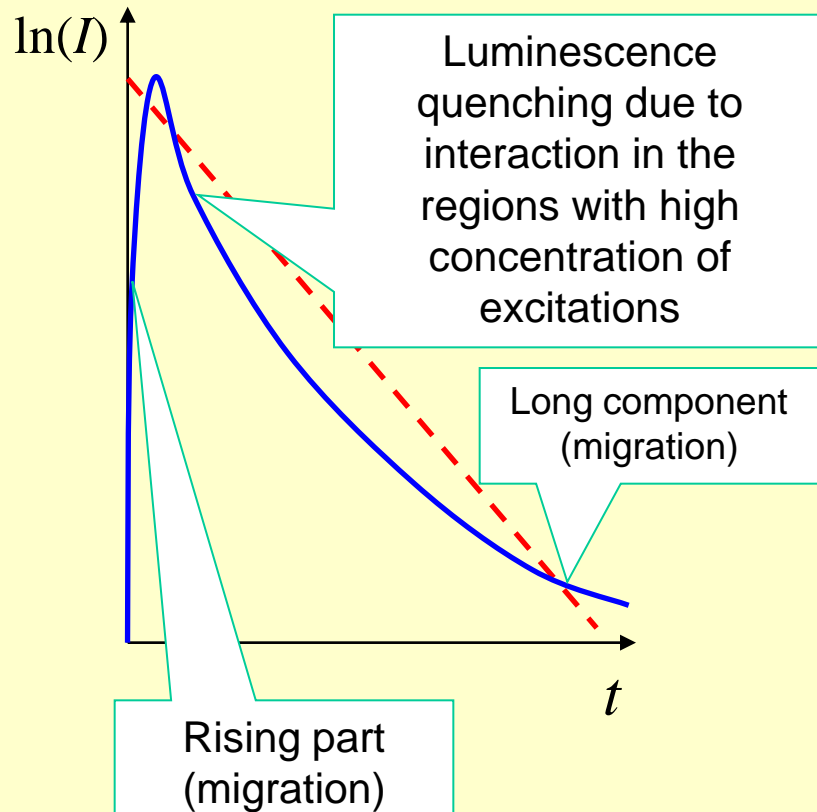
Gd₂(Al_xGa_{1-x})₅O₁₂:Ce



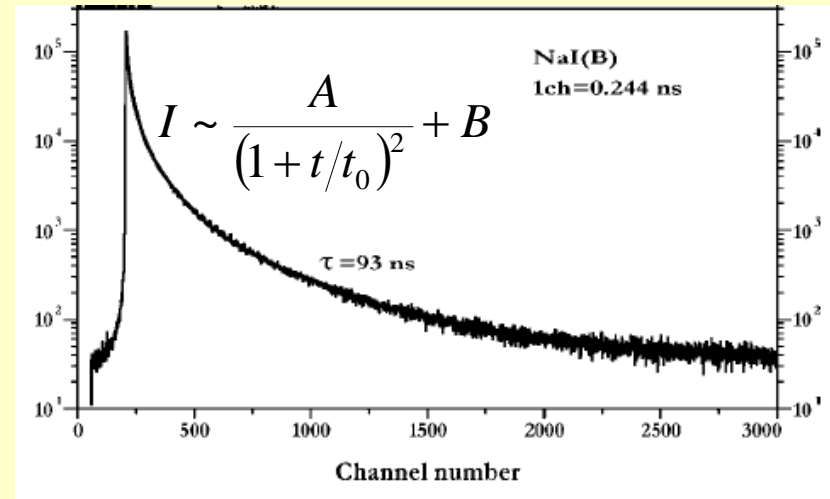


Modification of kinetics in scintillators

Scintillator kinetics



Essentially non-exponential decay kinetics for pure NaI



M. Moszyński, et al. Study of Pure NaI at RT and LNT, *IEEE TNS 2003*

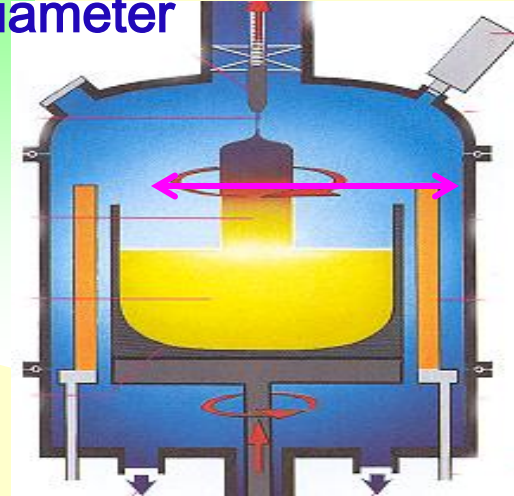
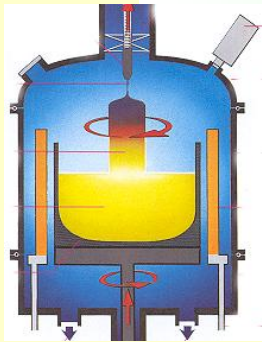


