

IDENTIFICATION OF GAS PHASE DNA BASES SPECIFIC ISOMERS VIA HIGH ACCURACY SINGLE PHOTON IONIZATION AND *AB INITIO* COMPUTATIONS

Majdi HOCHLAF

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OBJECTIVES

- **Accurate determination of ionization energies** of isolated gas phase medium-sized molecules and investigation of **the vibrational structure of involved ionic states**, including their mutual vibronic couplings.
- **Treatment of tautomers, rotamers, conformers** in the ground state, after photoionization, and in competition with other intramolecular processes.
- **Study of fragmentation of the molecular cations** at low and high internal energies.
- **Study of the evolution of the covalent character of hydrogen bonding** upon substitution, i.e., examination of electronic effects (acceptor, donor, etc.)

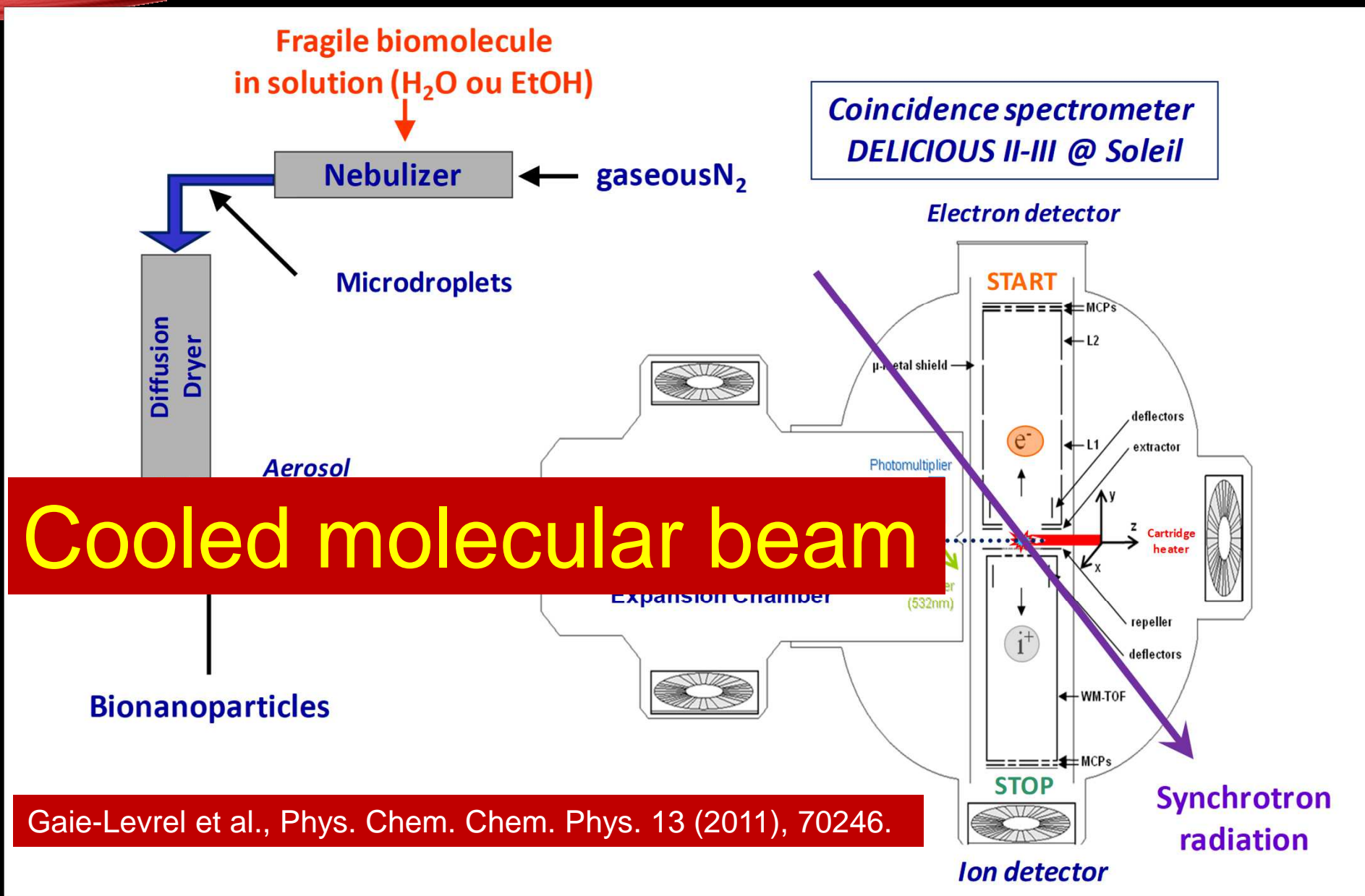
Major questions:

- How to get medium-sized molecules in gas phase?
- What to measure?
- Pb of autoionization vs. direct photoionization
- What to compute?
- How to compute accurate properties for these relatively large species



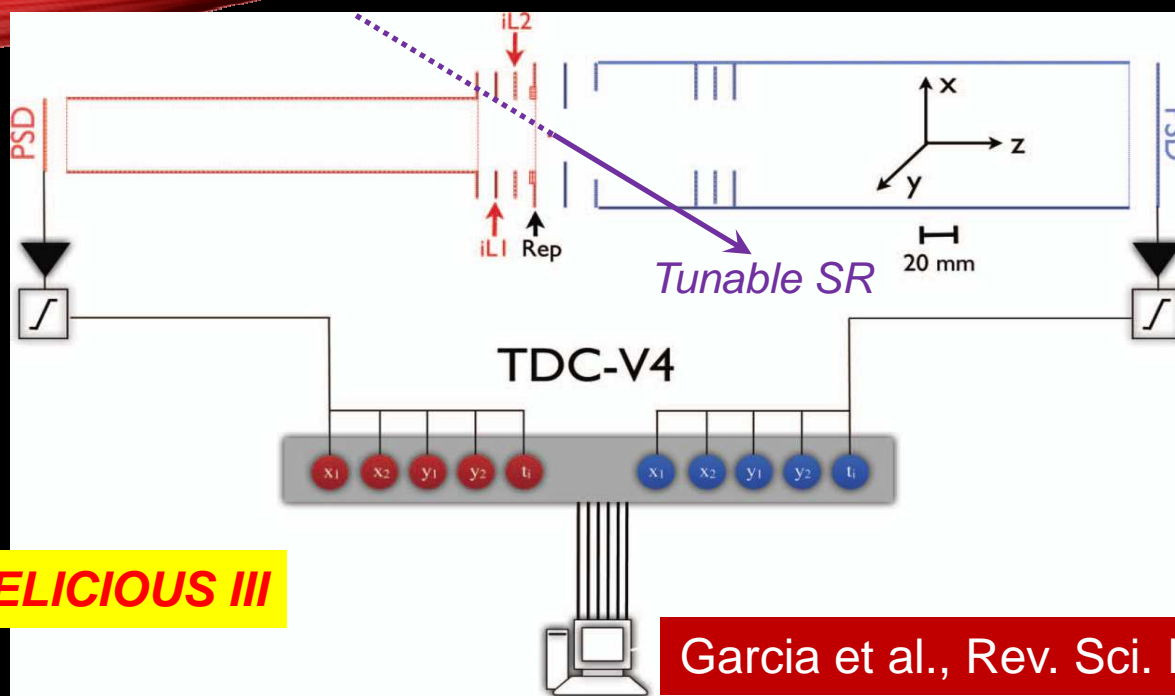
VUV SINGLE
PHOTOIONIZATION OF
NABS @
DESIRS BEAMLIN
(SYNCHROTRON
SOLEIL)

Desorption of "bionanoparticles" on a hot tip



Gaie-Levrel et al., Phys. Chem. Chem. Phys. 13 (2011), 70246.

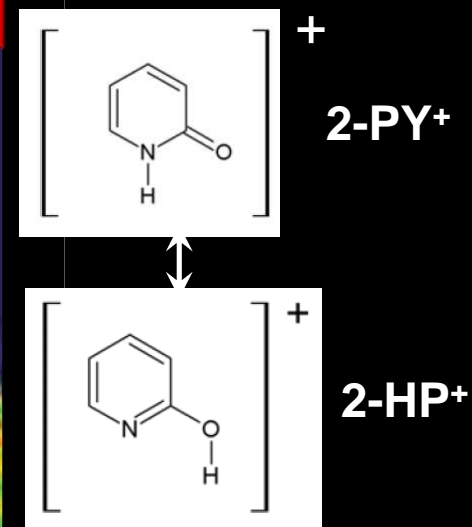
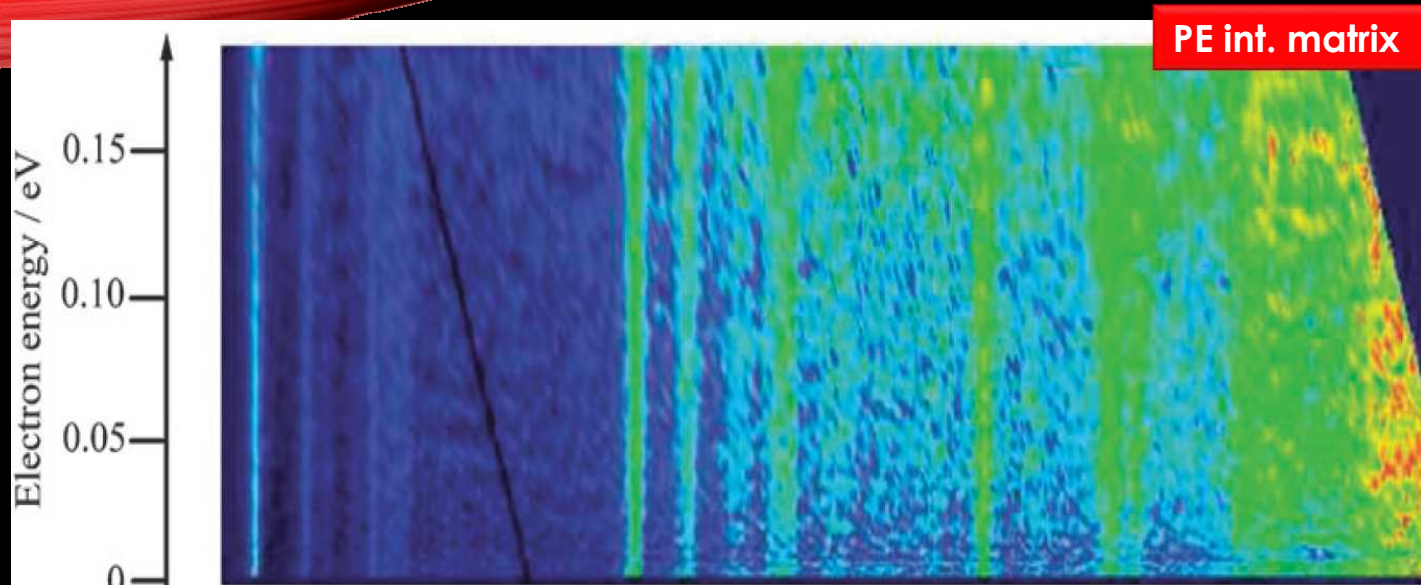
Methods used: DELICIOUS III, coincidence spectroscopy, imaging



DELICIOUS III: multipurpose instrument available @ DESIRS beamline (permanent end station)

- Jet-cooled molecules
- Detection of electrons and ions in coincidence , as a function of photon energy and kinetic energy released (using imaging techniques)
- PI mass spectrometry, PEPICO, TPEPICO
- High resolution (threshold) photoelectron spectroscopy (about 1 meV), TPES, Slow-PES (SPES)

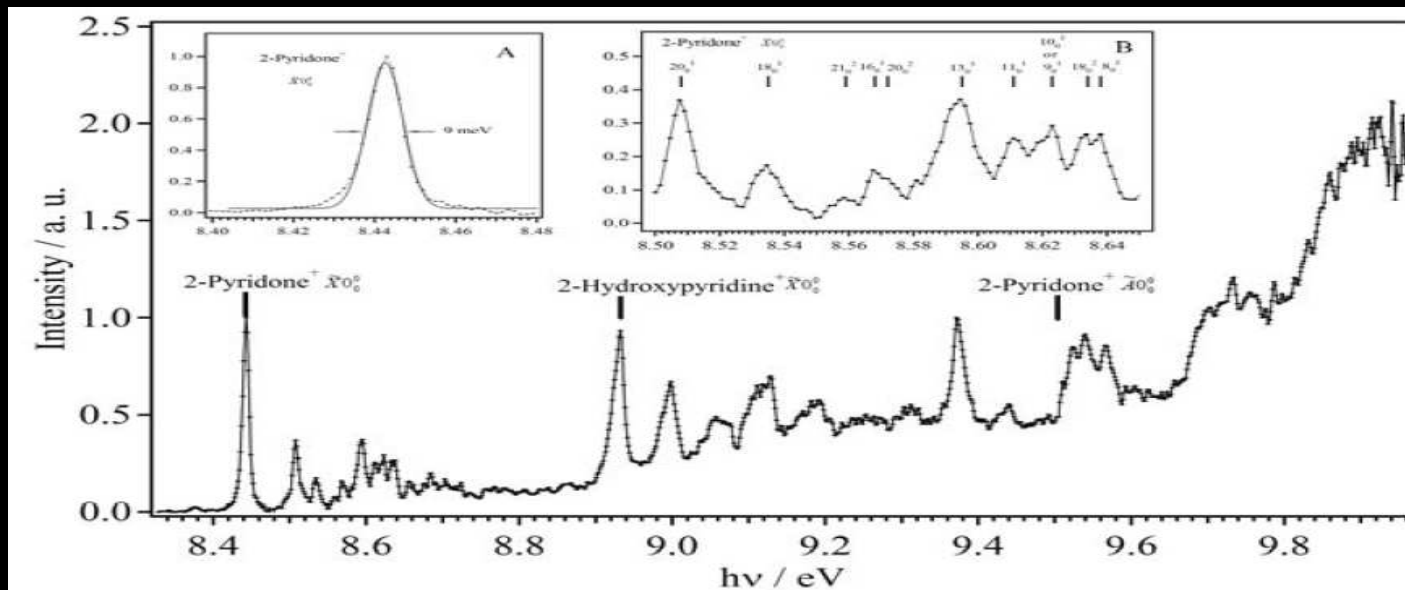
Slow PhotoElectron Spectroscopy: 2-Pyridone₇



