

La température des objets savonneux

François Boulogne

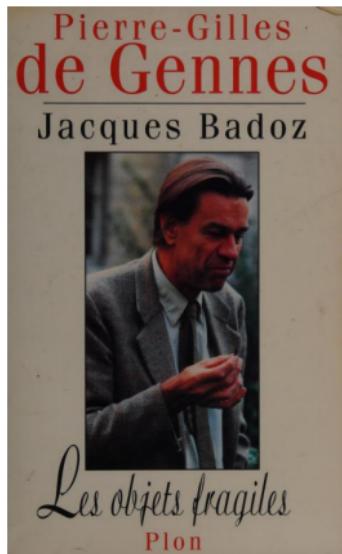


avec Frédéric Restagno & Emmanuelle Rio



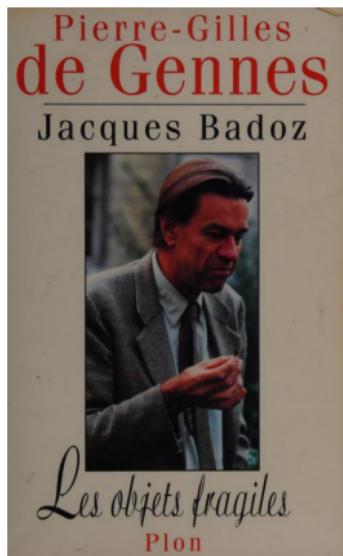
Soap Films and Foams, Fragile Objects

Dans une bulle de savon il y a aussi toutes les étapes de la vie. Elle naît, croît et se développe, vieillit et finalement... disparaît.



Soap Films and Foams, Fragile Objects

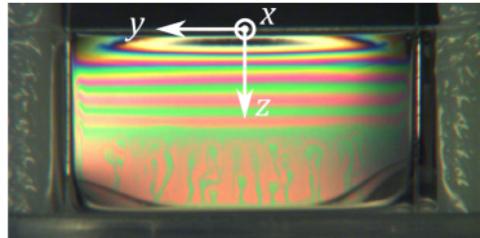
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Causes of soap film rupture?

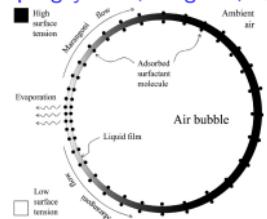
- ▶ Gravity and capillary drainage
- ▶ Marginal regeneration
- ▶ Evaporation

Evaporation contributes to soap film rupture

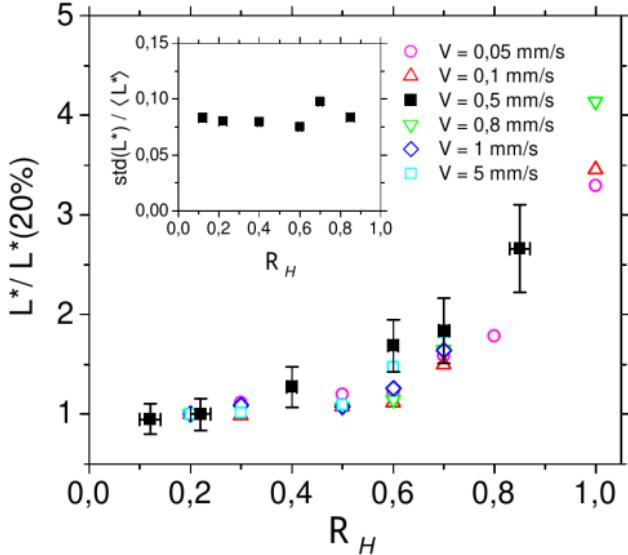


Rupture length L^*

Champougny et al., Langmuir, 2018



Li & Stevenson, Langmuir, 2012



Evaporation induces a film thinning

See also: Bourouiba et al., Villermaux et al., Lorenceau et al., Rouyer et al., Quéré et al., Stevenson et al. ...