**CRYSTAL GROWTH ,
RAW MATERIAL,
TARGET PRICE
FOR ALKALI-EARTH HALIDES**



Two ways to increase industrial output

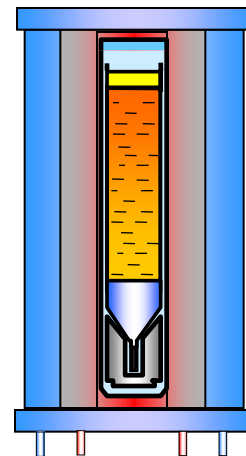
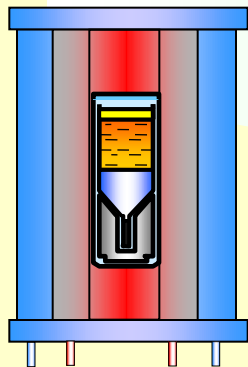
1. Increase of crystal / crucible diameter



Czochralski

- ✓ increased power input
- ✓ melt turbulences

2. Lengthening of crystal / melt height

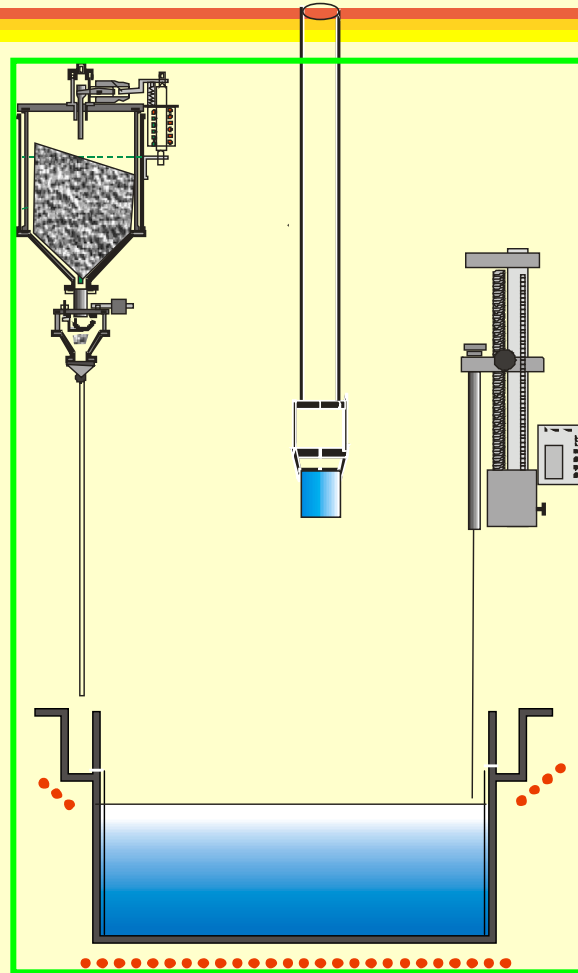


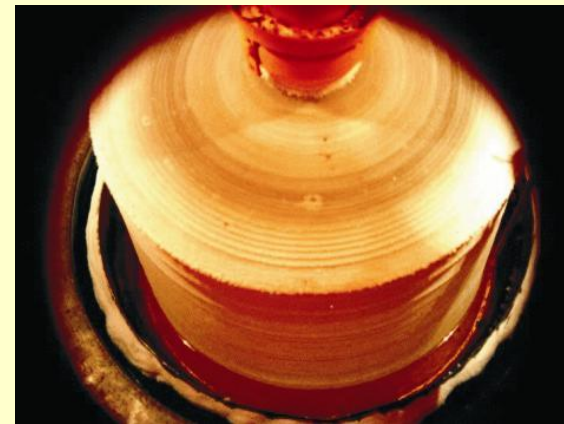
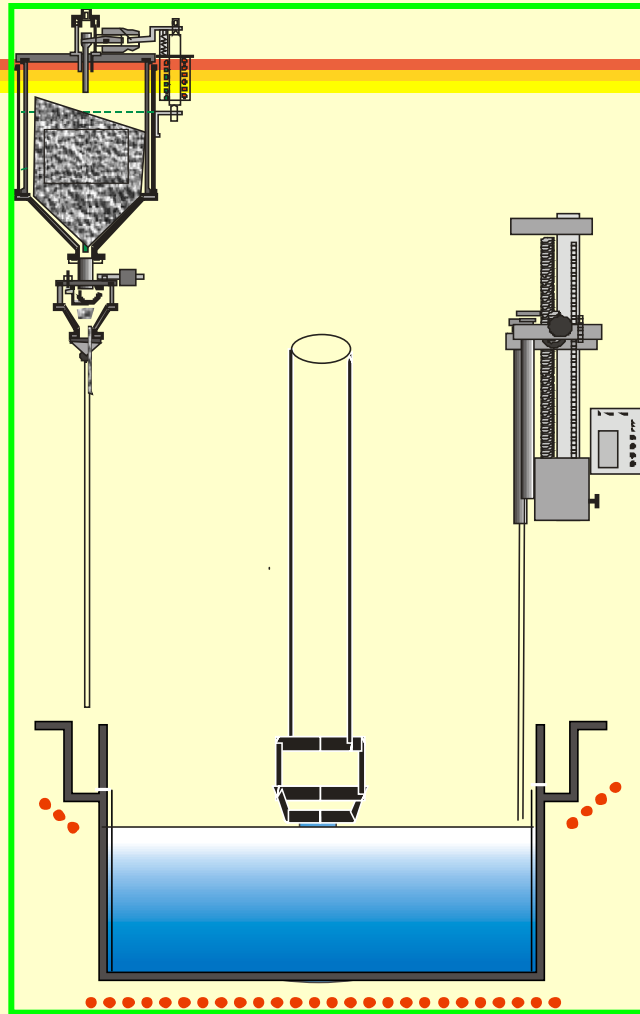
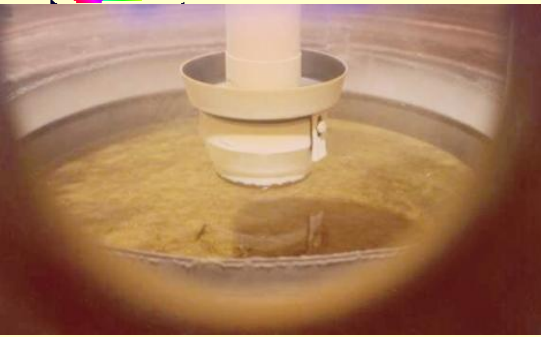
VGF

