

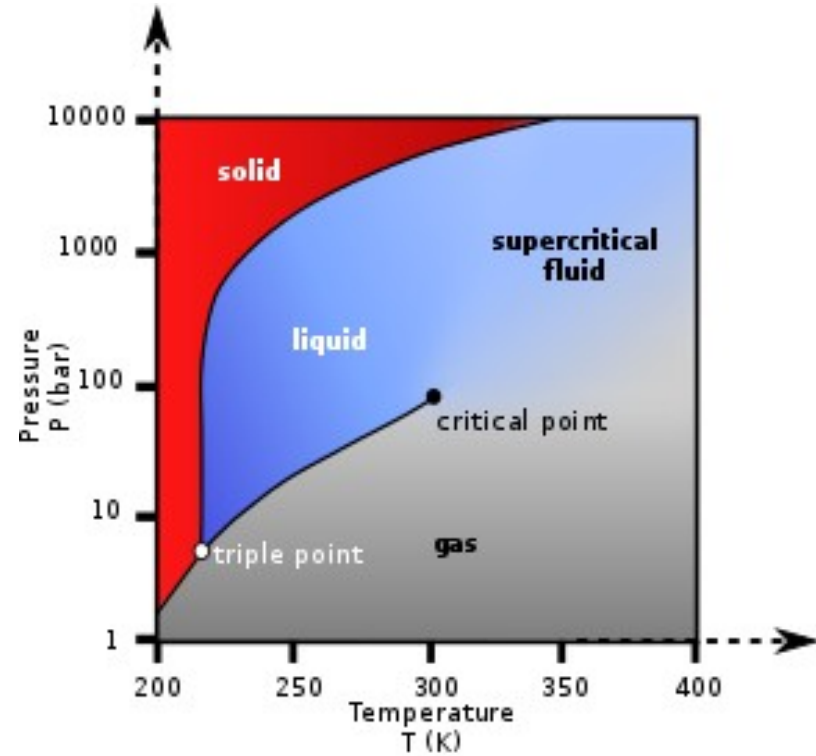
Classical density functional theory: Application to supercritical CO₂

Mohamed Houssein Mohamed, Francesca Ingrosso, Antoine Carof

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Why supercritical CO₂ ?

Supercritical Fluid (SCF): above the critical point!



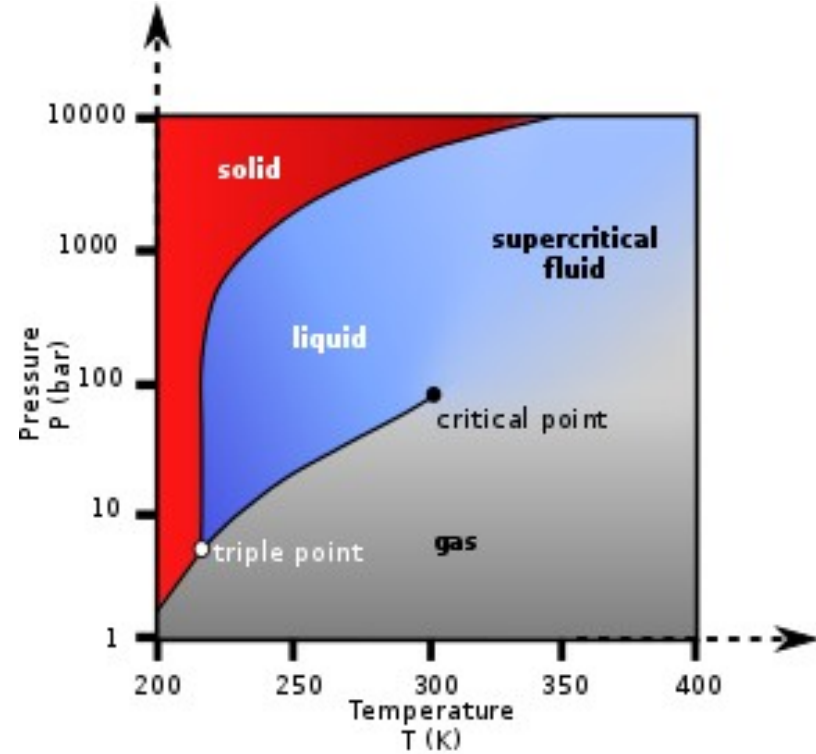
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Density sensitive to small changes of pressure