Psychrometry

Psychrometry



Psychrometry



$$\Delta e^* = \mathcal{A} P \Delta T^*$$

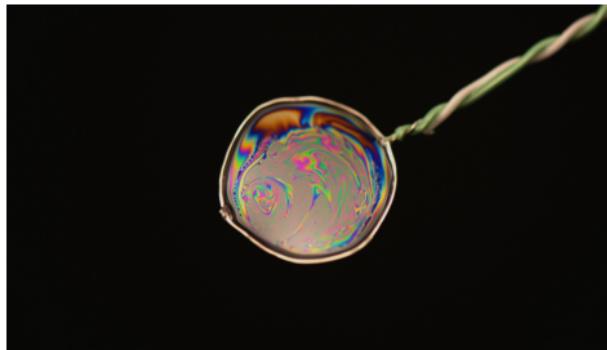
- ▶ $\Delta e^* = e_\infty - e_i$
vapor pressure difference dry & wet
bulbs
- ▶ P is the atmospheric pressure
- ▶ **\mathcal{A} the psychrometer coefficient**
... 150 years of research hidden in it!

From Δe^* we can derive \mathcal{R}_H

Temperature of Soap Films

Temperature of a Soap Film

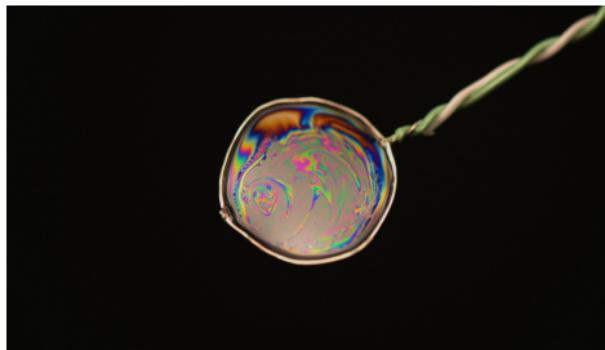
- ▶ Solution of Fairy (Dishwashing soap)
- ▶ Glycerol concentration Γ_g



Soap film on a thermocouple probe

Diameter $2R = 12$ mm

Temperature of a Soap Film



Soap film on a thermocouple probe
Diameter $2R = 12$ mm

- Solution of Fairy (Dishwashing soap)
- Glycerol concentration Γ_g : ability to tune the evaporation rate

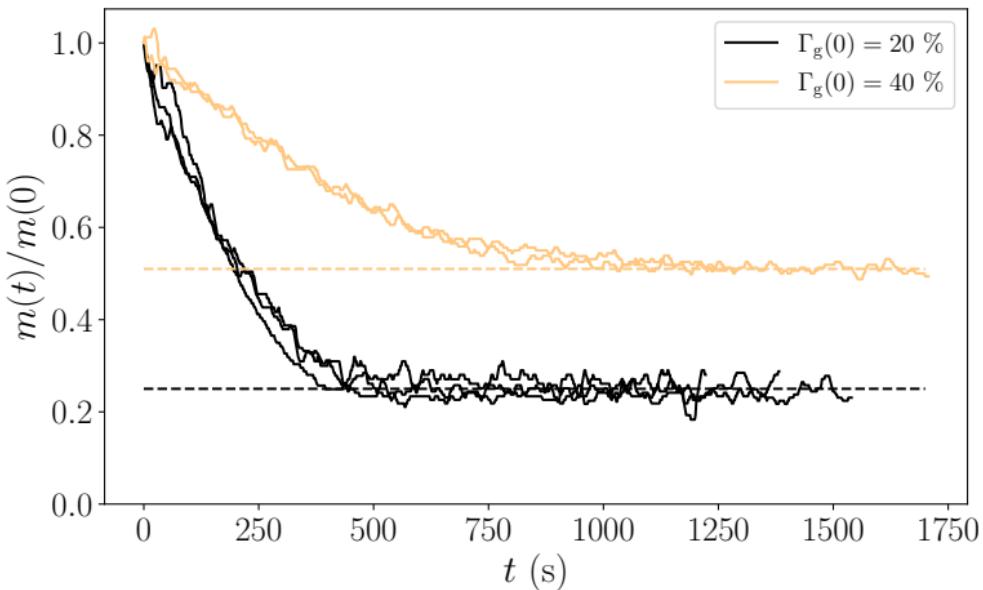
$$e_{\text{sat}}(\Gamma_g, T) = e_{\text{sat}}^0(T) \frac{1 - \Gamma_g}{1 + \Gamma_g(a - 1)}$$

