Induced Photoelectron Circular Dichroism in an Achiral Chromophore



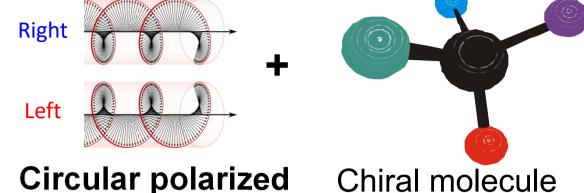
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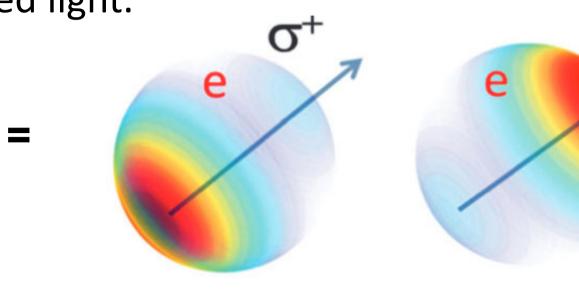


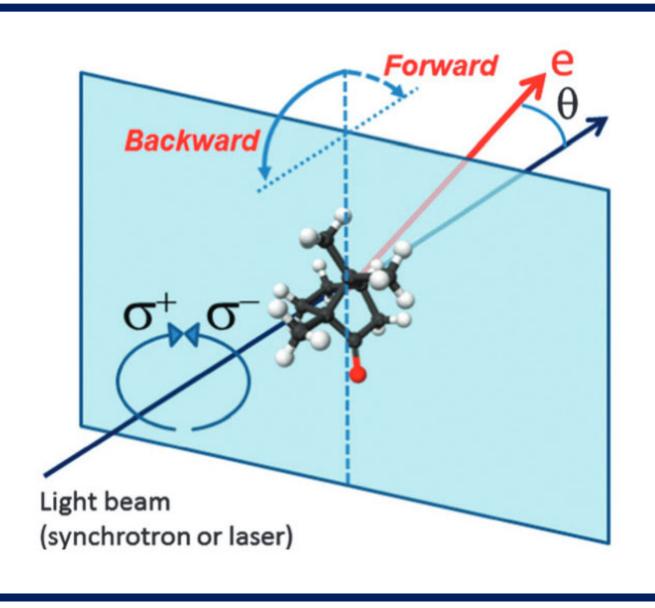
PHOTOELECTRON CIRCULAR DICHROISM (PECD)

PECD consists in a forward/backward asymmetry with respect to the light propagation axis in the photoelectron angular distribution after ionization of a chiral molecule by a circularly polarized light.



Chiral molecule

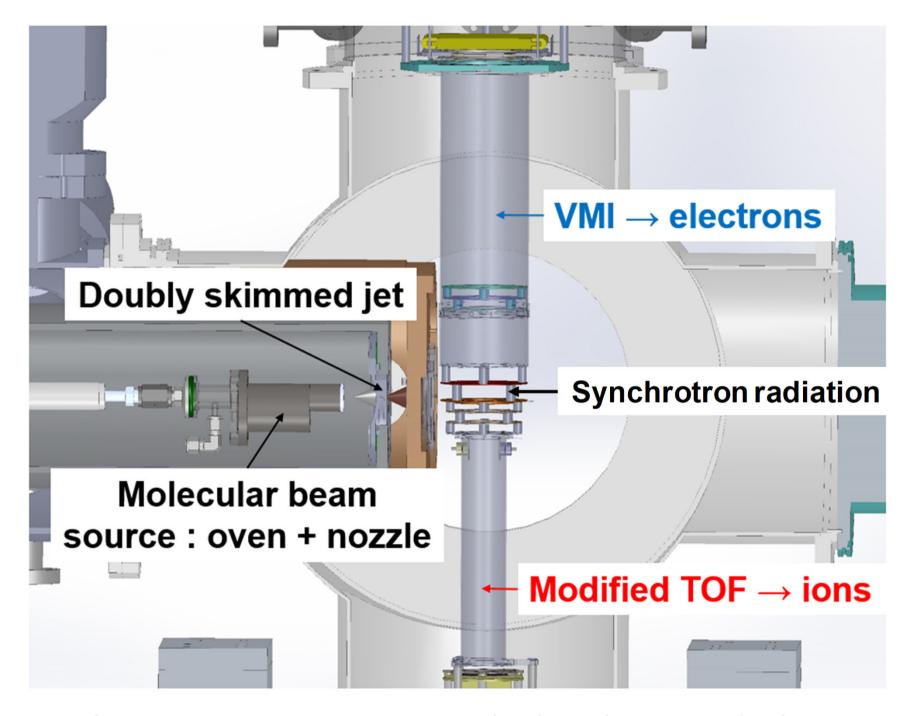




- **⇒** Absolute configurations determination
- ⇒ Sensitivity to conformation **Conformational analysis**
- ⇒ Sensitivity to molecular interactions : Microsolvation, hydrogen bond formation

