

Activation and spectroscopy of mass and charge selected ions

Alexandre Giuliani Synchrotron SOLEIL & INRAE



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- Spectroscopy on charged species: is it relevant?





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Number of proteins in the UniprotKB/swiss protein database versus pl

The isoelectric point (pI), is the pH at which a particular molecule carries no net electrical charge.

Figure 1. Number of proteins in the UniProtKB/Swiss-Prot database versus pl value, retrieved with the Tagldent tool (http://www.expasy.ch/tools/tagident.html) in June 2010.



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- Spectroscopy on ions
- Control over the target:
 - mass and charge selected species

























[1•2]⁷⁺ [1•3]⁷⁺ [1•1]⁷⁺ [1•2]⁸ [1•4]⁷⁺ [1•2]⁶⁺ [1.0] [1•1]⁶⁺ [1•5]7+ [1•3] [1•3]⁶⁺ [1•0]⁶⁺ [1•4]⁶ [1•4]⁸⁺ [1•8]^{7+[1•5]⁶⁺} [1•6]⁷⁺ [1•7]⁷⁺ [1•6]⁶⁺ [1•7]⁶⁺ 1000 1200 1400 1600 1800 m/z

ESI-MS of a protein interacting with its ligand. Mixture of stœchiométries (1•0 à 1•8) and charge states (6+ à 8+).



Experimental setup



Thermo LTQ XL

A. Milosavljević, C. Nicolas, J. Lemaire, C. Dehon, R. Thissen, J.-M. Bizau, M. Réfrégiers, L. Nahon and A. Giuliani, Phys Chem Chem Phys 13 (2011) 15432–6



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SRMS2 @ DESIRS







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Normalized ion abundance of [M+nH]^{(n+1)+•}

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Outline

• VUV activation of oligosaccharides

• Serine dimer







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Nomenclature



Nomenclature from Domon et Costello (1988)







Nomenclature



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Nomenclature



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DP3Me3









DP3Me3



















Relative Abundance



























SYNCHROTRON





SYNCHROTRON









DP5Me3















m/z



DP5Me3









DP5Me3

Me







DP5Me3





DP5Me3

Me







DP5Me3

Me

?

Me







DP5Me3











Liquid chromatography coupling







Liquid chromatography coupling





Intracluster bond formation

• VUV activation of ligosaccharides

Intracluster bond formation: the case of serine dimer





LE-CID of protonated serine dimer





Intracluster bond formation

Intracluster bond formation was never observed using LE-CID. \rightarrow Heating of the cluster leading to statistical products (evaporation more likely than ICBF)





Intracluster bond formation



- Excited states dynamics which can possible results in different, non statistical products

 \rightarrow ICBF has been reported at 157 nm irradiation (JACS 2011, 133, 15834)

- Ability to deposit a well defined amount of energy into the system
- Tunable source : identify the excited states involved



LE-CID versus Photon activation



ICBF region





- \rightarrow m/z 193 low abundant (<1%) below 7 eV
- \rightarrow PBF is present at higher energy







→ PBF is present at higher energy (above 10 eV)
Fragmentation of diserine is a two steps process:
- either PBF followed by fragmentation

- or fragmentation followed by bond formation







- States S_2 , S_3 , S_4 , S_5 and S_{12} relaxes into stable minima that could further evolve towards PBF
- PBF in the ground state S_0 implies an energetic barrier of 2.48 eV, compared to 0.5–1.6 eV from the excited states.

Min A: K Min B: K Min B1: K Min C: Min C1:



PBF

PBF sterically hindered

O. Licht et al. Angew. Chem. Int. Ed. 2023, 10.1002/ange.202218770.



- VUV activation of oligosaccharides
 - Better suited than CID
 - Brilliance of SR makes it compatible with LC
- Intracluster bond formation
 - Evidence for peptide bond within a cluster





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Thank you for your attention