2-Pyridone and
2-Hydroxypyridine

SPES

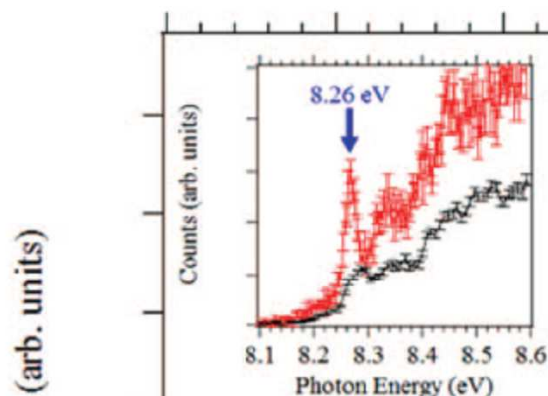
KE_{max} = 75 meV



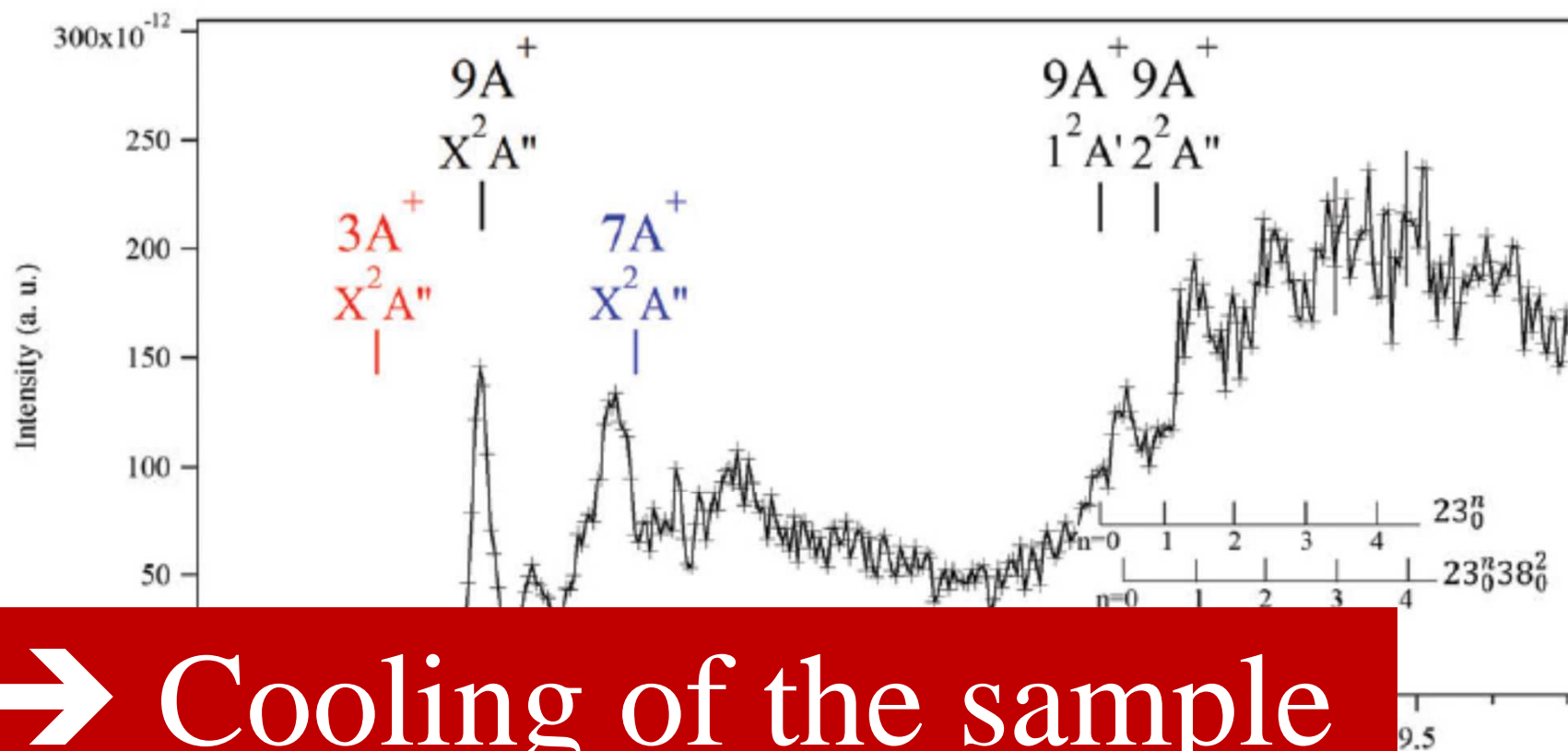
Briant et al., PRL 109, 193401 (2012).

Pouilly et al., PCCP 12, 3566 (2010).

Adenine



Cite this: *Phys. Chem. Chem. Phys.*,
2018, 20, 20756



→ Cooling of the sample



THEORETICAL COMPUTATIONS

Characterization of medium sized molecular systems

- Ground state computations: *Ab initio* (MP2, CCSD(T), **CCSD(T)-F12**) and DFT computations using the recently developed/implemented functionals and methods: Prediction of the equilibrium structures and their energetics and vibrational spectra

Benchmarking DFT methods vs. newly implemented explicitly correlated methods

- **F12 methods: Consideration of large amount of electron correlation**
- **F12 → DFT: Extension to large molecular systems**
- MOLPRO: <http://www.molpro.net>
- GAUSSIAN: www.gaussian.com

→ Consideration of Core-Valence (CV), Relativistic effects (RS), ...

- Electronic excited states: Multi configurational approaches: CASSCF, MRCI, MRCI+Q (**But size inconsistency ...**) or EOM-CCSD for monoconfigurational electronic states

Basis sets:

→ Standard methods: aug cc-pVXZ ($X > 4$ or $X = \text{CBS}$), Dunning and co-workers) + (R)CCSD(T) or CASSCF/MRCI

→ Explicitly correlated methods: aug-cc-pVXZ ($X=3$), cc-pVTZ-F12 (To be used with care, Peterson et al.) + (R)CCSD(T)-F12

→ **Efficient composite scheme for computation of IEs of medium sized molecular systems**

THYMINE

THE JOURNAL OF
PHYSICAL CHEMISTRY A

Article
pubs.acs.org/JPCA

Theoretical and Experimental Photoelectron Spectroscopy Characterization of the Ground State of Thymine Cation

Youssef Majdi

Laboratoire de Spectroscopie Atomique, Moléculaire et Applications (LSAMA), Université de Tunis El Manar, Tunis, Tunisia

Majdi Hochlaf*

THE JOURNAL OF
PHYSICAL CHEMISTRY A

Article
pubs.acs.org/JPCA

Vibrationally Resolved Photoelectron Spectroscopy of Electronic Excited States of DNA Bases: Application to the \tilde{A} State of Thymine Cation

Majdi Hochlaf*

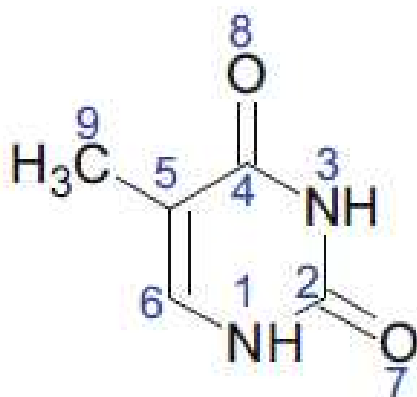
Laboratoire Modélisation et Simulation Multi Echelle, MSME UMR 8208 CNRS, Université Paris-Est, 5 bd Descartes,
77454 Marne-la-Vallée, France

Yi Pan and Kai-Chung Lau*



frontiers
in Astronomy and Space Sciences

ORIGINAL RESEARCH
published: 14 October 2021
doi: 10.3389/fspas.2021.757007



Thymine (T)

Identification of DNA Bases and Their Cations in Astrochemical Environments: Computational Spectroscopy of Thymine as a Test Case

Yage Zhao¹, Majdi Hochlaf² and Malgorzata Biczysko^{1*}

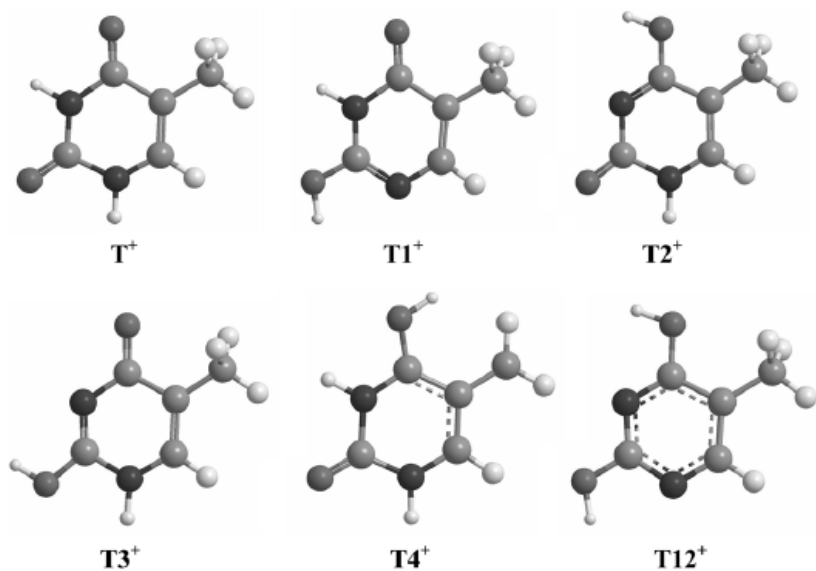
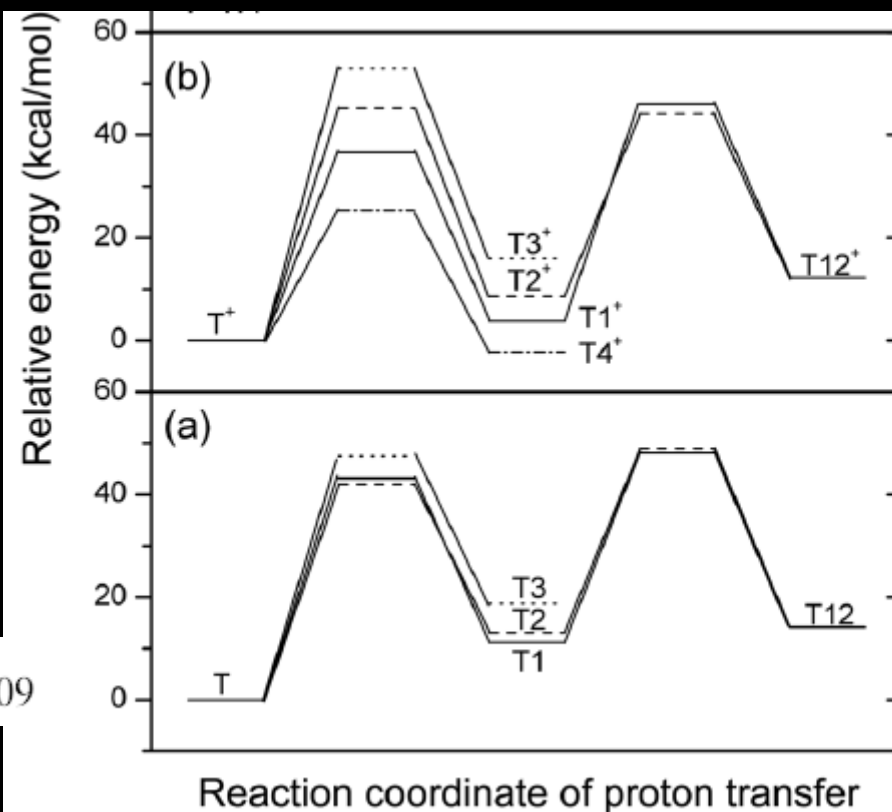


Figure 2. The structures of the tautomers of thymine radical cation optimized at the B3LYP/6-311++G(d,p) level.