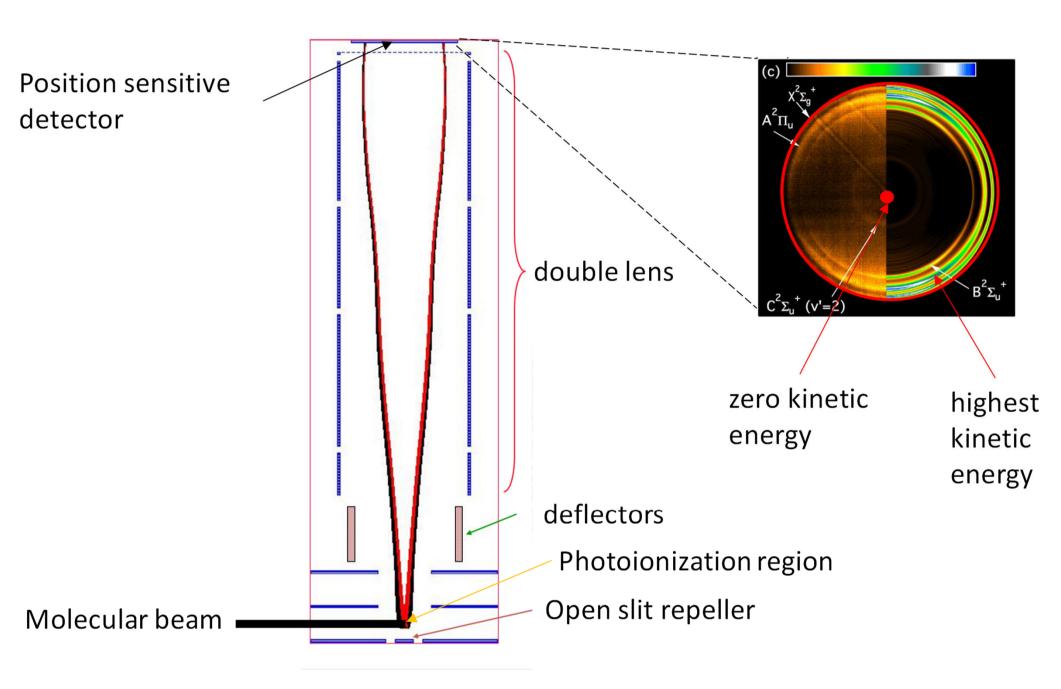
EXPERIMENT

light (CPL)



The SAPHIRS experimental chambers and the DELICIOUS3 spectrometer DESIRS Beam line of Synchrotron SOLEIL

VMI (Velocity Map Imaging)



- > Focusing of the electron as a function of their kinetic energy
- Microchannel plates

THEORY

Angular distribution of photoelectrons ejected from a randomly oriented molecule by single photon excitation process:

$$I_p(\theta) = 1 + b_1^{\{p\}} P_1(\cos \theta) + b_2^{\{p\}} P_2(\cos \theta)$$

p = 0 for linear polarization

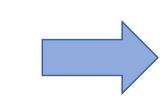
p = ±1 for left and right circular polarization

For achiral molecules,

PECD for Phe:(S)MOx

Phe:MOx

$$b_1^{\{\pm 1\}} \neq 0 \text{ and } b_1^{\{\pm 1\}}(R) = b_1^{\{-1\}}(S)$$



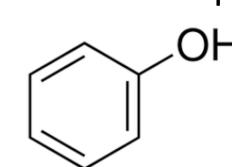
 $PECD = 2b_1$ > from one to a few tens %

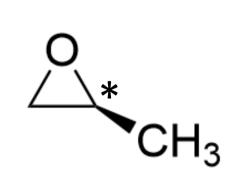
> depends on the initial orbital and the final state (photoelectron scattering by the molecular potential of the cation)