- ✓ increased interaction with ampoule
- ✓ increasing melt convection

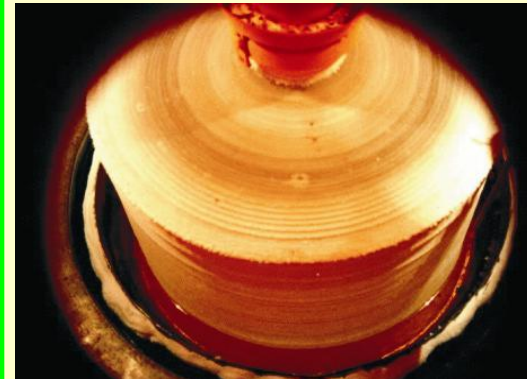
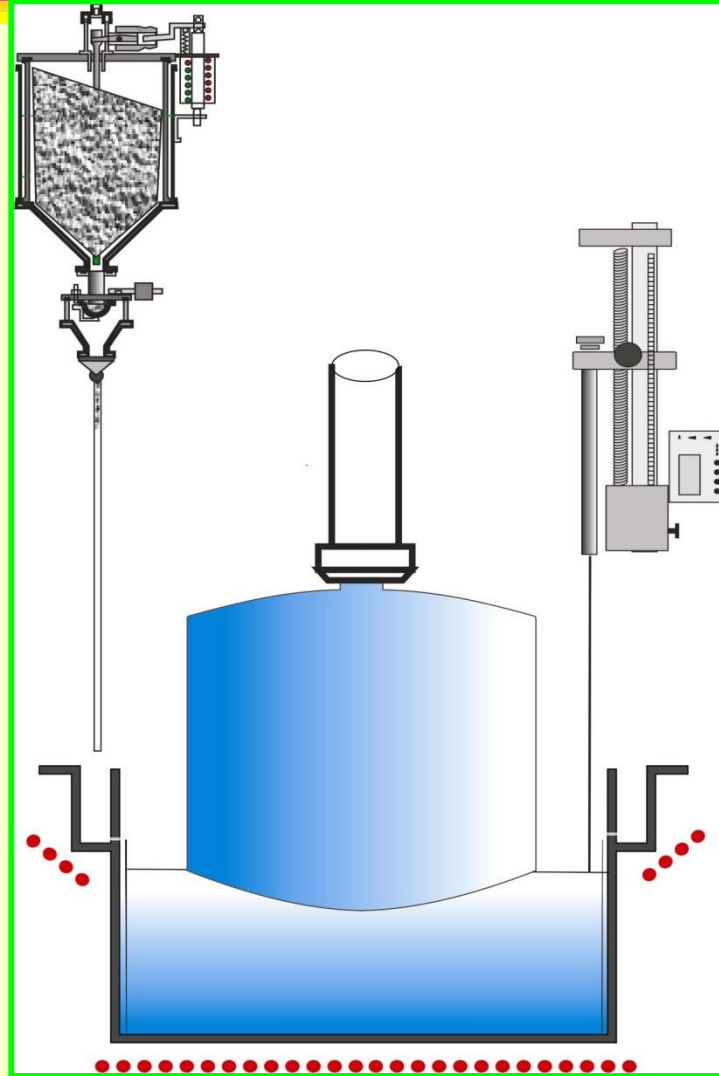
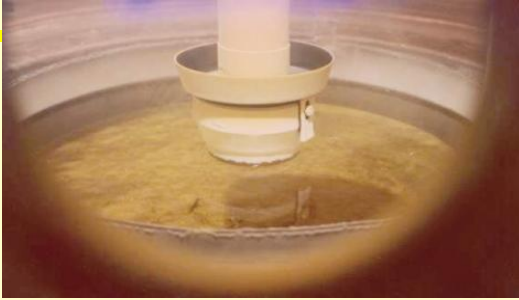