Used as a solvent



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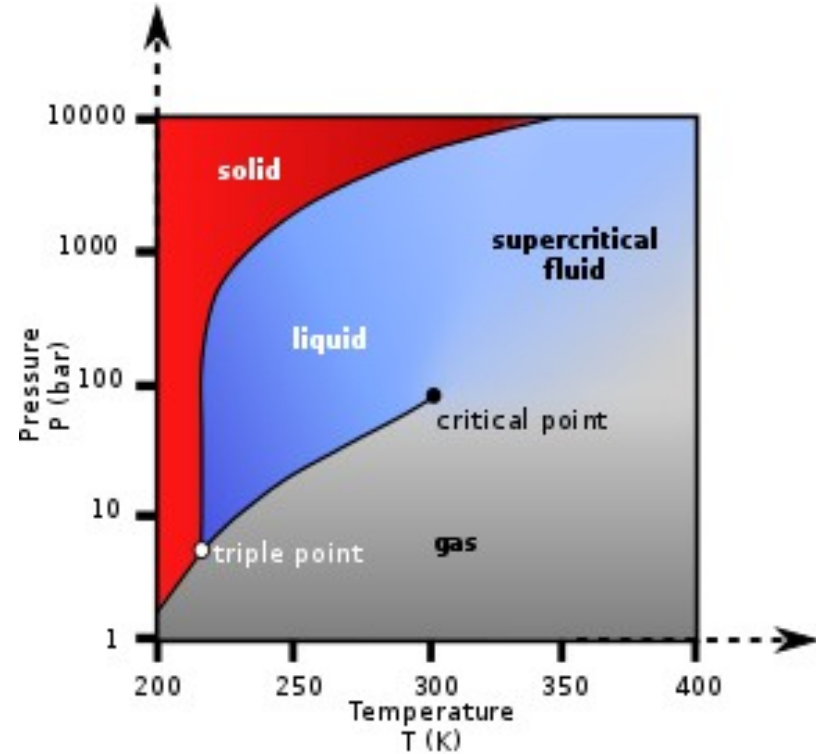


Used as a solvent

Supercritical CO₂

- Accesible ($T_c=35^\circ\text{C}$ et $P_c= 79$ atm)
- Abundant
- Non-toxic

Green solvent!



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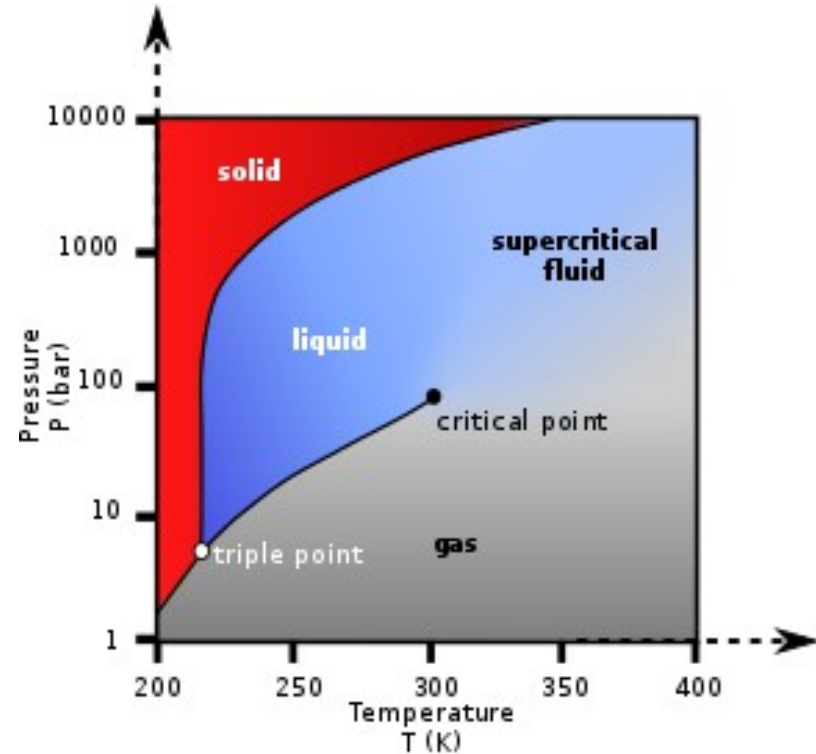
Green solvent!