$$a \approx 0.2$$

With Antoine equation

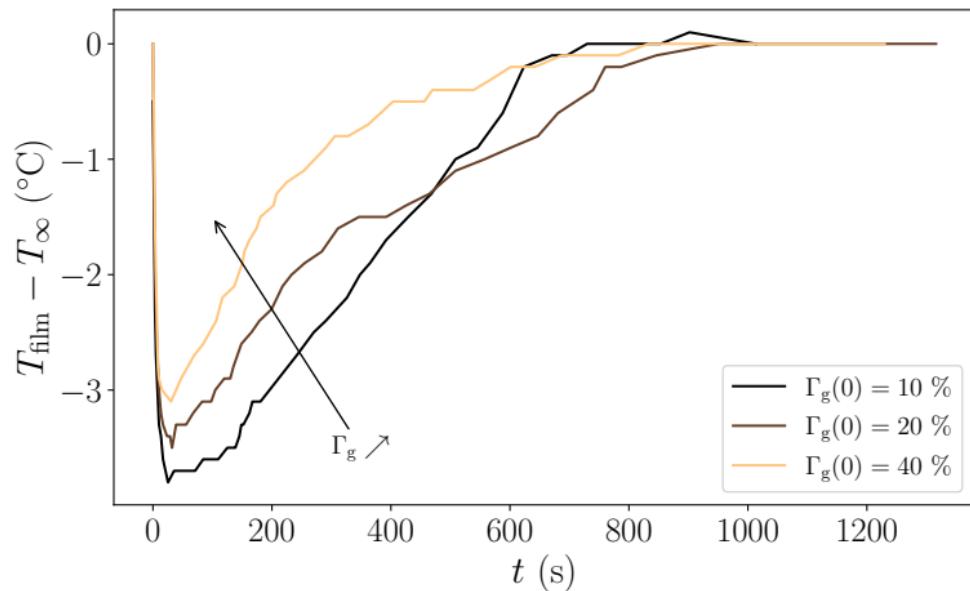
$$e_{\text{sat}}^0(T) = e^\circ 10^{A - \frac{B}{C+T}}$$