Complex studied

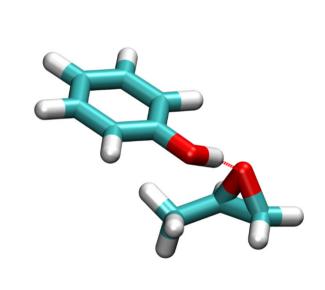
Phenol (Phe) = achiral chromophore

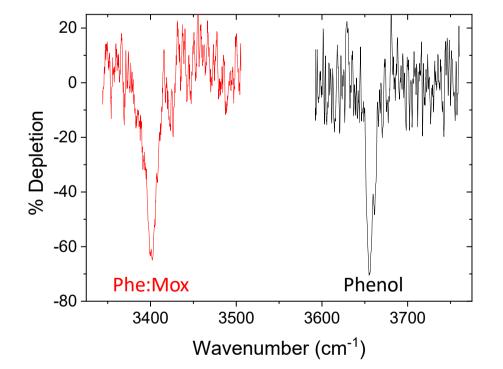
Methyloxirane (Mox) = chiral molecule





Bound by strong hydrogen bond OH...O



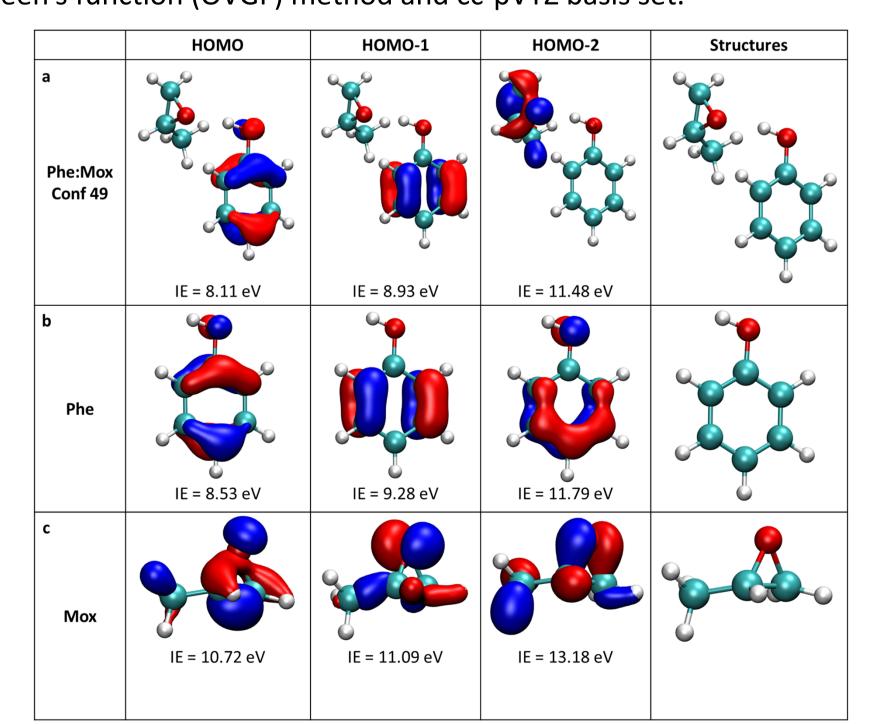


Most stable structure of the Phe-MOx complex calculated B3LYP-D3BJ/6-311++G(d,p) level of theory

Vibrational spectrum of isolated phenol and of the Phe-MOx complex recorded by IR-UV double resonance spectroscopy

Isosurfaces of occupied orbitals for Phe:Mox, Phe and MOx calculated at the MP2/6-31++G(d,p) level.

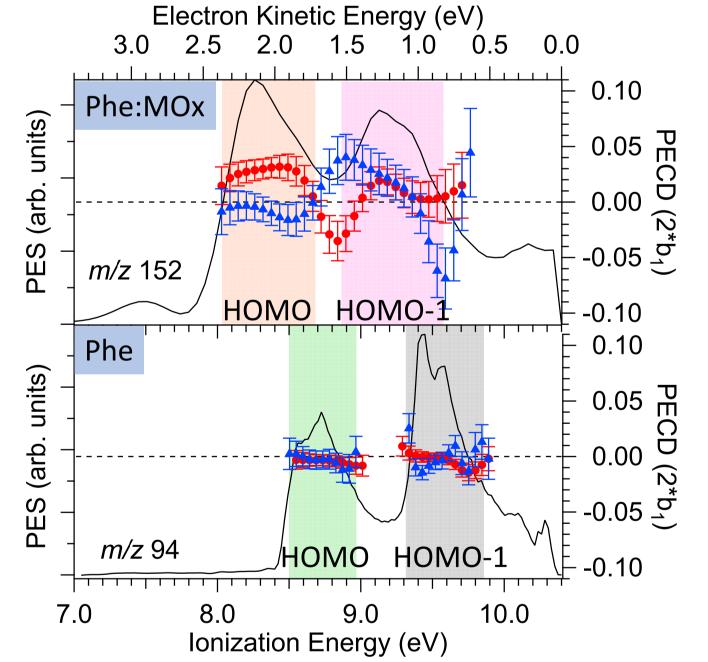
Vertical ionisation energies (VIE) calculated with the outer valence Green's function (OVGF) method and cc-pVTZ basis set.