Application: Decaffeination



But not efficient....



How to predict the solubility for a new molecule?





How to calculate the solubility?

	Simulation	Empiric
Efficiency		
Cost		
Flexibility		





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We need a fast and accurate method

Classical density functional theory (cDFT)

Free energy functional: $\mathcal{F}[\rho] = \Omega[\rho] - \Omega[\rho_0]$ **Not electronic DFT!**

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cDFT works well for small & rigid molecule \longrightarrow **Perfect for CO₂!**

How to construct the free energy functional?

Classical density functional theory (cDFT)

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$$\mathcal{F}[\rho] = \mathcal{F}_{\text{id}}[\rho] + \mathcal{F}_{\text{ext}}[\rho] + \mathcal{F}_{\text{exc}}[\rho]$$

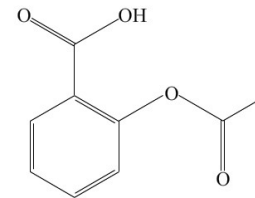
Ideal part
(Entropy)

Solute-
solvent
interaction

Solvent-
solvent interaction \longrightarrow Unknown

Litterature: $\mathcal{F}_{\text{exc}}[\rho] = \text{Hard sphere} + \text{Attrative potential}$

CO₂ and solute as a sphere



Objectives: cDFT with good accuracy + molecular description + evolution with P,T

Construction of $\mathcal{F}_{\text{exc}}[\rho]$

Molecular DFT

$$\rho = \rho(\mathbf{r}, \omega)$$

Position + orientation!

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$$\mathcal{F}_{\text{exc}}[\rho(r, \omega)] = -\frac{1}{2}\beta^{-1} \int d1d2c(12) \Delta\rho(1) \Delta\rho(2) + \mathcal{F}_B$$

quadratic term beyond the second order

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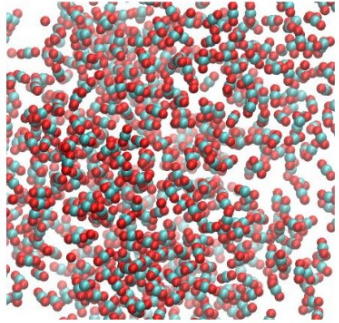
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Direct correlation function

MD
simulations



$$\rho = 0.8 \rho_c$$

$$T = 1.05 T_c$$

13500 CO₂

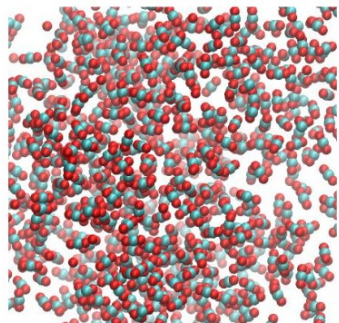
60 ns

Direct correlation function

MD
simulations



Pair correlation
function

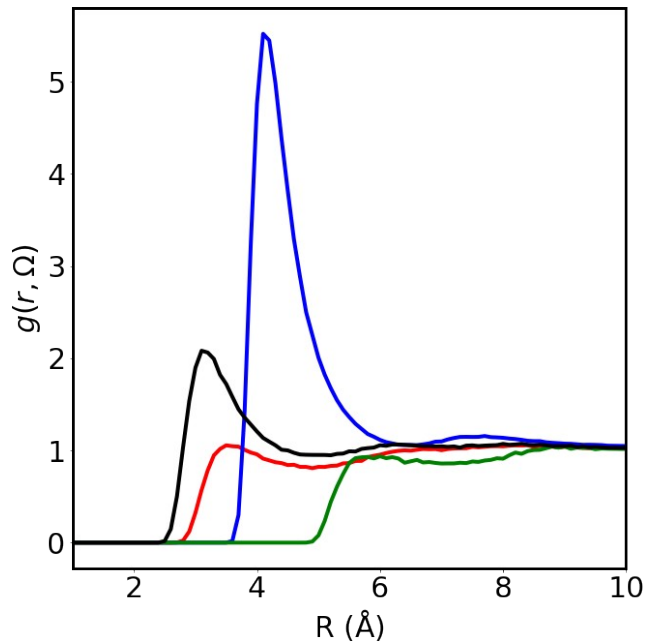


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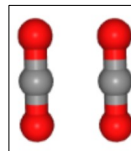
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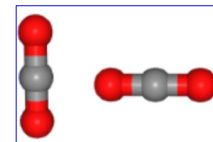
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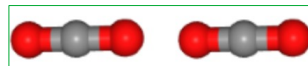
1