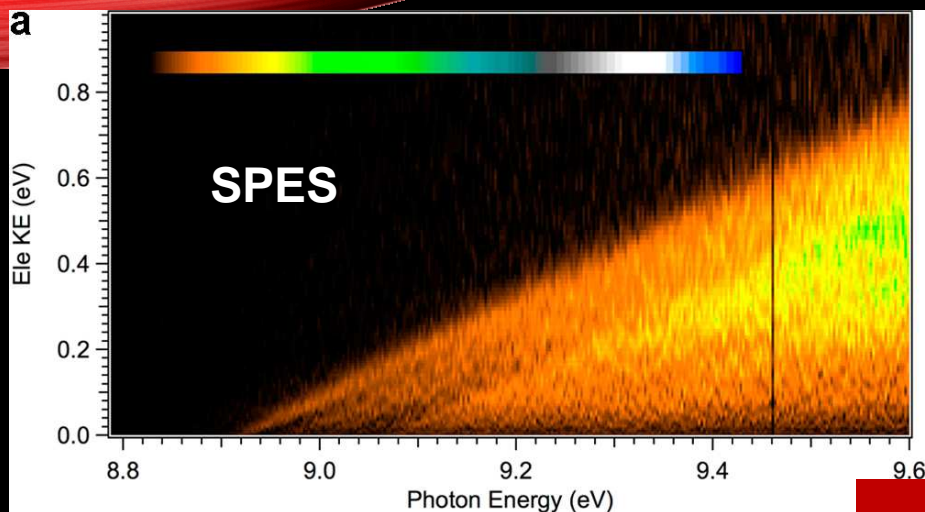
Nam Joon Kim

Bull. Korean Chem. Soc. **2006**, Vol. 27, No. 7 1009

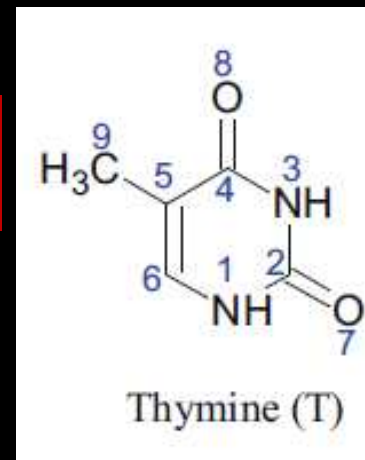
Neutral	ΔE_0^b	kc
T	0.00	
T1	11.19	
T2	13.08	
T3	18.76	
T12	14.17	



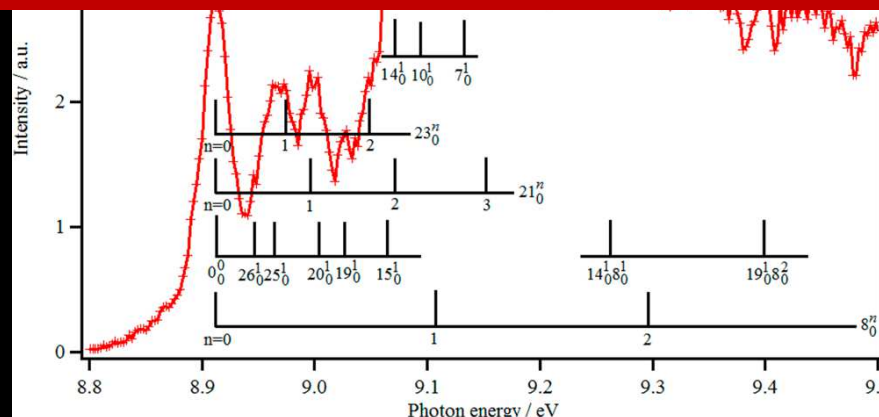
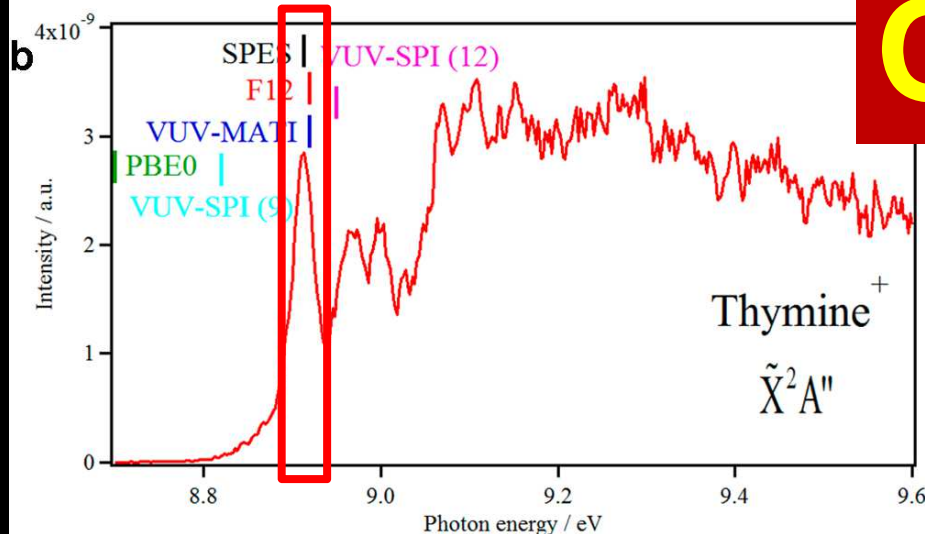
Thymine⁺ - cation ground state



SOLEIL



One tautomer



Theo.: **F12** = (R)CCSD(T)-F12/cc-pVTZ-F12+ZPVE
PBE0 (DFT) = PBE0/aug-cc-pVDZ+ZPVE
 Expt.: **VUV-SPI**, **VUV-SPI**, **VUV MATI**, **SPES**

Majdi et al., JPC. A (2015)

ADENINE

PCCP

PAPER



Volume 20 | Number 32 | 28 August 2018 | Pages 20737–21206

PCCP

Physical Chemistry Chemical Physics
rsc.li/pccp

Unveiling the complex vibronic structure of the canonical adenine cation†

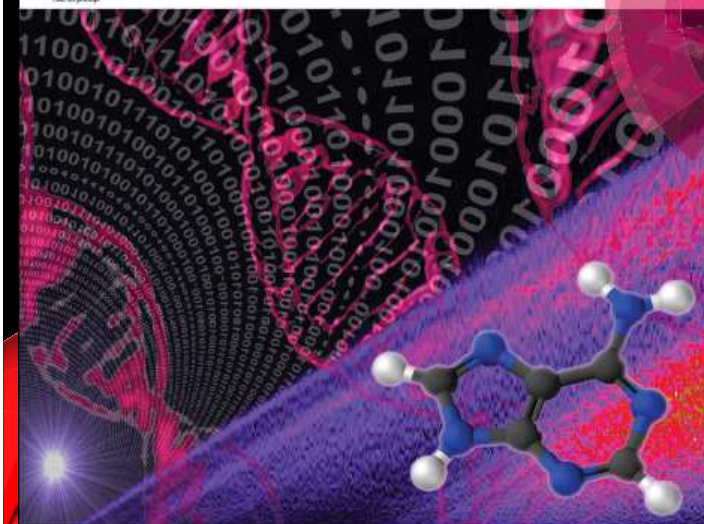
Hong Yan Zhao,^a Kai-Chung Lau,^{id}^a Gustavo A. Garcia,^{id}^b Laurent Nahon,^{id}^b Stéphane Carniato,^c Lionel Poisson,^{id}^d Martin Schwell,^e Muneerah Mogren Al-Mogren^f and Majdi Hochlaf^{id}^{*g}

THE JOURNAL OF CHEMICAL PHYSICS 138, 094203 (2013)



VUV photoionization of gas phase adenine and cytosine: A comparison between oven and aerosol vaporization

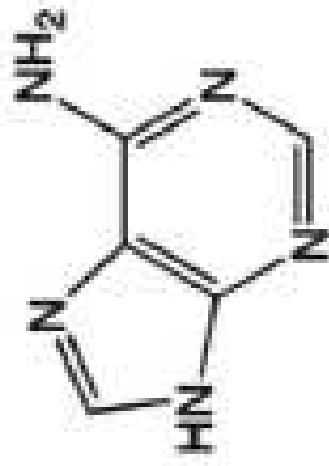
D. Touboul,¹ F. Gaie-Levrel,^{2,3} G. A. Garcia,² L. Nahon,² L. Poisson,⁴ M. Schwell,⁵ and M. Hochlaf^{6, a)}



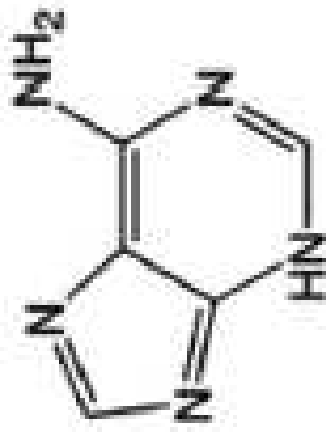
ISSN 1463-9076



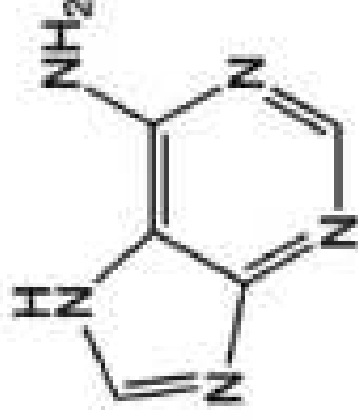
PAPER
Majdi Hochlaf et al.
Unveiling the complex vibronic structure of the canonical adenine cation



9H-adenine (9A)



3H-adenine (3A)



7H-adenine (7A)

Scheme 1 Structures of 9H-adenine, 3H-adenine and 7H-adenine. Between parentheses, we give the symbols used for their designation in the present study. For common atom numbering, please refer to the ESI.†

Table: Total energies (in Hartrees) and ionization energies (IE in eV) at the (R)CCSD(T)-F12(b)/cc-pVTZ-F12 (+CV+SR+ZPVE) level. ZPVE corresponds to the zero point vibrational energy (in cm^{-1}).

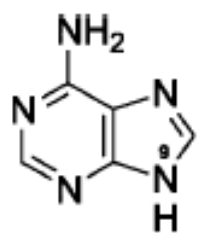
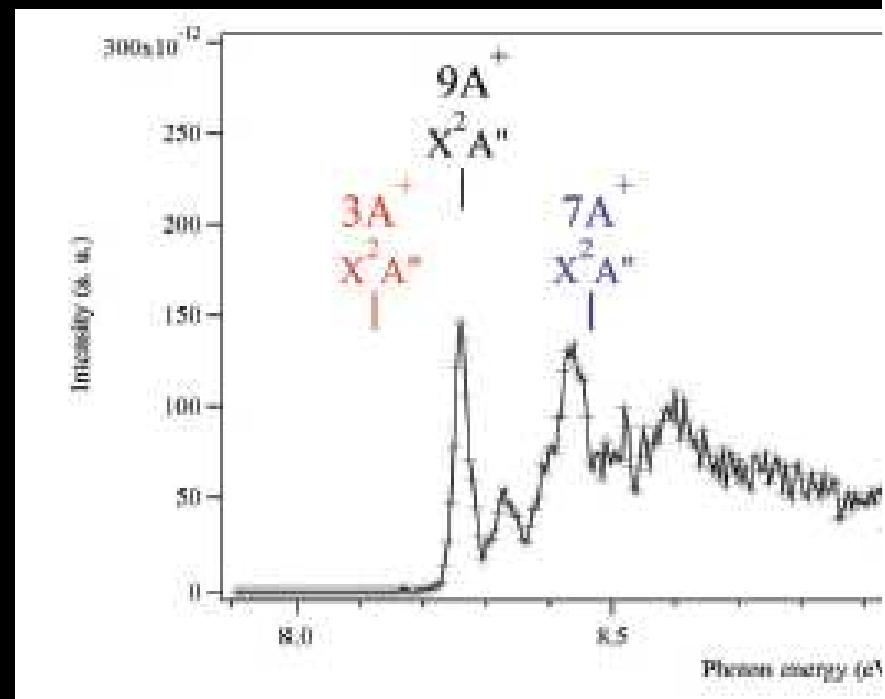
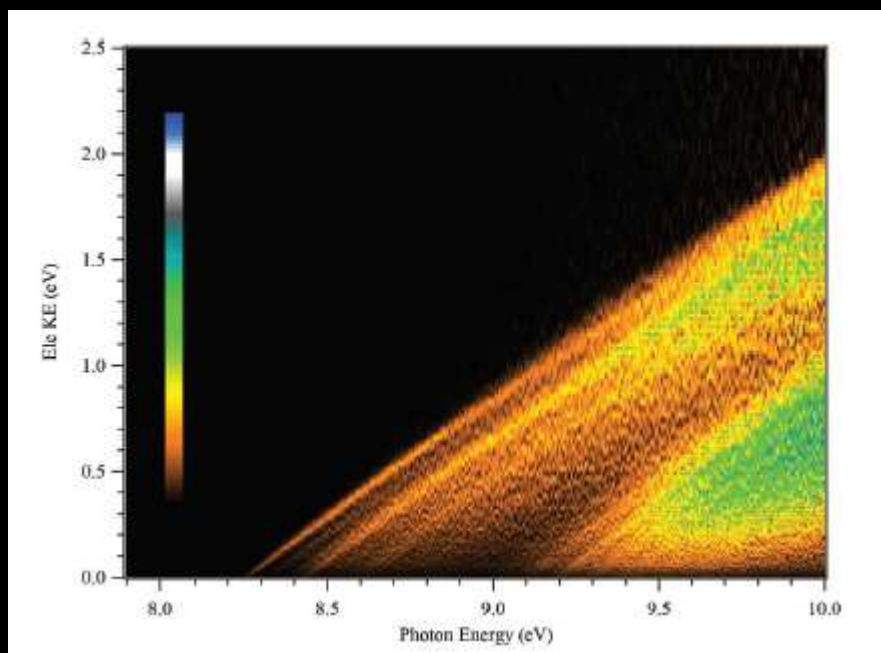
Species	Total energies						
	9H-adenine	9H-adenine ⁺	3H-adenine	3H-adenine ⁺	7H-adenine	7H-adenine ⁺	
PBE0/aug-cc-pVDZ	-466.87444	-466.57908	-466.86206	-466.56945	-466.86187	-466.55799	
ZPVE ^a	24738	24828	24865	24768	24771	24650	
CCSD(T)-F12b/cc-pVTZ-F12	-466.73389	-466.43108	-466.72056	-466.42223	-466.72200	-466.41086	
CCSD(T,fc)/cc-pwCVTZ	-466.57363	-466.27777	-466.55994	-466.26894	-466.56182	-466.25734	
CCSD(T,full)/cc-pwCVTZ	-467.07047	-466.77404	-467.05671	-466.76506	-467.05849	-466.75351	
CCSD(T)/cc-pVTZ	-466.54498	-466.24939	-466.53130	-466.24060	-466.53316	-466.22896	
CCSD(T)/cc-pVTZ-DK	-466.76288	-466.46745	-466.74922	-466.45867	-466.75111	-466.44704	
Ionization energies(in eV)							
CCSD(T)-F12b	8.240		8.118		8.467		
CV	0.016		0.018		0.014		
SR	-0.004		-0.004		-0.004		
ZPVE	0.011		-0.012		-0.015		
IE(F12b+CV+SR+ZPVE)	8.262		8.120		8.462		

^a Based on harmonic vibrational frequencies at PBE/aug-cc-pVDZ level.