HOMO & HOMO-1 of the complex localized on the phenol part, Mox moiety being a mere spectator

PECD Results

hv = 10,4 eV < EI (Mox)Electron Kinetic Energy (eV) 2.0 1.5 1.0 0.5 0.0 Phe:MOx 0.10 Phe:MOx PES (arb. units) 0.05 문 0.00 D $-0.05 \frac{6}{3}$ *m/z* 152 HOMO HOMO-1 -0.10 0.10 0.05 꿈 PES (arb. 0.00 $-0.05 \frac{6}{1}$ номо номо-1 -0.10 10.0 Ionization Energy (eV)



- ✓ PECD of Phenol = 0
- Mirorring effect between PECD of both enantiomers of the complex
- PECD around 3% for the complex

$$I_p(\theta) = 1 + b_1^{\{p\}} P_1(\cos \theta) + b_2^{\{p\}} P_2(\cos \theta)$$

 $b_1^{\{p\}} = 0$

For chiral molecules,

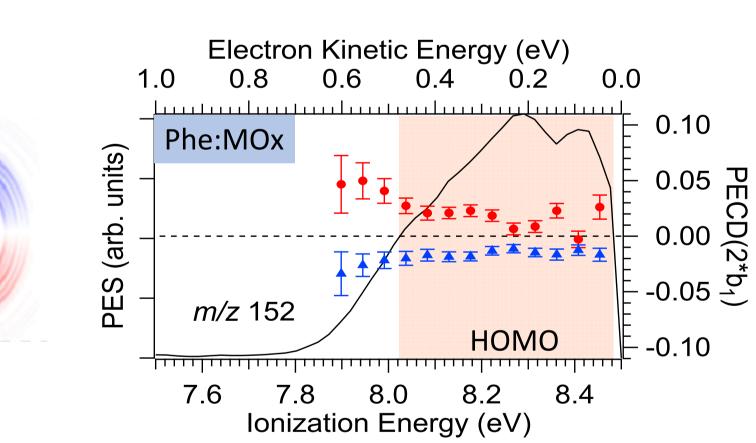
$$b_1^{\{\pm 1\}} \neq 0 \text{ and } b_1^{\{\pm 1\}}(R) = b_1^{\{\pm 1\}}(S)$$

PECD for Phe:(R)MOx the velocity map imaging (VMI) spectrometer.

hv=8,5 eV < EI (Phe) or EI (Mox)

Raw images (left part) and Abel-inverted (right

part) difference (LCP-RCP) images obtained with

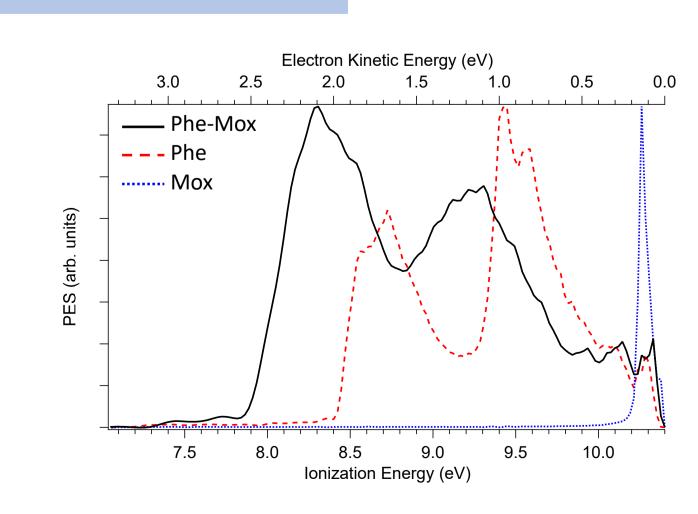


Induced PECD on orbitals of achiral phenol



chiral potential

PES Results



Comparison of Photoelectron spectra (PES) of phenol, MOx and the Phe:MOx complex, recorded at a photon energy of 10.4 eV, slightly above the adiabatic ionisation energy of MOx (10.24 eV).

HOMO & HOMO-1 of the complex lower in energy than those of phenol

CONCLUSION AND AFTER...

- > First example of the manifestation of induced chirality in PECD on the HOMO & HOMO-1 orbitals of an achiral chromophore by complexation with a chiral tag.
- > Complexation with the chiral MOx host offers the phenol departing electron a chiral scattering potential.
- > Role of the long-range chiral scattering potential extending to intermolecular distances, here up to ~5 Å
- > Laser chiroptical measurements possible of chiral species which cannot be excited in UV thanks to complexation with a achiral chromophore
- > Induced chirality = important structural and analytical implications
- > REMPI-PECD at ISMO to record conformer-selective PECD