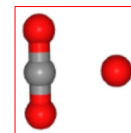
2



3



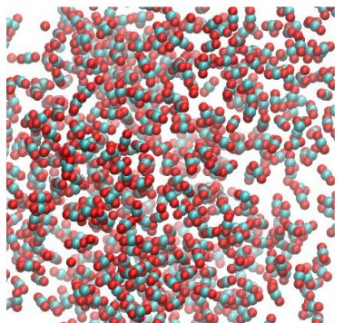
4



Direct correlation function

$$c(12) = g(12) - 1 - \rho_0 \int (g(13) - 1) c(32) d3$$

MD
simulations



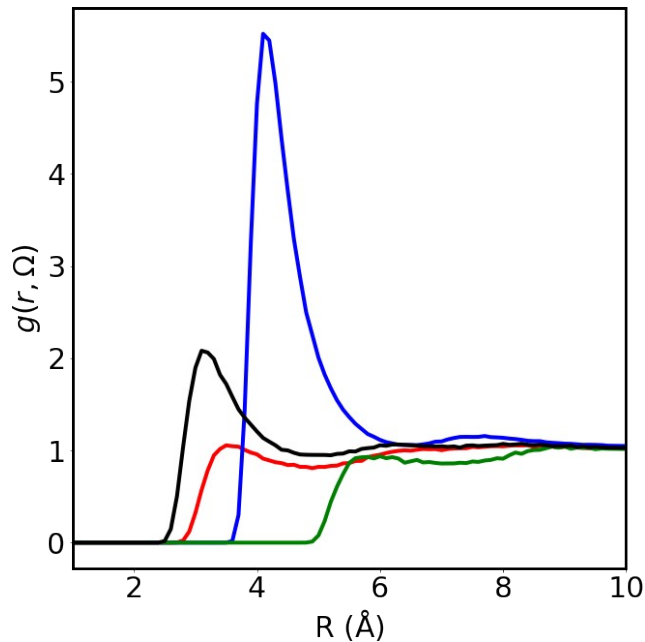
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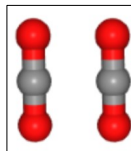
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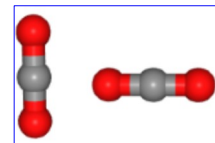
Pair correlation
function



1



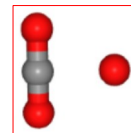
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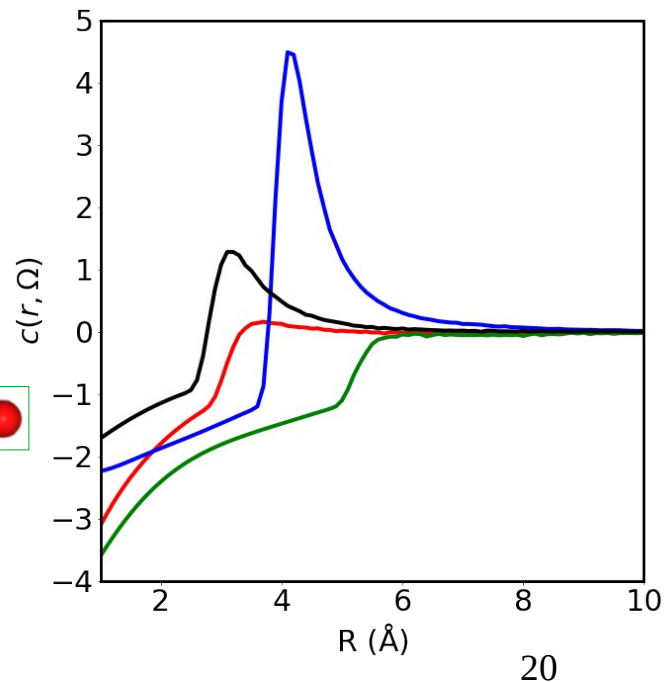
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4