Continuous growth procedure



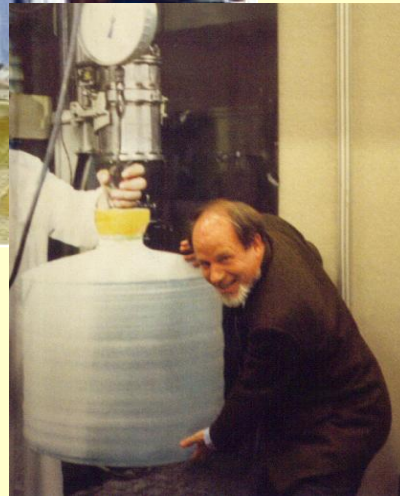
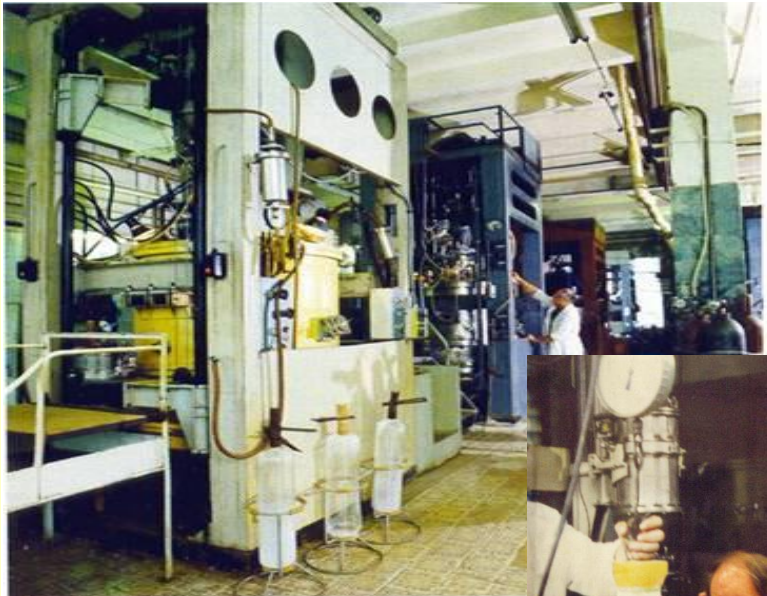


Nal(Tl) crystal continuous growth





From Principles to Practice



NaI(Tl)
Industrial growth

Hygroscopicity is not a problem!

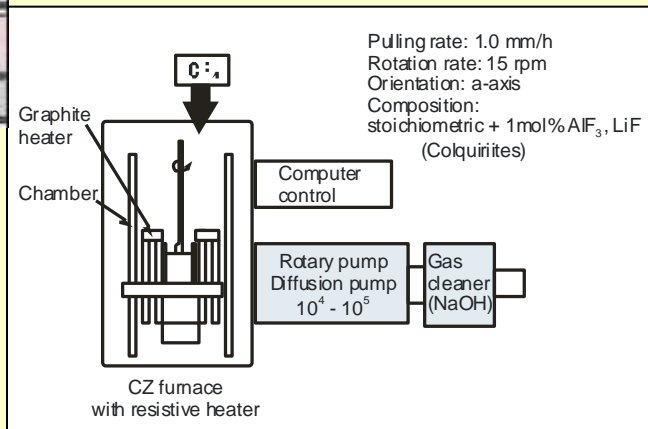
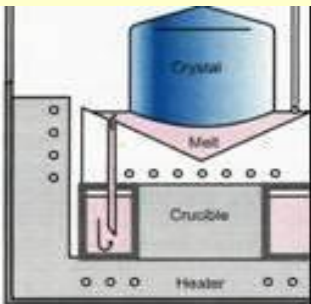


