Rechauffement et acidification des océans

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The ocean is influenced by many factors



Gattuso et al. 2015

The atmospheric CO₂ increase is well documented



More atmospheric CO₂ means increased ocean acidity



Schematic: Sam Dupont, University of Gothenburg

Some numbers:

4 kg CO₂ per day **per person** added to the ocean

+30% acidity in surface ocean since start of industrial era

+100 to 200% by 2100

Ocean acidification is largely from atmospheric CO_2 increase but other factors may also affect coastal regions



Kelly et al (2011, Science)

Change in pH from ocean acidification is already measurable



<u>Data</u>:

Bates (2007) Dore et al. (2009) Santana-Casiano et al. (2007) Gonzàles-Dàvila et al. (2010)



IPCC AR5 WG1 Report (2013)

Today's atmospheric CO_2 level appears as instantaneous spike relative to glacial-interglacial variations



Luthi et al. (2008, Nature)

Today's rate of ocean acidfication is unprecedented



Current change:

Barker and Ridgwell (2012)

- overwhelms natural variations (last 800 000 years)
- 10x faster than natural event (55 million years ago)
- unprecedented (over last 300 million of years)
- 26% increase in acidity (H+) during industrial era
- 100% increase (or more) projected by 2100

Atmospheric CO₂ scenarios used in projections describe very different possible futures



Future emissions cause future additional warming



IPCC AR6 WG1 report

Surface air temperature warms less over the ocean

Simulated changes...





IPCC AR6 WG1 report

The sea surface warms everywhere



Animation Copyright: James C. Orr

Projections (IPCC AR5 WG1, 2013) <u>scenario RCP8.5</u>

Surface ocean acidity increases everywhere



Animation Copyright: James C. Orr

Projections (IPCC, AR5 WG1, 2013) <u>scenario RCP8.5</u>

> see also Bopp et al. (2013, Biogeosciences) Kwiatkowski et al. (2013, Biogeosciences)

Large regional differences, but the intensity depends on us



Differences between models is less than between scenarios, especially for pH





IPCC AR5 WG1 Report (2013)

There is an increase not only in annual average acidity, but also the seasonal amplitude, doubling by 2100

21st century increase in seasonal amplitude (summer–winter difference)



Kwiatkowski & Orr (2018, Nat. Clim. Change)

The formation and dissolution of $CaCO_3$ depend on the *saturation state*

$$\Omega = [Ca^{2+}][CO_3^{2-}]/K_{sp}^*$$

$$K_{sp}^{*} = [Ca^{2+}]_{sat}[CO_{3}^{2-}]_{sat}$$

$$CO_2 + CO_3^2 + H_2O \rightarrow 2 HCO_3^2$$

The *corrosivity* of surface seawater increases this century

Seawater corrosivity to aragonite, a CaCO₃ mineral that certain organisms secrete to build their skeletal material (corals, shell builders)



These corrosive conditions dissolve shells of sea butterflies



Movie: Brad Seibel, University of Rhode Island

Orr et al. (2005)

Fabry et al. (2008)

Comeau et al. (2009; 2011; 2012)

Lischka et al. (2011); Lischka & Riebesell (2012)

Bednarsek et al. (2012)

Sea butterfly shells $(CaCO_3)$ exposed to corrosive conditions expected by 2100



Image: Victoria Fabry, California State University San Marcos

Most tropical corals projected to be exposed to unsustainable chemical conditions by mid-century (e.g., $\Omega_{arag} < 3.0$)



Analysis of 13 Earth System Models (CMIP5)



Marine heatwaves have already resulted in large-scale coral bleaching events causing worldwide reef degradation Vulnerable Ecosystems identified in AR5, SR1.5, SROCC



IPCC SROCC report

By 2100 there are large changes in subsurface corrosivity to $CaCO_3$

- Surface undersaturation
 - Southern Ocean
 - Subarctic Pacific
- Shoaling of aragonite saturation horizon
 - S. Ocean by 1000 m
 - N. Atlantic by 3000 m



Orr et al. 2005 (Nature)

Most cold-water corals will be exposed to corrosive conditions during this century



L. pertusa with expanded tentacles ready to capture zooplankton

Guinotte et al. (2006) Davies et al. (2008) Fautin et al. (2009) Tittensor et al. (2010)

70

50

Regions that are naturally rich in CO₂ confirm expected trends

- less biodiversity
- fewer shells & corals
- more fragile shells
- invasive species

seafloor at Ischia, Bay of Naples, a natural lab to study acidification

CO₂ bubbles rise from

Hall-Spencer et al. (2008) Rodolfo-Metalpa et al. (2008)

more seaweed, coral degradation

Photo: Steve Ringman, Seattle Times





Photo: Jason Hall-Spencer, Plymouth University

Another natural CO₂ vent site in Papua, New Guinea, used to study effects of acidification on corals