Direct correlation
function

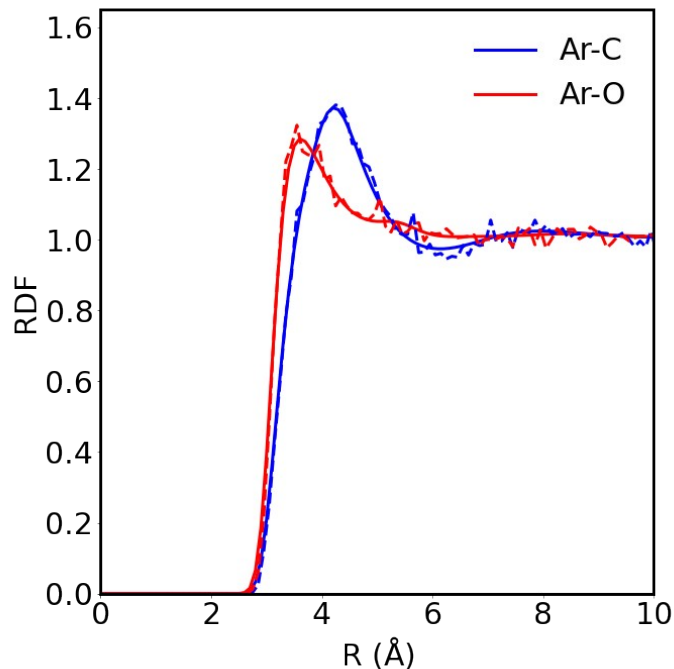


20

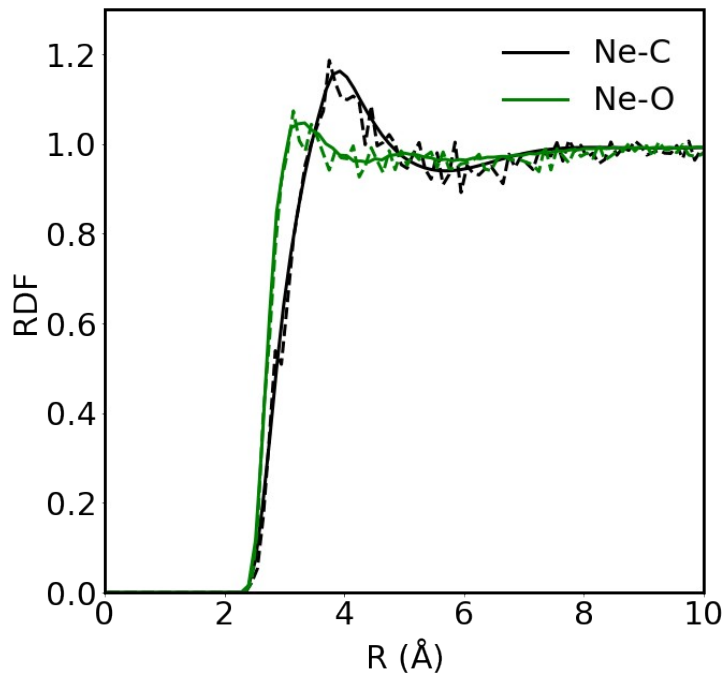
Application of cDFT with spherical solute

We constructed our functional!

$$\mathcal{F}[\rho] = \mathcal{F}_{\text{id}}[\rho] + \mathcal{F}_{\text{ext}}[\rho] + \mathcal{F}_{\text{exc}}[\rho]$$