Si – large size crystal growth

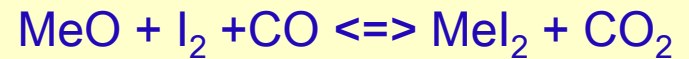
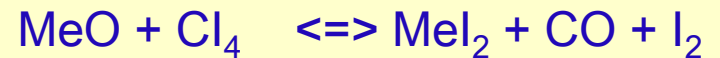
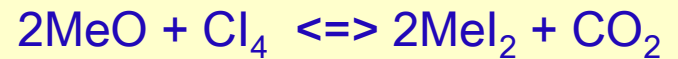
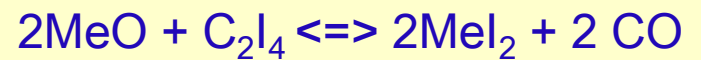
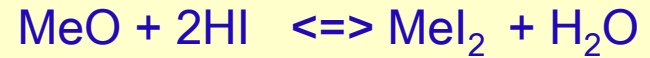
Si - industry is an example of efficient and cost reasonable crystals production



Crystal purity – crystal quality

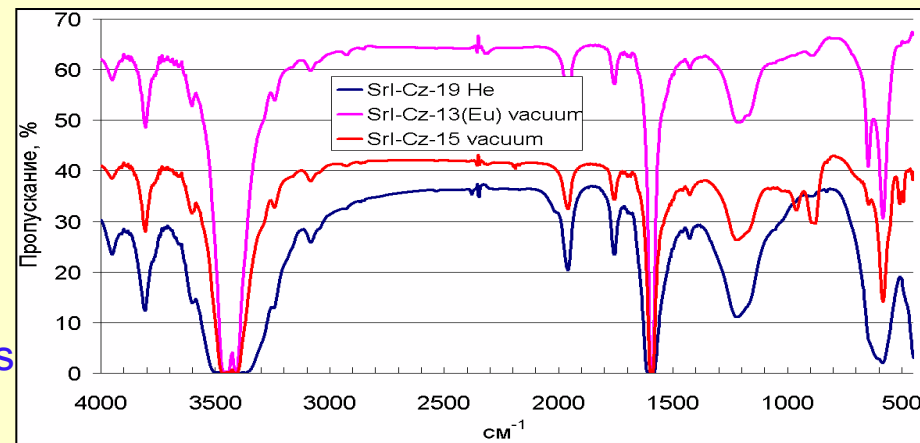


Growth "chemistry"



IR spectra of SrI₂:Eu

Peaks demonstrate oxygen impurities inside

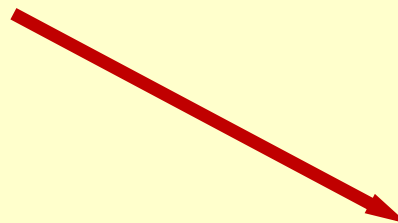




Current prices for halides powder

<i>Product</i>	<i>Producer</i>	<i>Price</i>
Srl ₂	Russia	4N \$1200/kg
	USA	4N \$3500/kg
	India	2N \$1300/kg
Eul ₂ *	Russia	4N ~ \$8000/kg
	USA	4N ~ \$20000/kg
BaBr ₂	Russia	4N ~\$1000/kg
Bal ₂	Russia	4N ~\$1000/kg
Csl	USA	5N \$150/kg
	Germany	5N \$160/kg
	Ukraine	5N \$150/kg

Current prices



Target price estimation
for hydrated AE halides

*Eu₂O₃ price \$3500-4500/kg (base for Eul₂)

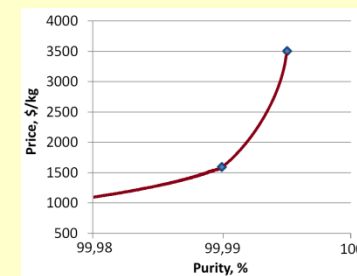
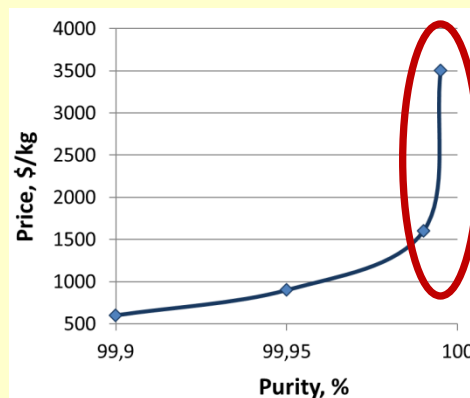
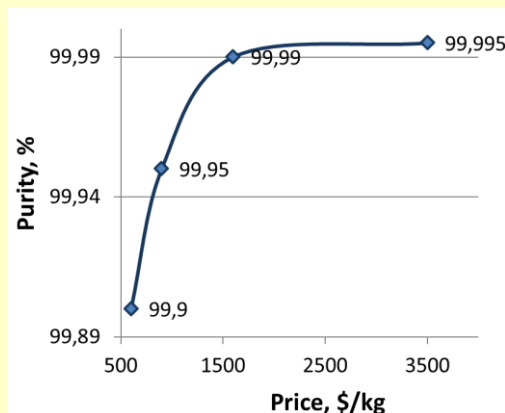
<i>Product name</i>	<i>Price per anhydrous</i>
Srl ₂ hydrate	\$220-270/kilo
BaBr ₂ hydrate	\$150-200/kilo
Bal ₂ hydrate	\$150-200/kilo