Single Point computations

OPTG

Spectroscopy of a unique tautomer of Adenine⁺



9H - amino

Assignment of the ground state spectrum of 9H-Adenine⁺



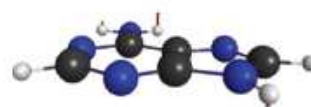
$\nu_9^+(a')$



$\nu_{23}^+(a')$



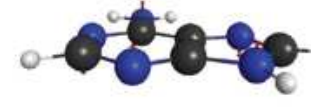
$\nu_{24}^+(a')$



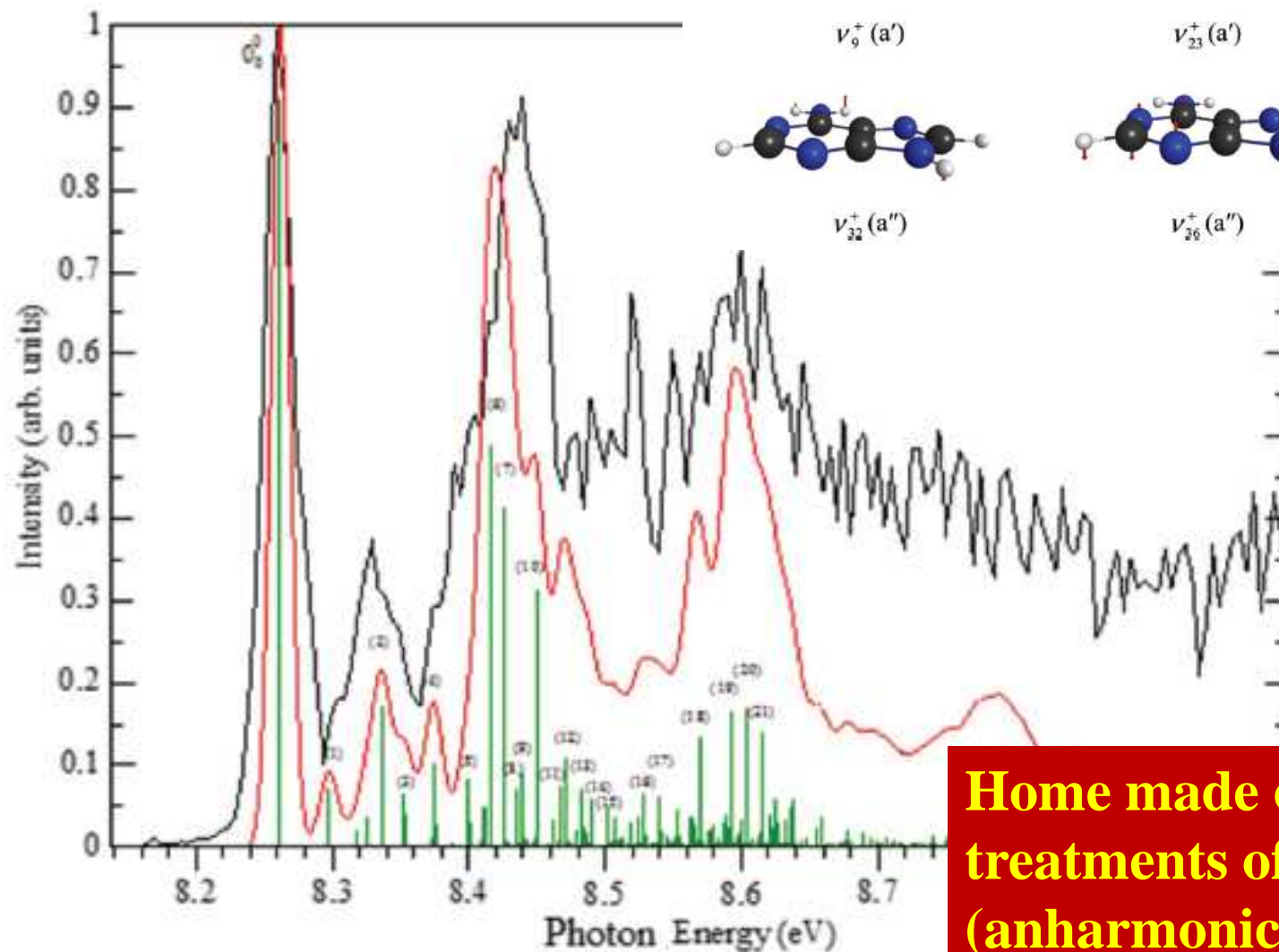
$\nu_{32}^+(a'')$



$\nu_{26}^+(a'')$

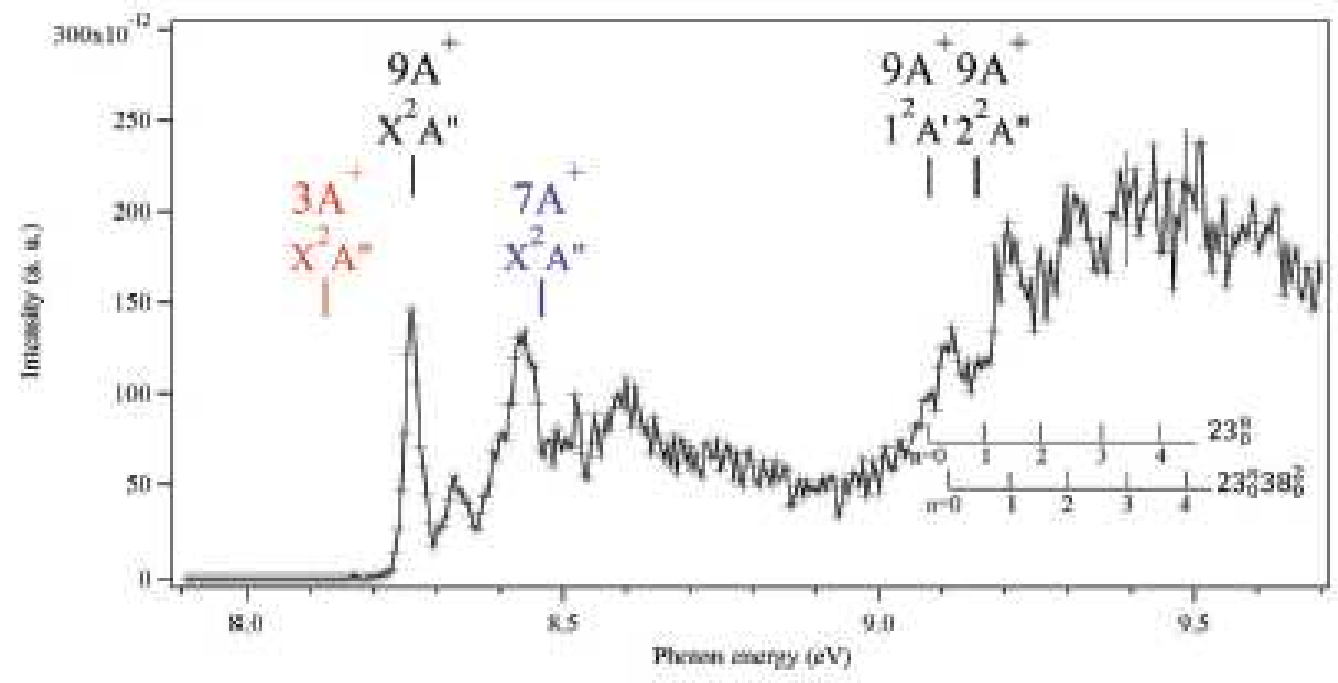
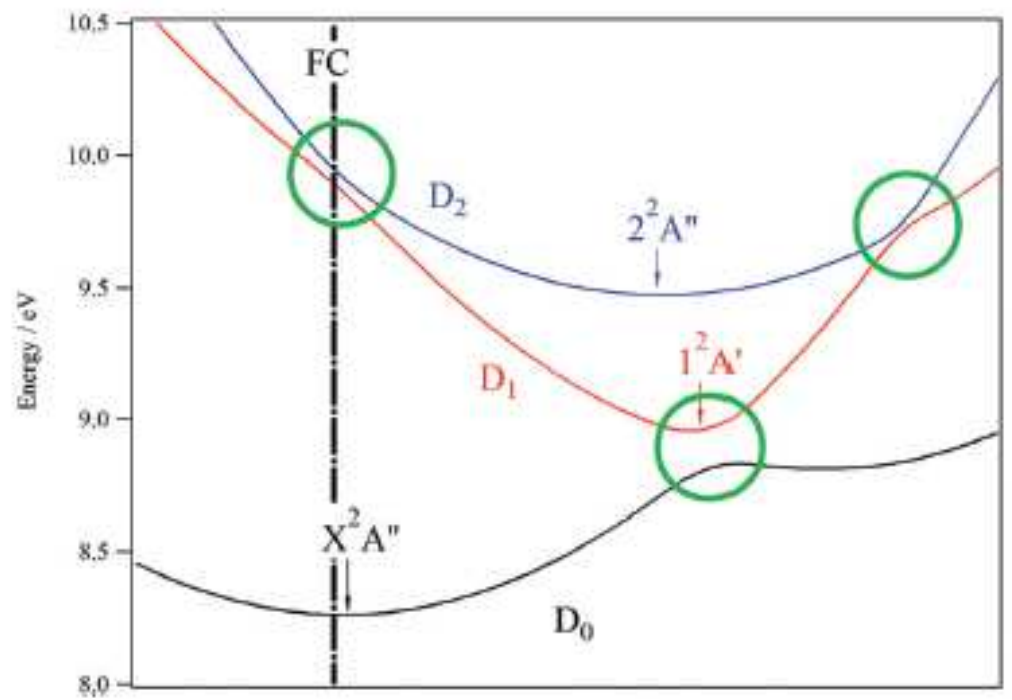


$\nu_{29}^+(a'')$



Home made codes for the treatments of vibrations (anharmonic resonances, ...)

Electronic excited states of 9H-Adenine⁺



ADENINE - THYMINE

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LETTERS

A JOURNAL OF THE AMERICAN CHEMICAL SOCIETY

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Letter

Photoionization Dynamics and Proton Transfer within the Adenine-Thymine Nucleobase Pair

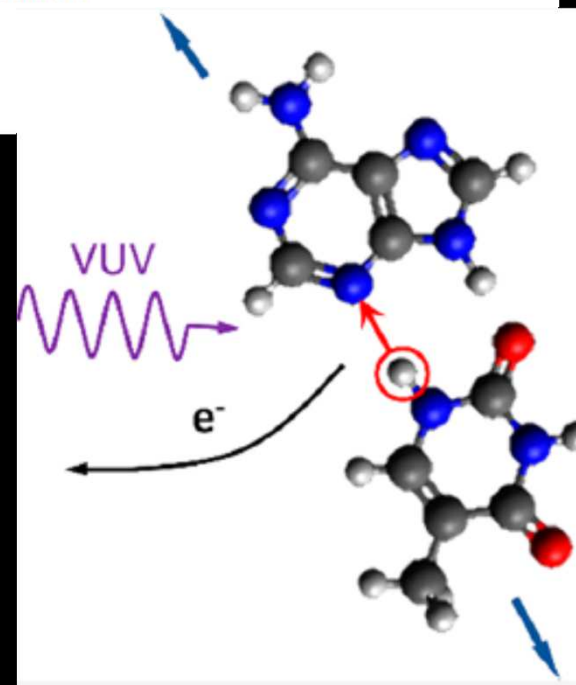
Sebastian Hartweg,* Majdi Hochlaf,* Gustavo A. Garcia, and Laurent Nahon

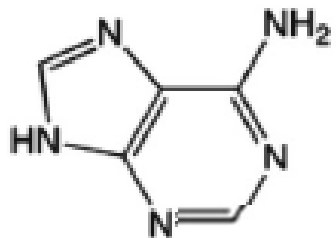


Cite This: *J. Phys. Chem. Lett.* 2023, 14, 3698–3705



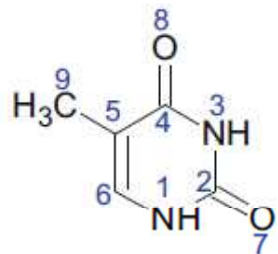
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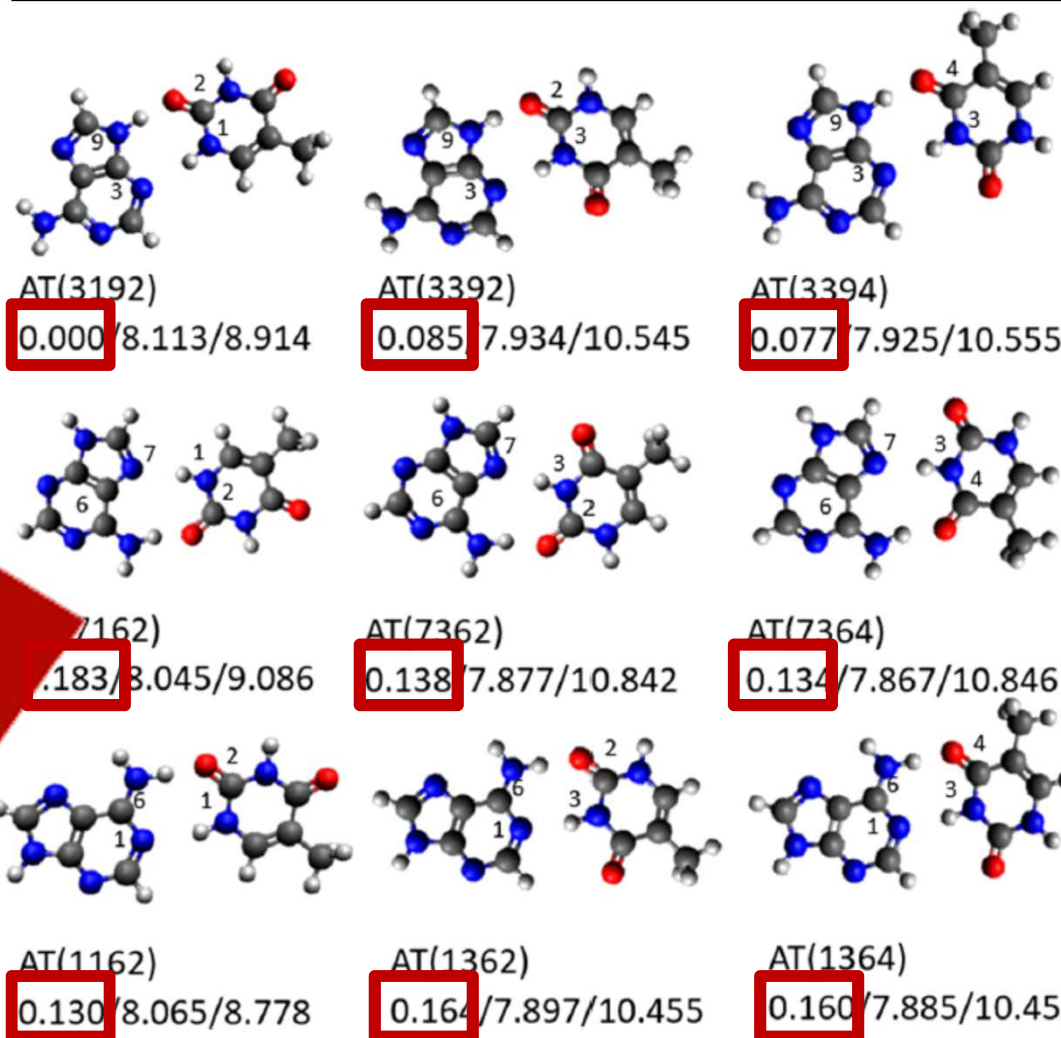