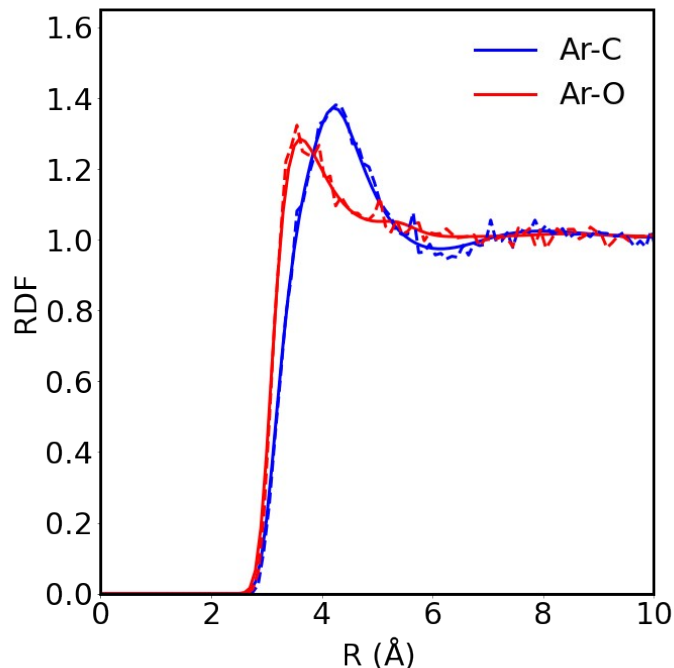
— cDFT
- - MD



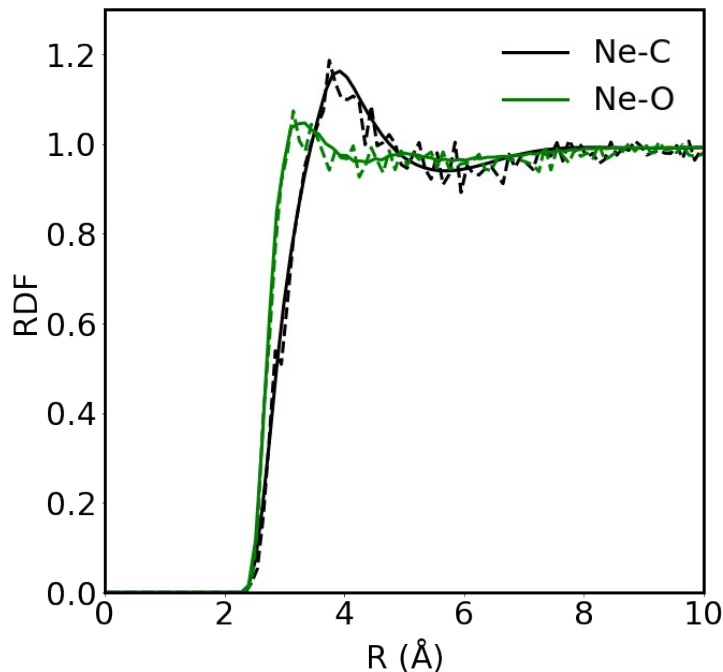
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— cDFT
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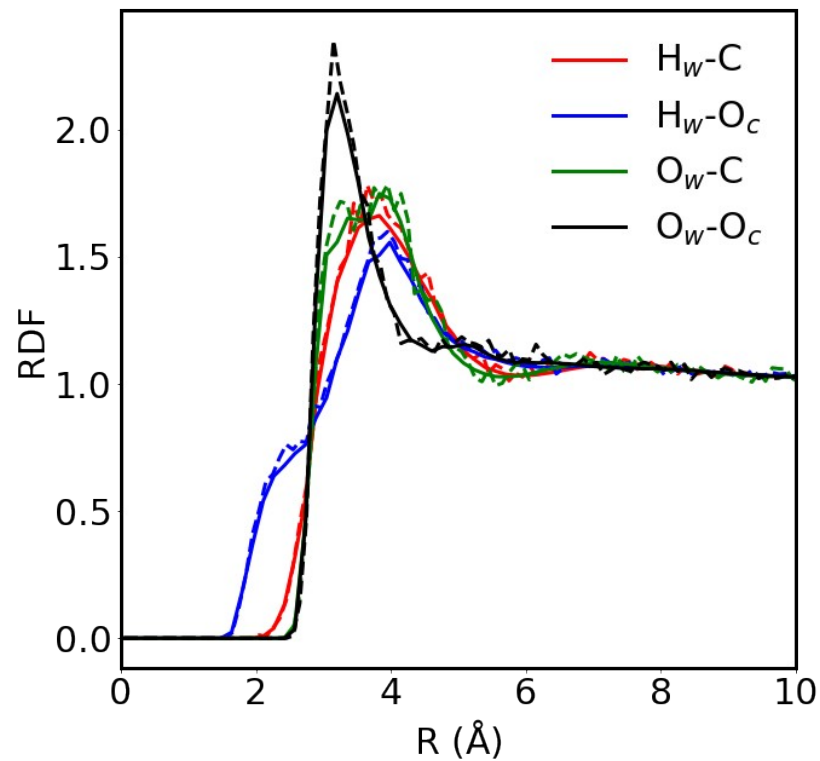
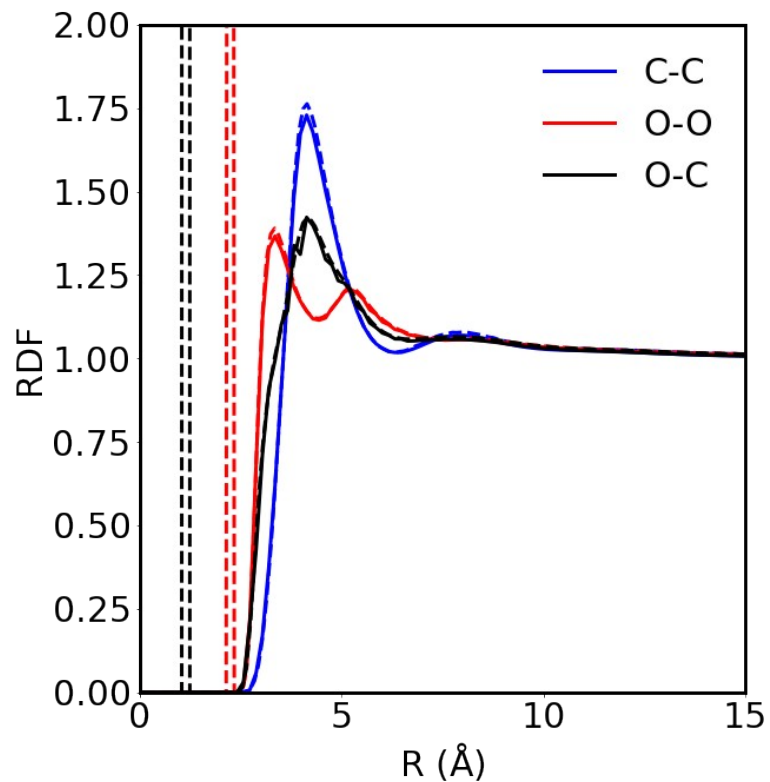