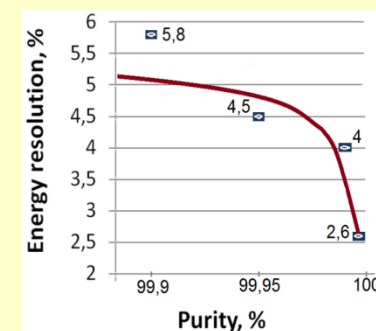
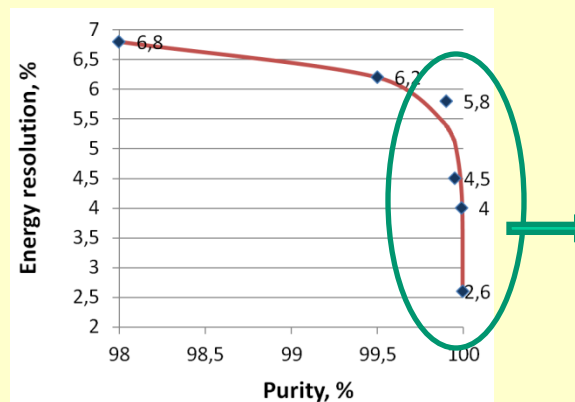
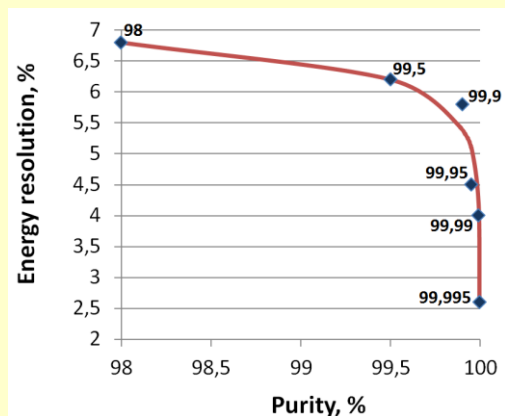
$\text{SrI}_2:\text{Eu}$ performance and cost vs. raw material purity

SrI_2 . Raw material cost depending on purity

(Lab level)

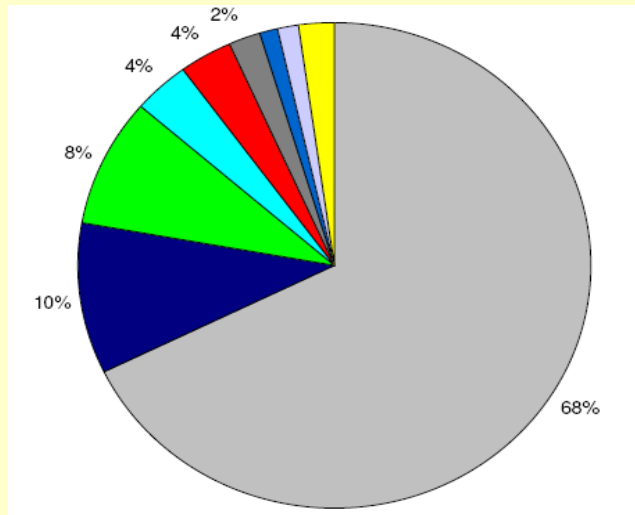


$\text{SrI}_2:\text{Eu}$. Energy resolution vs raw material purity





Cost structure for single crystal growth

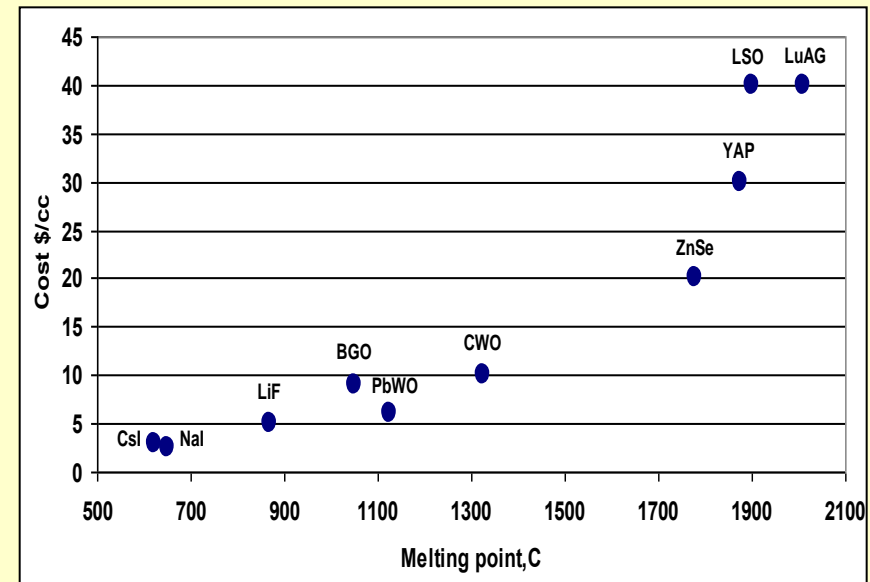


Crystal cost structure (Si)