Weight dynamics

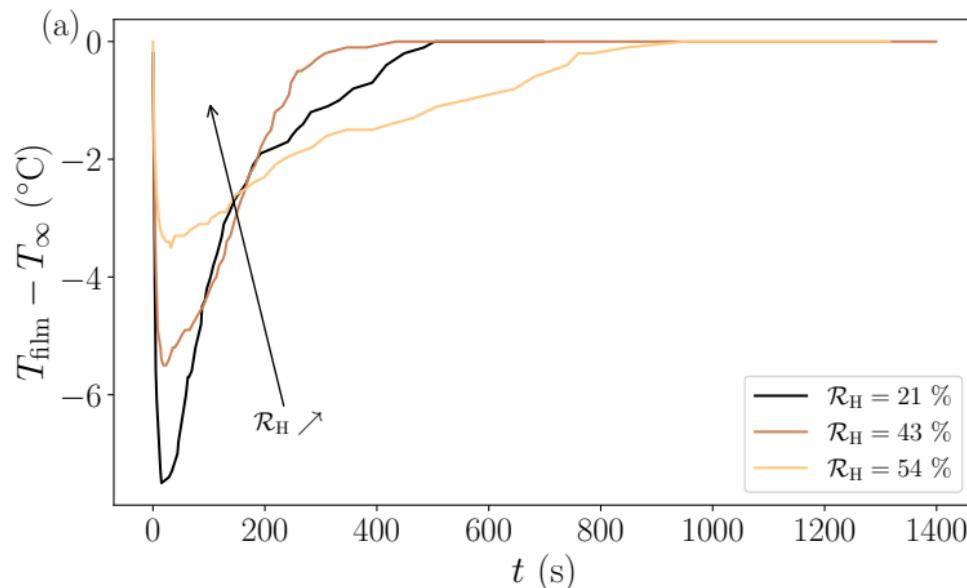


$$\Gamma_g^{\text{eq}} = \frac{1 - \mathcal{R}_H}{1 + (a - 1)\mathcal{R}_H}$$

Temperature: Effect of glycerol



Temperature: Effect of relative humidity



How to predict the minimum temperature?

Psychrometer Coefficient

h_{ev} enthalpy of vaporization

Heat flux balance

$$h_{\text{ev}} \Phi_{\text{ev}} + Q_h + Q_{\text{rad}} = 0$$

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Heat flux balance

$$h_{\text{ev}}\Phi_{\text{ev}} + Q_h + Q_{\text{rad}} = 0$$

Evaporative flux ($\Delta c = 0$)

$$\Phi_{\text{ev}} = 4\mathcal{D}R\Delta c^*$$

$$\Delta c^* = c_\infty - c_{\text{sat}}(\Gamma, \textcolor{red}{T_i})$$

Cooke, 1967; Lebedev 1965

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Heat flux from the atmosphere

$$Q_h = 4\lambda_{\text{air}} R \Delta T^*$$

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Psychrometric equation

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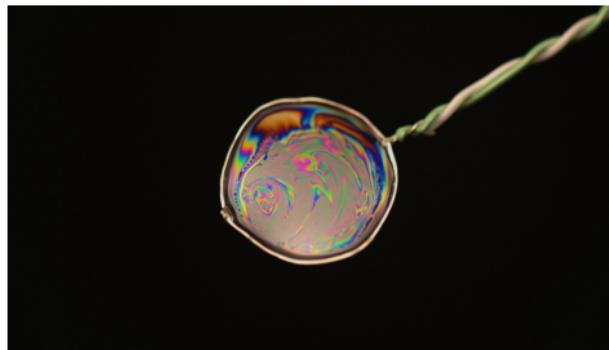
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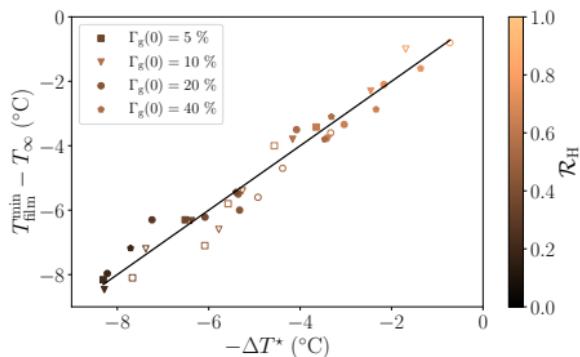
Minimum Temperature

$$\Delta e^* = -P \frac{M_{\text{air}}}{\rho_{\text{air}} M_w} \frac{\lambda_{\text{air}}}{D h_{\text{ev}}} \left(1 + \frac{\pi R \epsilon \sigma T_\infty^3}{\lambda_{\text{air}}} \right) \Delta T^*$$

Characteristic radius for $Q_{\text{rad}} = Q_h$: 7 mm.



Soap film on a thermocouple probe
Diameter $2R = 12$ mm



No adjustable parameter.

Perspectives

Evaporation is not only a **thinning effect**
but also a **cooling effect**
not quantified in the litterature on soap films

Order of magnitude: 2 to 8°C

Consequence on stability?

Thank you for your attention