MD : 32 ns, 4000 CO₂, NVT (34h on 32 cores)

cDFT : 2 minutes on 1 core

cDFT is 10⁵ faster than MD!

Application of cDFT with molecular solute



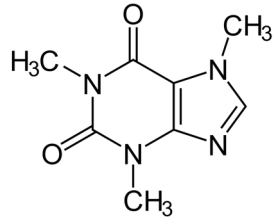
Conclusion

Classical DFT gives excellent structure!

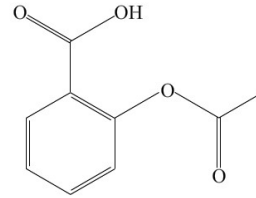
Look into solvation free energy

Study more complex solutes and test cDFT with MD

Caffeine



Aspirin



Find a method to generalise the calculation of direct correlation function for a given P and T

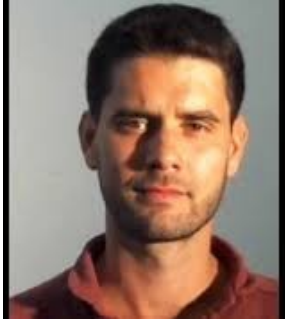
The team



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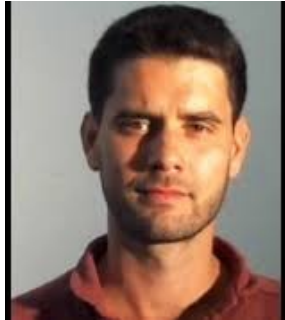
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Thank you!



Luc Belloni



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MD simulations details

Canonical ensemble NVT at $T = 1.05 T_c$ $N=13500$ and $\rho = 0.8 \rho_c$

Rigid CO_2 EPM2 Force field : 3 sites = LJ + charge

Near critical point : simulation of 62 ns

Trajectories were printed every 5 ps : 12000 configurations

Periodic boundary condition (PBC)