- 68% - raw material
- 10% - crucible
- 8% - system cost
- 4% - labor cost
- 4% - power
- 6% - other

Oxides

- 20% - crucible
- 17% - power



2010 prices



- Natural “bottle neck” (self-absorption)
and *overpass ways*

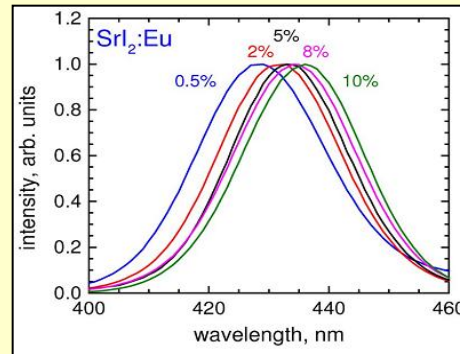


Self-absorption

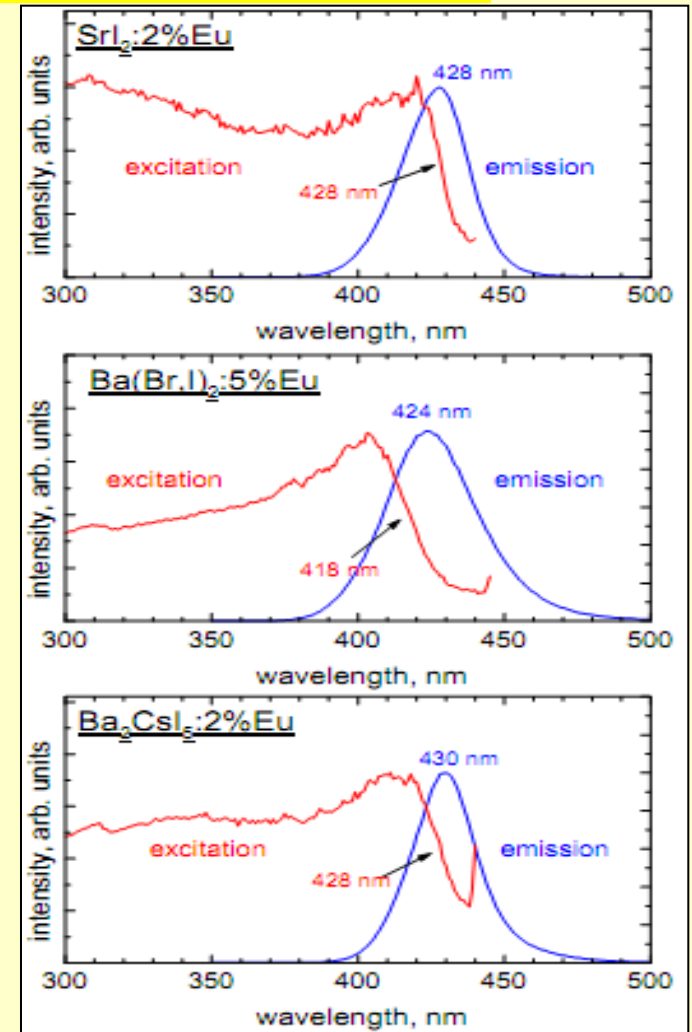
Self-absorption due to small Stokes shift is the key problem for large bulk Eu doped scintillator use

This is a typical for all Eu doped crystals !!!

Self-absorption lead to the low transparency and scintillator efficiency loss



J.Glodo et al. 2010



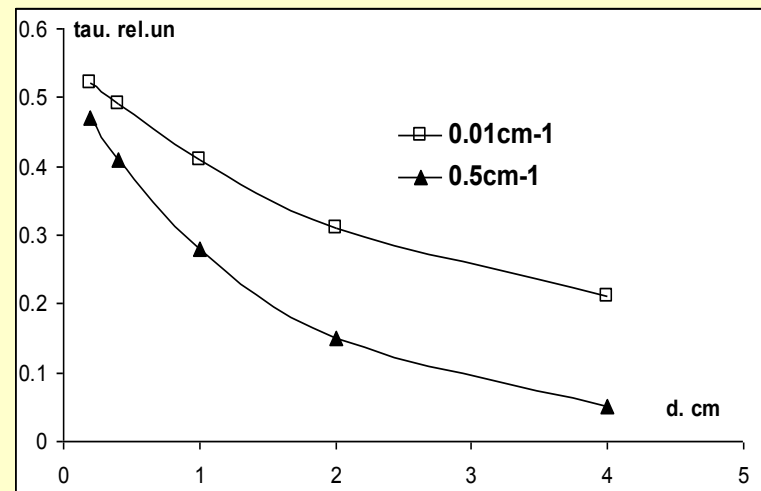
E. Bourret, S. Derenzo et al., 2009, 2010, 2011