9H-adenine (9A)

+



Thymine (T)



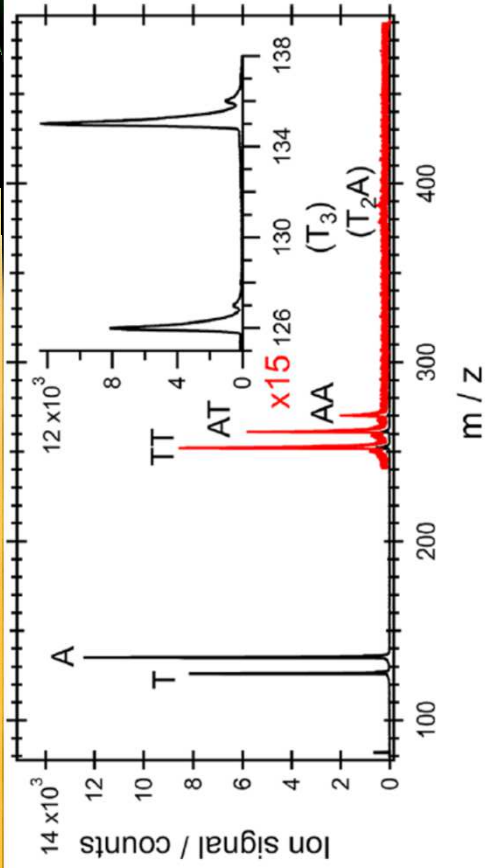


Figure 1. Representative mass spectrum of species in the molecular beam obtained while scanning the photon energy from 8 to 10 eV. Only trace amounts of trimers T_3 and T_2A can be observed.

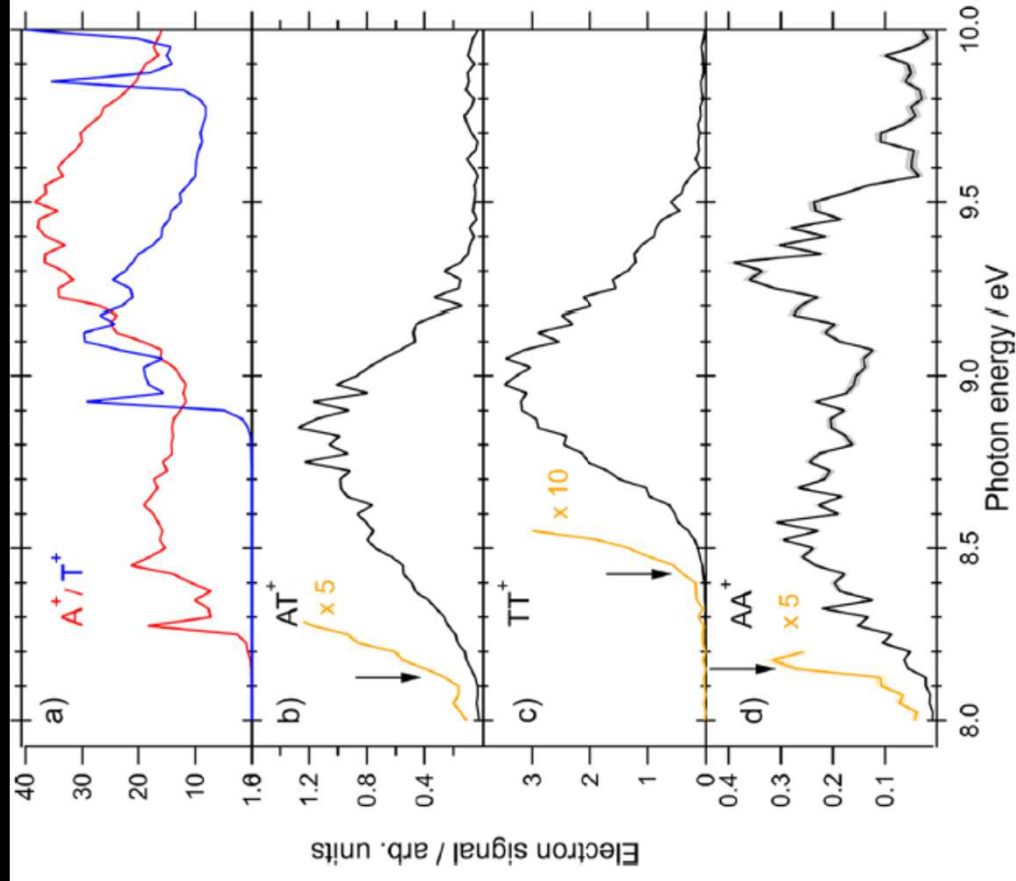


Figure 2. TPES of (a) adenine and thymine (red and blue lines, respectively), (b) the adenine-thymine heterodimer, and the (c) thymine and (d) adenine dimers.

- | | | |
|-------------------------------|---------------|------|
| $A \rightarrow A^+$ | m/z 135/136 | (1) |
| $AT \rightarrow T + A^+$ | m/z 135 | (2) |
| $AA \rightarrow A + A^+$ | m/z 135 | (3) |
| $AT \rightarrow T(-H) + AH^+$ | m/z 136 | (4) |
| $AA \rightarrow A(-H) + AH^+$ | m/z 136 | (5) |
| $T \rightarrow T^+$ | m/z 126/127 | (6) |
| $AT \rightarrow A + T^+$ | m/z 126 | (7) |
| $TT \rightarrow T + T^+$ | m/z 126 | (8) |
| $AT \rightarrow A(-H) + TH^+$ | m/z 127 | (9) |
| $TT \rightarrow T(-H) + TH^+$ | m/z 127 | (10) |

Krylov *et al.*
 The effect of π -stacking, H-bonding
 and electrostatic interactions on the
 ionization energies of nucleic acid
 bases

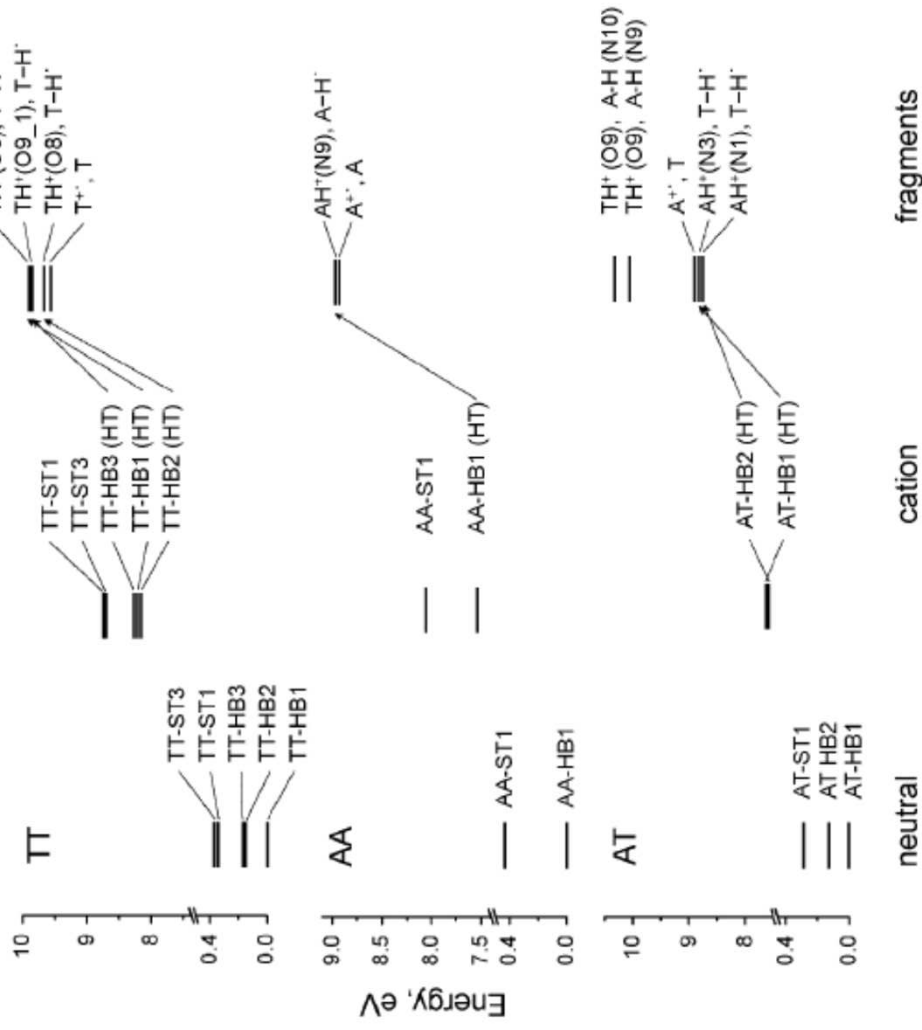


Fig. 13 Relative energies of the ionized dimers and the dissociation products with respect to the lowest energy neutral dimer for each system (TT, AA and AT). The energies are computed with ω B97X-D/6-311 + G(2df,2dp)// ω B97X-D/6-31 + G(d,p) and include ZPEs. The arrows indicate the channels for direct dissociation of the cationic H-transferred h-bonded dimers.

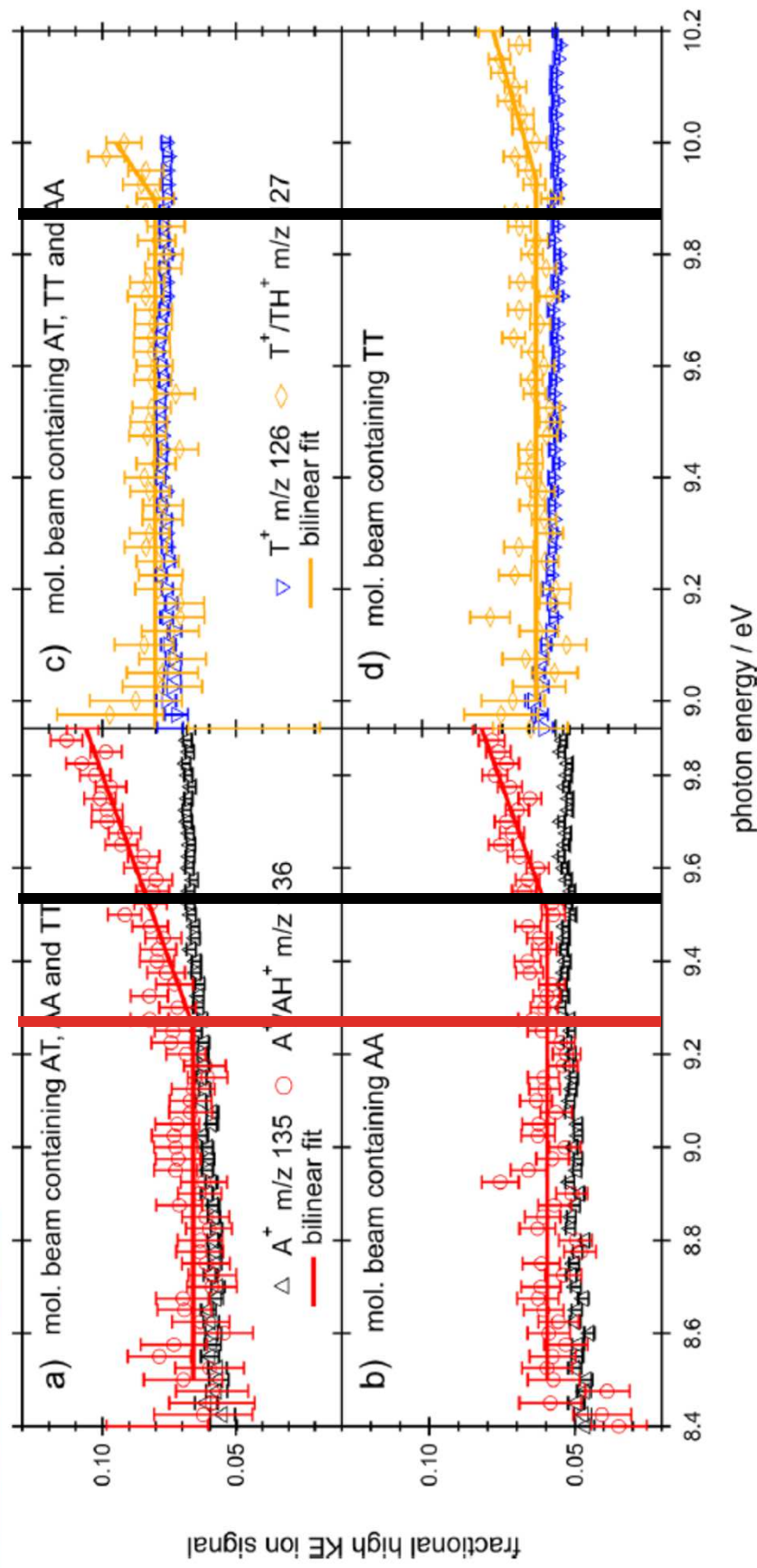
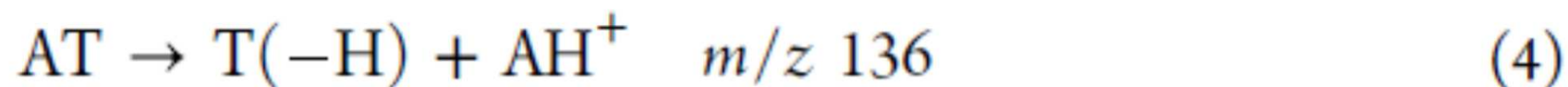


Figure 3. Fractional contributions of high-kinetic energy ions to the total ion signal of a given m/z as a function of the photon energy for a KER threshold fixed at 30 meV. (a and c) Data obtained for mixed molecular beams containing AT, AA, and TT clusters and (b and d) data obtained for pure molecular beams containing either pure A and its clusters or pure T and its clusters but no AT pairs, respectively.