Acidification likely to change marine ecosystems

Organisms react differently

Corals and shell builders decline

Seagrasses may increase

Fish become disoriented

Predators affected by prey loss

Potential fish catch decline

Synthesis of existing experimental studies



Wittmann & Pörtner (2013, Nature Clim. Change)

Ocean acidification will also affect humans

- Fish is primary source of animal protein for 1 billion people, mostly in developing countries (FAO)
- Coral reefs provide
 - home for millions of species
 - storm protection for coastlines
 - income from tourism
 - biodiversity legacy for future
- Ocean acidification already affecting oyster industry (U.S. west coast)
- Ocean acidification may well affect aquaculture, fisheries, and human livelihoods



Photo: Rodolfo Quevenco, IAEA



Photo: Jean-Louis Teyssié, IAEA



IPCC SROCC report

Summary

Ocean Acidification

Already detectable

Fast

Caused by CO₂ emissions

Negative impacts on ecosystems



Bopp et al. 2013

IGBP, IOC, SCOR (2013) Ocean Acidification Summary for Policy makers

Future projections of egg survival for polar and Atlantic cod across the Arctic

James Orr (LSCE) Fanny Dubanton (LSCE, Ecole Polytechnique)

Photo: Peter Leopold (NPL)







Global warming leads to poleward migration of fish biomass

Change in marine biomass of fish & invetebrates (1986-2005 to 2081-2100)

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Percent change Average by 2081–2100, relative to 1986–2005

IPCC SROCC report

Arctic Ocean is mostly shallow continental shelves and includes many regional seas





Can **Polar cod** survive climate change?

Will climate change allow Atlantic cod to invade the Arctic?

Polar cod (Boreogadus saida) is omnipresent in the Arctic Ocean



Polar cod is sympagic. It relies on sea ice for spawning, habitat, food, etc





Polar cod (*Boreogadus saida*) is a central element of the Arctic Ocean ecosystem



Polar cod (Boreogadus saida) is a central element of the Arctic Ocean ecosystem



Fish have limited thermal tolerance, especially eggs & spawners





Pörtner and Farrell (2008)

Warming & acidification affect embryos (eggs) more than adults



Egg survival was measured experimentally as a function of temperature and pCO_2



Dahlke et al. (2018, Sci. Adv.)

Current spawning areas have exhibit high habitat suitability (PES > 90%)



Dahlke et al. (2018, Sci. Adv.)

Arctic surface sea-surface temperature increases less in winter than summer (opposite of air surface temperature)



18 CMIP6 models: mean $\pm 1\sigma$

Ocean pCO_2 has increased greatly, nearly uniformly, closely following the atmospheric CO₂ increase



18 CMIP6 models: mean $\pm 1\sigma$

For Polar cod eggs, all suitable habitat disappears by 2100 except in the low-end, and perhaps mid-range, scenarios



For Atlantic cod, traditional spawning grounds disappear, but might move to the Barents Sea in lower scenarios



Fraction of area with suitable habitat (PES > 90%) remains if we consider only warming but disappears with added acidification



The thermal windows for the two cod species do not coincide



Conclusion

All polar cod spawning habitat lost when atm CO2 > 700 ppm

Atlantic cod spawning forced to move & lost under high scenario

Most of the heat from anthropogenic global warming is absorbed by the ocean



Manipulative studies used to evaluate biological resp

• Lab perturbation experiments

• Field observations near CO₂ vents (natural, long term)

- Mesocosm experiments (in the water; on the sediments)
- Free Ocean CO₂ Enrichment (FOCE) experiments

Conclusions

- General *Amplification* in seasonal cycles of surface $pCO_2 \& [H^+]$ vs. *Attenuation* in seasonal cycles of surface pH and Ω_A
- In the Arctic, today's seasonal minima in pCO₂ & [H⁺] become tomorrow's seasonal maxima
- That phase change worsens summer acidity by ~30% compared to an amplification with no phase change
- How these big increases in summer SST, pCO₂, and [H⁺] will affect the Arctic Ocean BGC & ecosystems has been ignored

Future changes in animal biomass including fish and invertebrates

Percent change Average by 2081–2100, relative to 1986–2005



IPCC SROCC report



IPCC SROCC report

Seasonal amplitude of surface ocean acidity & pCO_2 increases, particularly in polar oceans



9 CMIP5 models under RCP8.5

Kwiatkowski & Orr (2018, Nat. Clim Change)

Different scenarios result in large future differences in atm CO₂



Every tonne of CO₂ emissions adds to global warming



IPCC AR6 WG1 report