Stokes shifts in Eu-doped scintillators

Crystal	Stokes shift, eV
<i>Alkali halides</i>	
NaI (Tl)	1.35
NaI (Eu)	0.8
CsI (Tl)	1.93
CsI (Na)	2.07
CsI (In)	1.83
CsI (Eu)	0.8
<i>Alkali-earth halides</i>	
SrI ₂ (Eu)	0.15
CaI ₂ (Eu)	0.30
Ba ₂ CsI ₅ (Eu)	0.15
CsSrI ₃ (Eu)	0.55
BaBrI (Eu)	0.40

AE scintillator have a small Stokes shift and low transparency



Light collection for high (0.01 cm⁻¹) and low (0.5 cm⁻¹) transparent crystals

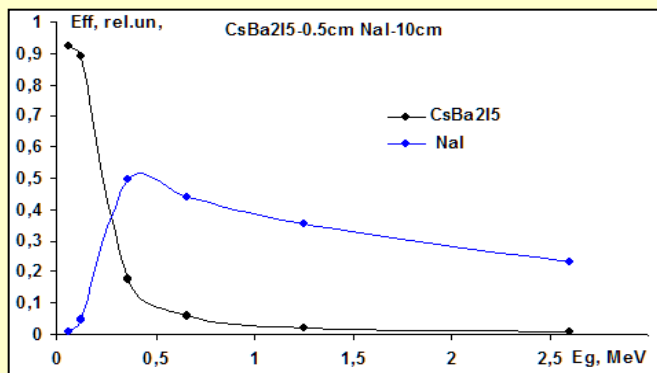
V.A. Tarasov, ISMA, 2011

Resume:

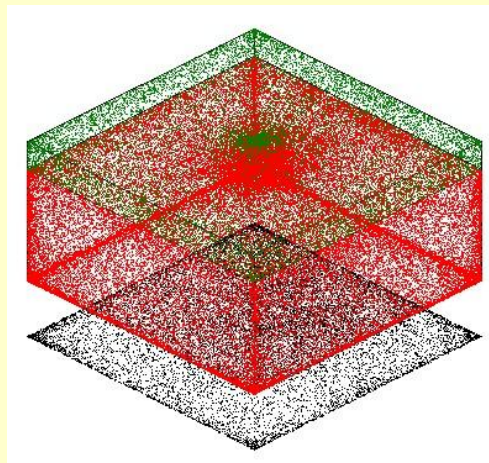
Bulk Eu-doped scintillator could not be efficient scintillators



Combined (phoswich) detectors as a way to increase scintillator efficiency

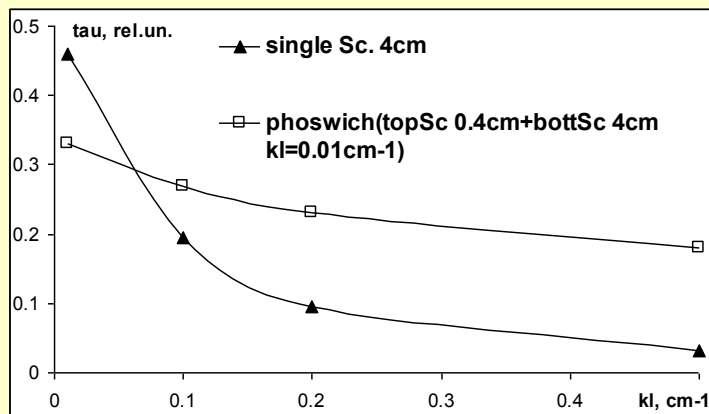


Phoswich detector scheme



Gamma registration efficiency at 0.06 – 2.65 MeV for 0.5 cm thick $\text{SrI}_2(\text{Eu})$ combined with 10 cm $\text{NaI}(\text{TI})$

Signal separation due to decay time difference



$\text{SrI}_2 : \text{Eu}$	1200ns
$\text{NaI} : \text{TI}$	230ns
$\text{BaBrI} : \text{Eu}$	500ns
$\text{CsBa}_2\text{I}_5 : \text{Eu}$	383; 1200ns
$\text{CsI} : \text{TI}$	980ns

Light collection coefficient for $\text{SrI}_2 : \text{Eu}$ and $\text{SrI}_2(\text{Eu}) + \text{NaI}(\text{TI})$ phoswich detectors 4 cm thickness



Conclusions

- ✓ The last years scintillators developments significantly upscale our vision of the perfect scintillator.
- ✓ Practical needs rise up theoretical studies and view to the fundamental process in radiation absorptions and primary stages of excited states relaxation (evolution).
- ✓ Theoretical simulations allow to select the process “skeleton” to make new material search more predictable.
- ✓ Next years forecast –
 - community has to select two-three best candidates for the technology development
 - we need in criteria for the best scintillators selection



THANK YOU FOR ATTENTION!