Table 2. Summary of Computations for the Various AT Conformers and Comparison to Experimental Findings^a

	relative stability (eV)	neutral binding energy (eV)	adiabatic ionization energy (eV)	dissociative ionization energy of AH ⁺ (eV)	dissociative ionization energy of A ⁺
AT(3192)*	0.000	0.793	8.113	8.914	9.062
AT(3392)	0.085	0.713	7.934	10.545	8.982
AT(3394)	0.077	0.723	7.925	10.555	8.992
AT(7162)*	0.183	0.610	8.045	9.086	8.879
AT(7362)	0.138	0.655	7.877	10.842	8.924
AT(7364)	0.134	0.659	7.867	10.846	8.928
AT(1162)*	0.130	0.663	8.065	8.778	8.932
AT(1362)	0.164	0.629	7.897	10.455	8.898
AT(1364)	0.160	0.633	7.885	10.459	8.902
experimental			8.107	9.231 ± 0.050	

^aAll values were obtained at the (R)CCSD(T)-F12/aug-cc-pVDZ level after geometry optimization using the wB97xD level of theory and include zero-point vibrational energy correction at the wB97xD/aug-cc-pVDZ level. Asterisks mark the conformers in agreement with our experimental observations.



→ Unique AT conformer
 → Solely proton/hydrogen transfer upon
 dissociative photoionization

General comments and conclusions

→ Theory:

- Treatment of large or medium sizes (**complex**) using composite DFT-CCSD(T)-F12 based schemes.
- We have to develop new approaches for FC simulations to consider low frequency modes, anharmonic resonances, ...

→ Experiments are still needed to disentangle the complex processes...

→ NICE COMPLEMENTARY BETWEEN THEORY AND EXPERIMENT.



Atoms, Molecules and Clusters in Motion

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Jérôme Palaudoux (Sorbonne U)

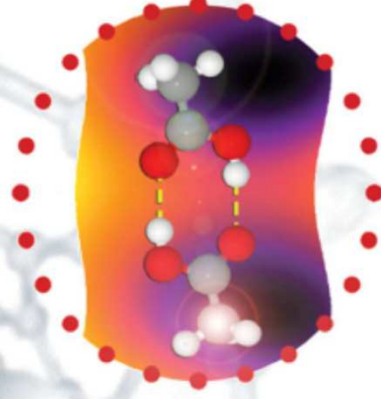
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ACKNOWLEDGMENTS

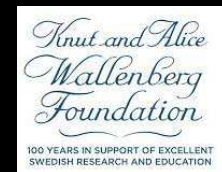


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THANK YOU FOR YOUR ATTENTION

CYTOSINE

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Communication

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Identifying Cytosine-Specific Isomers via High-Accuracy Single Photon Ionization

Ziyong Chen,[†] Kai-Chung Lau,^{*,†,‡} Gustavo A. Garcia,^{‡,§} Laurent Nahon,[‡] Dušan K. Božanić,[‡] Lionel Poisson,[§] Muneerah Mogren Al-Mogren,^{||} Martin Schwell,[⊥] Joseph S. Francisco,[#] Ayad Bellili,[▽] and Majdi Hochlaf^{*,▽,§}

Scheme 1. Nine Low-Lying Tautomer and Rotamer Isomers of Cytosine

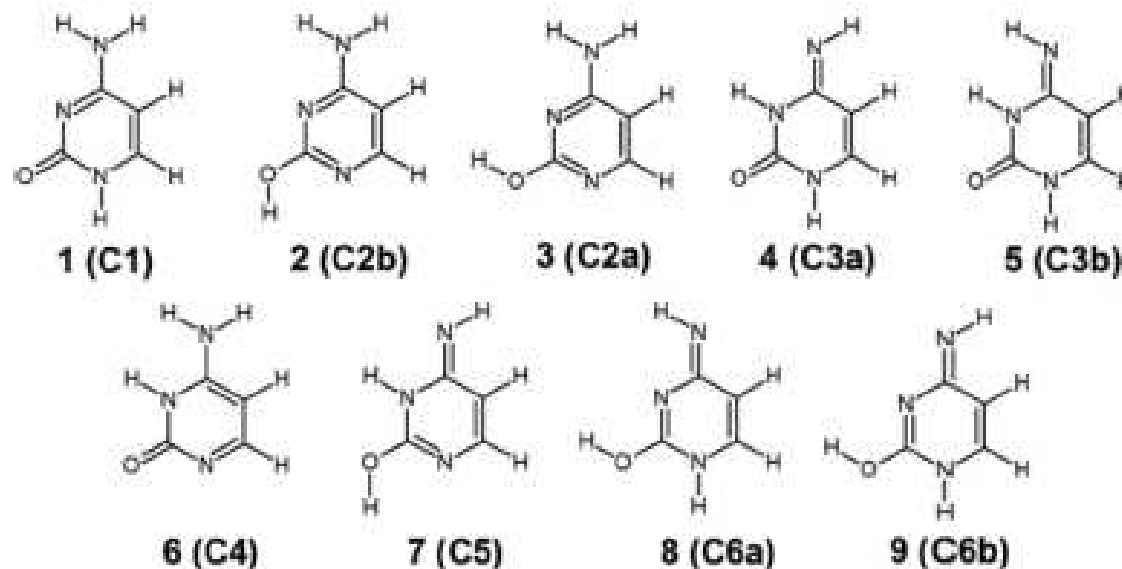


Table 1. Adiabatic Ionization Energies (in eV) of Cytosine Isomers As Deduced from the TPEPICO Spectrum and As Computed at the CCSD(T)-F12/cc-pVTZ-F12(+CV+SR+ZPVE) Level of Theory Using the Equilibrium Structures Optimized at the PBE0/aug-cc-pVDZ Level

	C1 → C1 [*]	C2a → C2a [*]	C2b → C2b [*]	C3a → C3a [*]	C3b → C3b [*]	C4 → C4 [*]	C5 → C5 [*]	C6a → C6a [*]	C6b → C6b [*]
	Theory								
this work	8.741	8.661	8.671	8.727	8.700	8.341	8.223	8.188	8.193
ref 6 ^d	8.71	8.62	8.64	8.58	8.64	8.31			
ref 8 ^b	8.64	8.51	8.53	8.67	8.64				
	Experiment								
this work	8.738 ± 0.003	8.652 ± 0.003	8.669 ± 0.003	8.717 ± 0.003	8.695 (not resolved)	no peak	no peak	no peak	no peak
VUV-SPI ^c					8.68 ± 0.05				
EI ^d					9.0 ± 0.1				
PES ^e					8.45				
VUV-SPI ^b					8.60 ± 0.05				
SPES ^f					8.66 ± 0.01				

^aCCSD(T), ZPE-corrected [B3LYP/6-31+G(d,p) frequencies]. ^bEOM-IP-CCSD/cc-pVTZ//IP-CISD/6-31+G(d), ZPE-corrected [ω B97X-D/6-31+G(d,p) frequencies]. ^cFrom ref 10. ^dFrom ref 9. ^eFrom ref 11. ^fFrom ref 14.

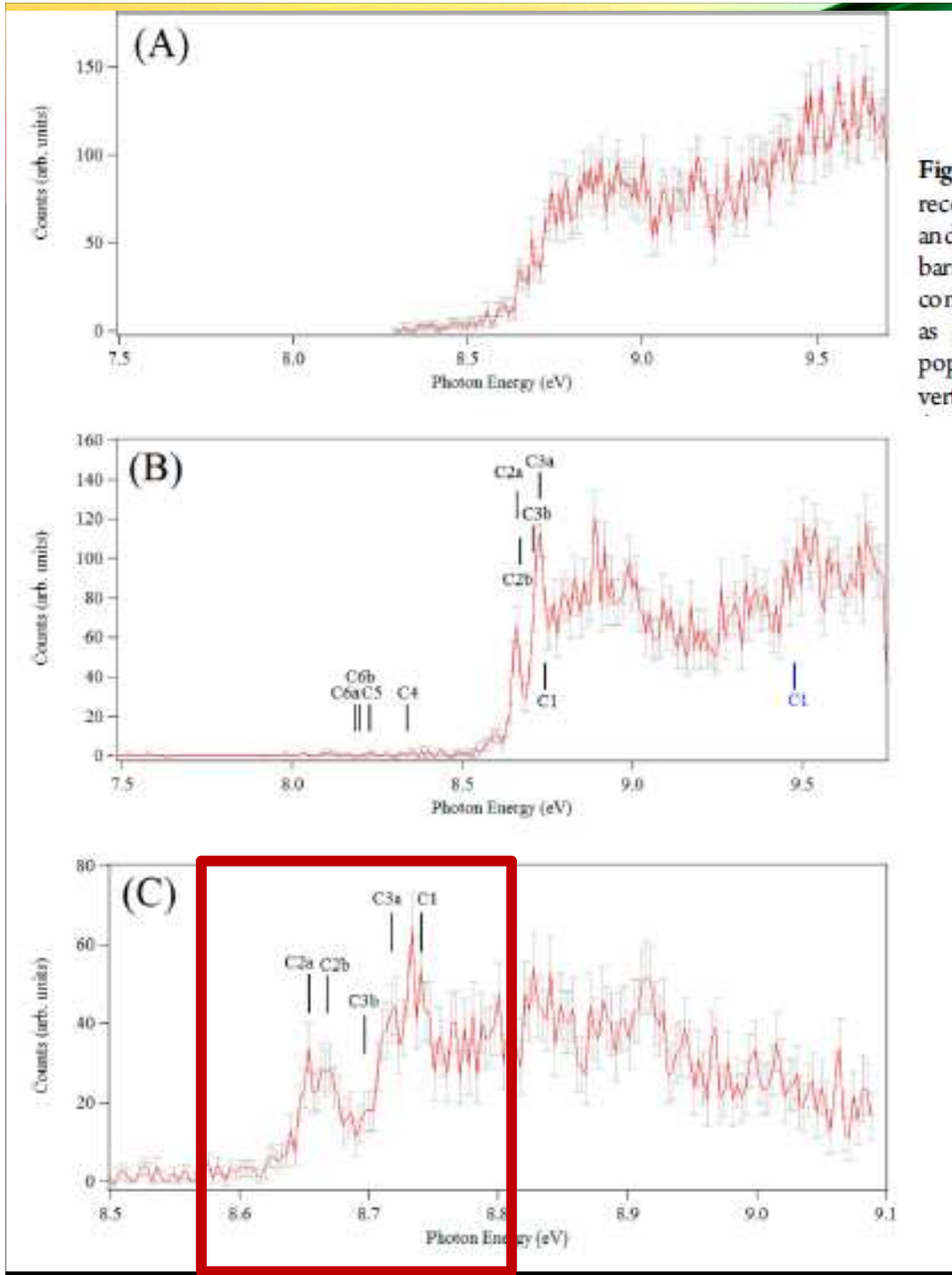
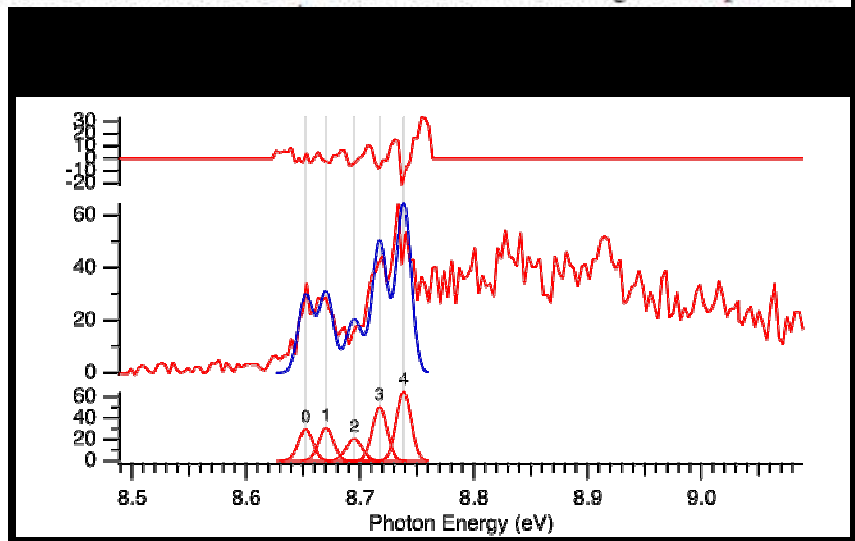


Figure 2. TPEPICO spectra of the cytosine parent ($m/z = 111$ amu) recorded using (A) an Ar backing pressure of 0.5 bar, (B) Ar at 3 bar and low resolution, and (C) Ar at 3 bar and high resolution. The error bars are shown in gray. In (B), the vertical combs correspond to the computed adiabatic ionization energies (AIEs) of cytosine tautomers as given in Tables 1 and 2. The black (blue) ones are for the population of the cationic ground (first excited) states. In (C), the vertical combs are for the AIEs derived from the fitting of the spectrum



Multi peak fitting of the high resolution spectrum. The upper trace is the residue.

THYMINE

THE JOURNAL OF
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Article
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Theoretical and Experimental Photoelectron Spectroscopy Characterization of the Ground State of Thymine Cation

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Vibrationally Resolved Photoelectron Spectroscopy of Electronic Excited States of DNA Bases: Application to the \tilde{A} State of Thymine Cation

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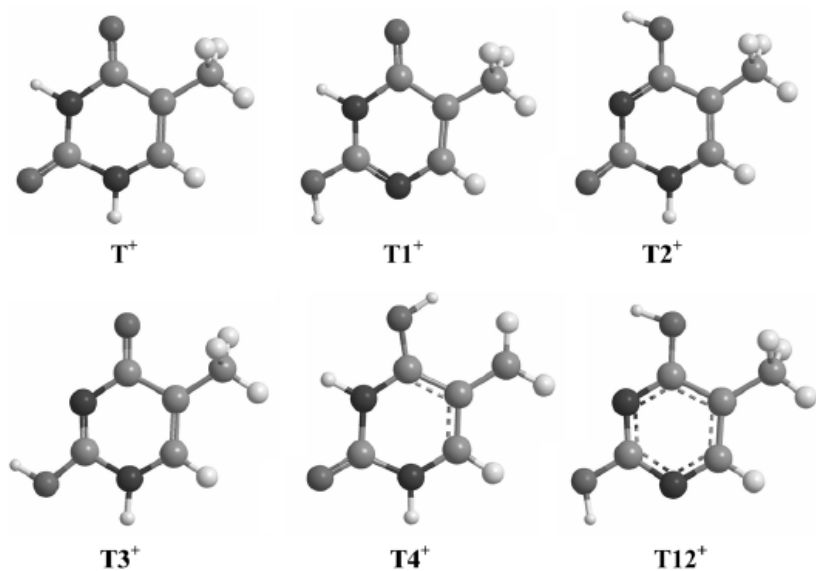
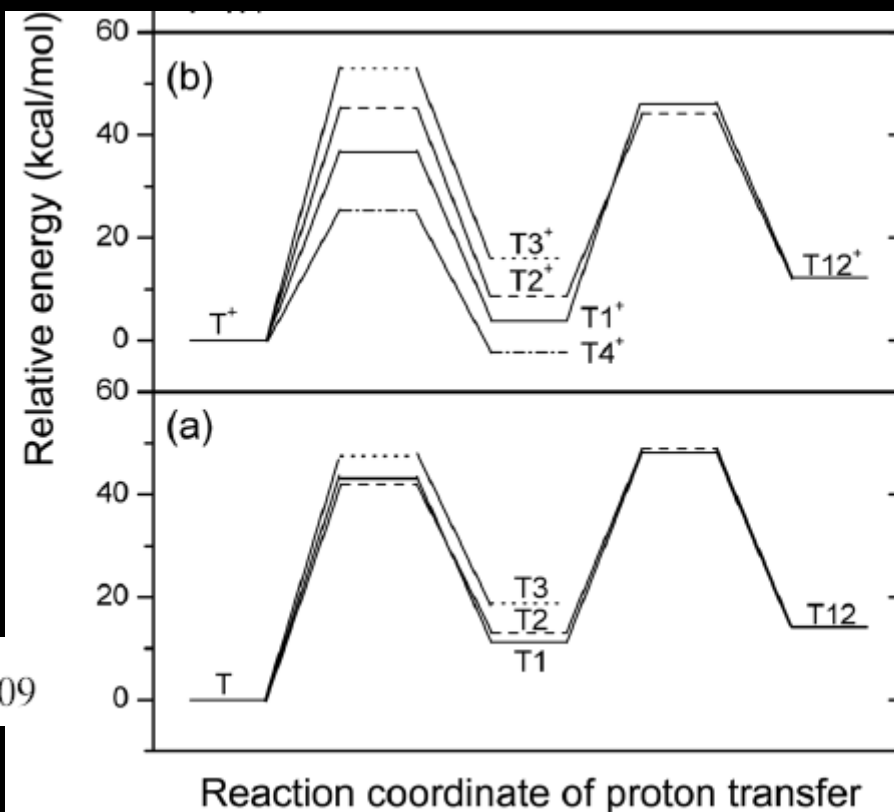


Figure 2. The structures of the tautomers of thymine radical cation optimized at the B3LYP/6-311++G(d,p) level.

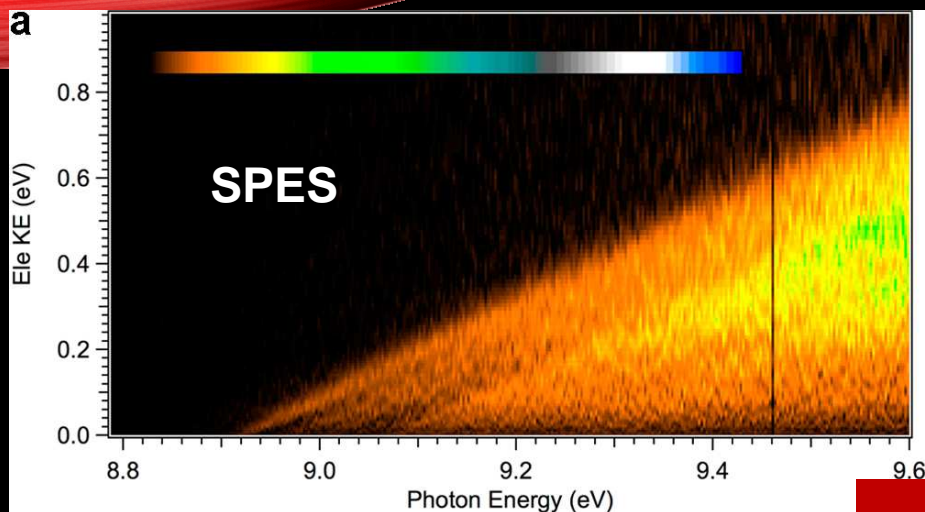
Nam Joon Kim

Bull. Korean Chem. Soc. **2006**, Vol. 27, No. 7 1009

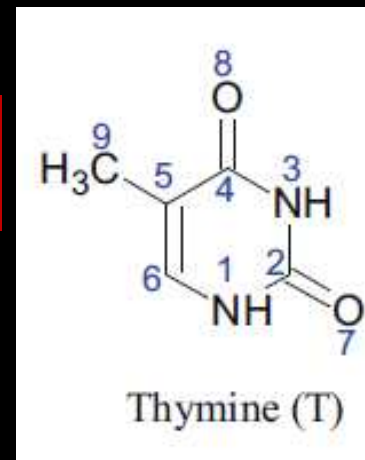
Neutral	ΔE_0^b	kc
T	0.00	
T1	11.19	
T2	13.08	
T3	18.76	
T12	14.17	



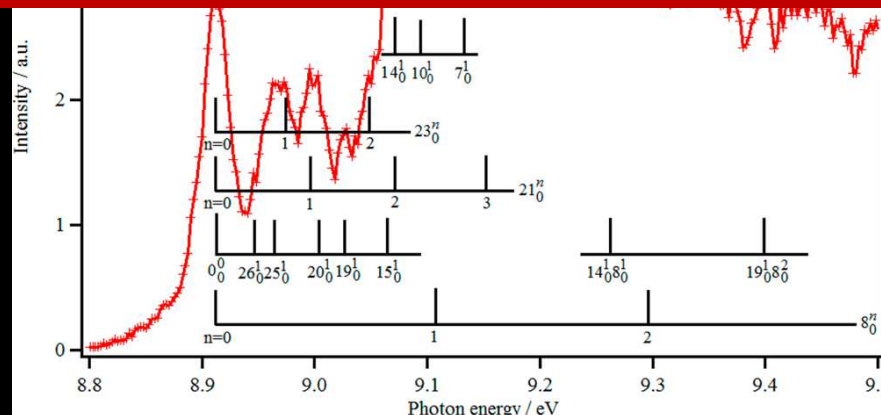
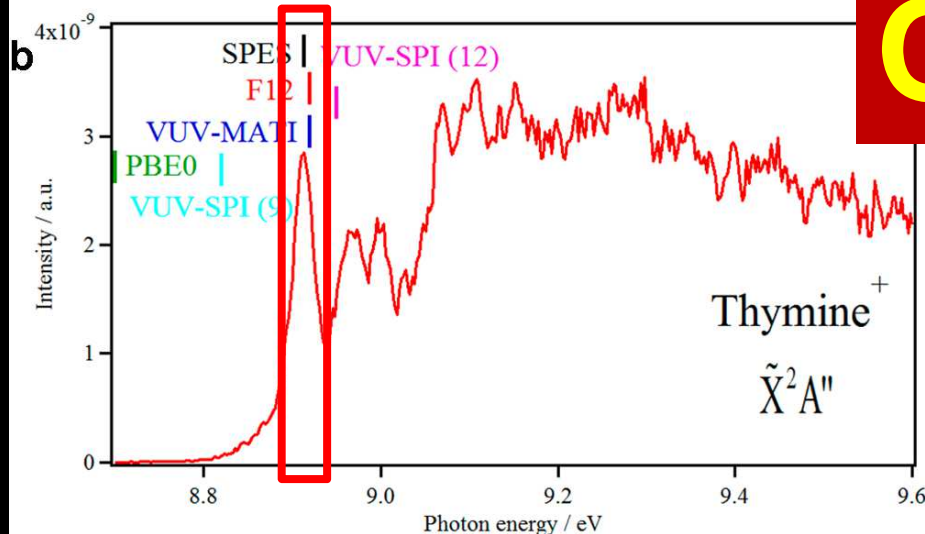
Thymine⁺ - cation ground state



SOLEIL



One tautomer



Theo: **F12** = (R)CCSD(T)-F12cc-pVTZ-F12
PBE0 (DFT) = PBE0/aug-cc-pVDZ+ZPVE
 Exptl: **VUV-SPI**, **VUV-SPI**, **VUV-MATI**, SPES

Majdi et al., JPC. A (2015)



Identification of DNA Bases and Their Cations in Astrochemical Environments: Computational Spectroscopy of Thymine as a Test Case

Yage Zhao¹, Majdi Hochlaf² and Malgorzata Biczysko^{1*}

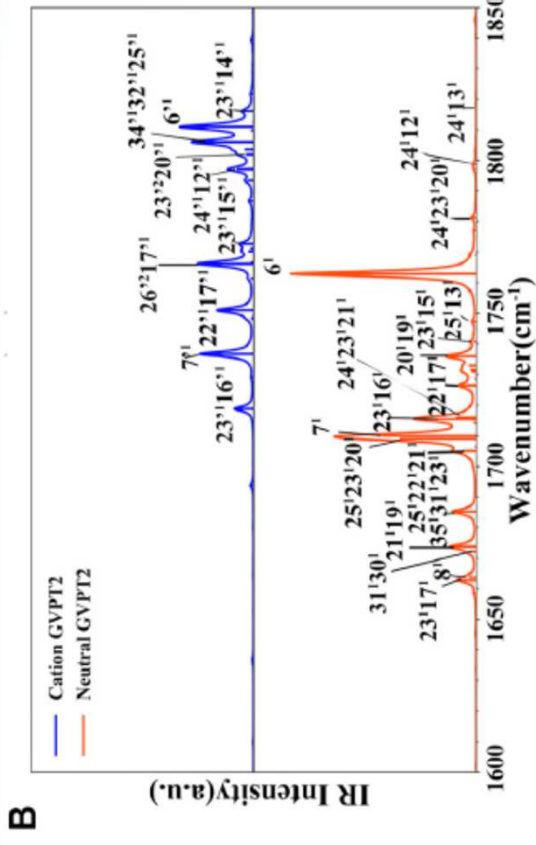
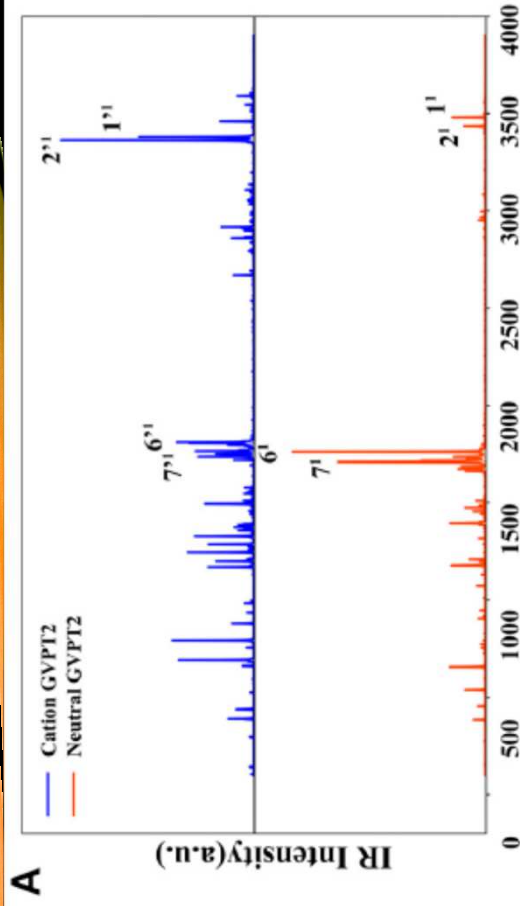
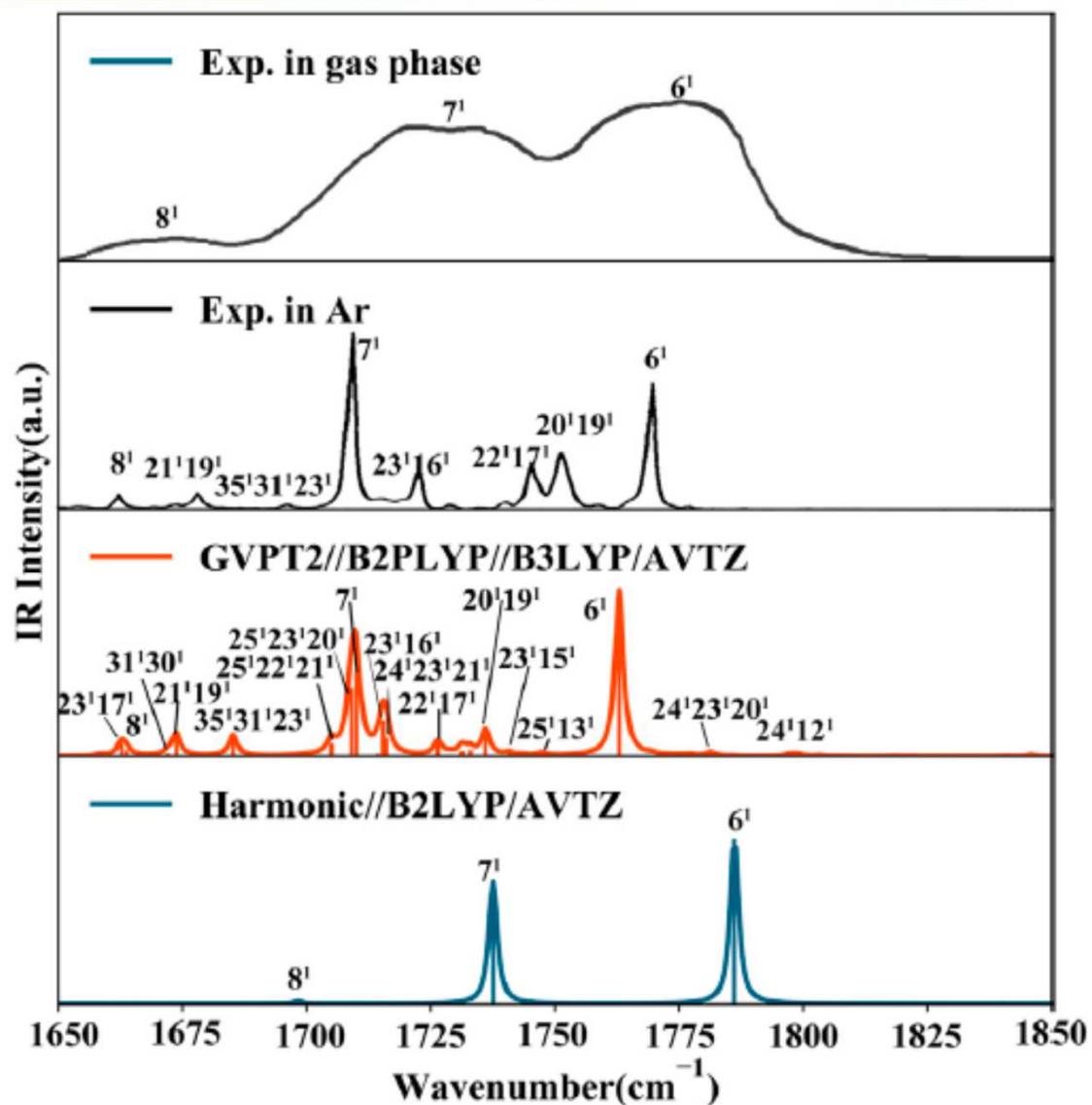


FIGURE 6 | MIR spectra simulated at GVPT2//B2PLYP//B3LYP/AVTZ level. Spectra line-shapes were broadened by Lorentzian function with $\Gamma = 0.1 \text{ cm}^{-1}$. **(A)** $100 \text{ cm}^{-1} - 4000 \text{ cm}^{-1}$ range **(B)** The $1600 \text{ cm}^{-1} - 1850 \text{ cm}^{-1}$ range Most intense bands are assigned as m^n where m is normal mode and n number of quanta (1 fundamentals, 2 overtones).



Theo. vs. Exp. 1: IR region

FIGURE 5 | IR spectrum of thymine in the CO stretching range (1650 cm⁻¹ - 1850 cm⁻¹). Computed stick spectra were broadened by Lorentzian functions with half width at half-maximum (HWHM) of 1 cm⁻¹. Most significant contributions to intense bands are assigned as m^n where m is the normal mode and n is the number of quanta on this mode (1 fundamentals, 2 overtones). Experimental spectra of gas phase thymine (Colarusso et al., 1997) and of Ar low temperature matrix trapped thymine (Szczepaniak et al., 2000) are shown for comparison.

Theo. vs. Exp. 2: VUV region

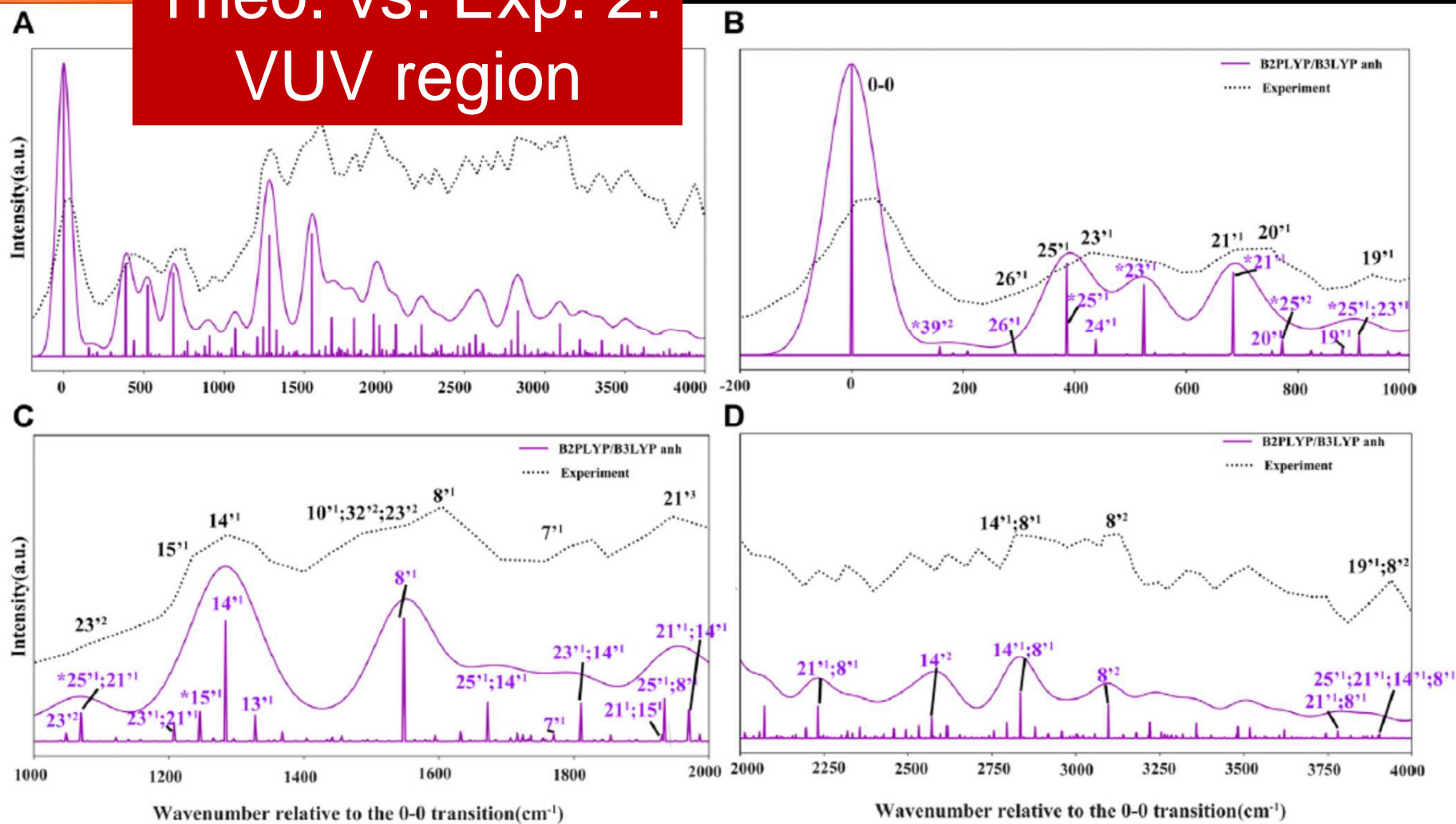


FIGURE 7 | Photoelectron TI AH|FC spectrum (60 K) and SPES experimental spectrum (Majdi et al., 2015). The computed spectrum is represented as single transitions, with most intense ones assigned as m^n , which represents the final vibrational state with n quanta associated to mode m . The convoluted spectra line-shapes have been obtained by broadening with Gaussian function with HWHM of 50 cm⁻¹. Bands observed also in the VUV-MATI experimental spectrum (Choi et al., 2005; Bravaya et al., 2010) are marked by asterisk (*).

Thymine Theo.

Thymine + Theo.

Tholins Exp.

Application:
Comparison to
Tholins spectra

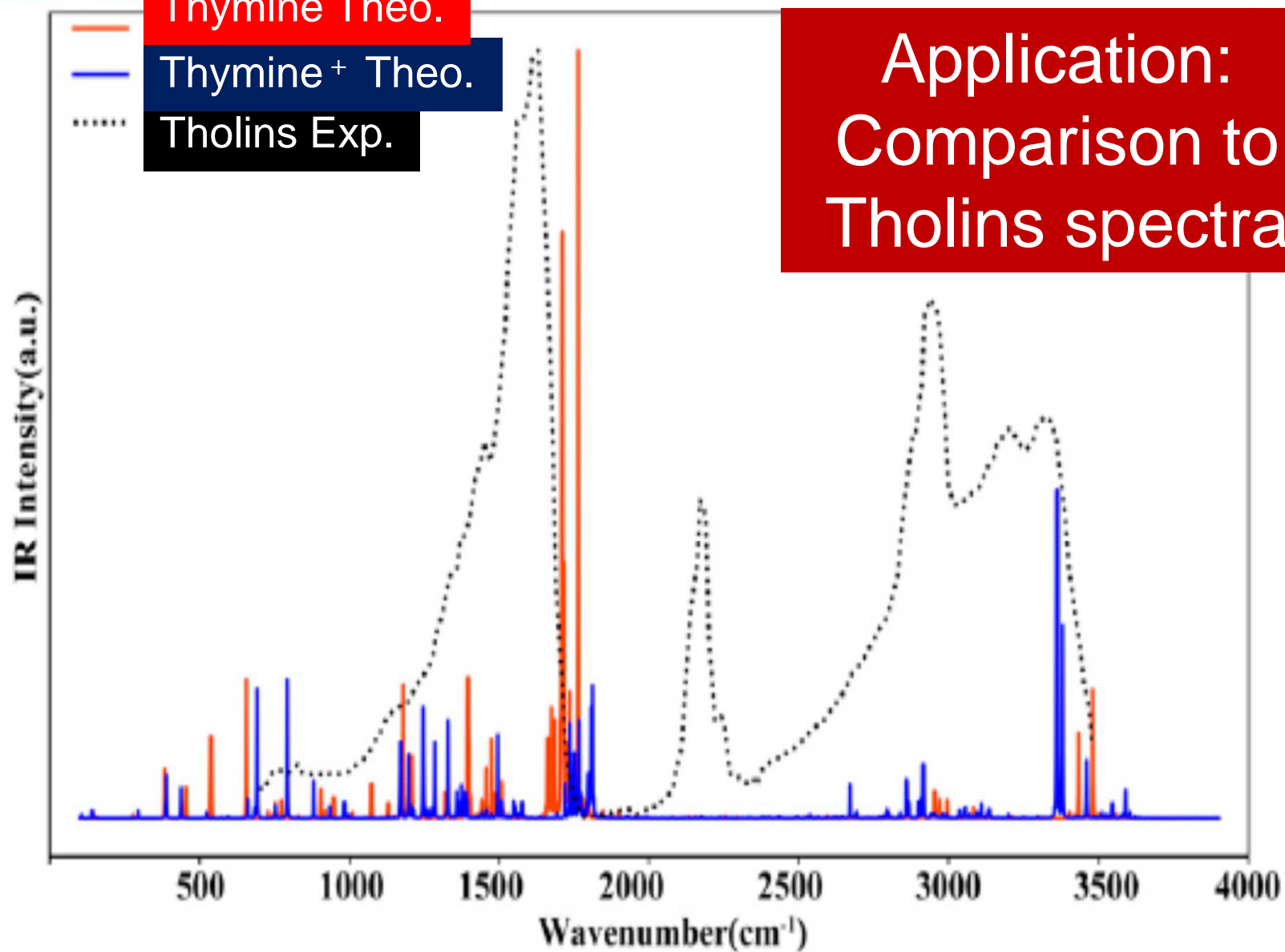


FIGURE 8 | Comparison of the simulated IR spectrum of thymine with the experimental spectrum of Tholins (Gautier et al., 2012). GVPT2//B2PLYP//B3LYP/AVTZ stick spectra were broadened by Lorentzian functions with HWHM of 1 cm⁻¹.